

Engineering + Environmental

November 22, 2016

Jeremy Miller Maintenance Manager Department of Administrative Services Enterprise Asset Management Division 1225 Ferry Street SE Salem, Oregon 97301

Via email: Jeremy.W.MILLER@oregon.gov

Regarding: Drinking Water Sampling for Lead Adolphson Building 635 Capitol Street NE Salem, Oregon 97301 PBS Project #: 25103.003 Phase 0002

Dear Mr. Miller:

On October 10, 2016, PBS Engineering and Environmental Inc. (PBS) performed drinking water sampling at the Adolphson building located at 635 Capitol Street NE in Salem, Oregon. The testing was requested by State of Oregon Department of Administrative Services in an effort to ensure that concentrations of lead in drinking water remain below the EPA action level.

Sampling methodology and the interpretation of laboratory results were based on the EPA Lead and Copper Rule (LCR). Following LCR sampling guidelines, PBS collected the first 1000 milliliters (mL) of water from each test location (first draw) early in the morning following an overnight stagnation period. The LCR's stagnation period, and sampling protocol specifying the first 1000 mL samples, is designed to maximize the likelihood that the highest concentrations of lead are identified in water used for consumption. At each sample location, immediately following first draw sampling, a flush sample was collected after the water had been allowed to run for 30 seconds.

The water sampling process was supervised by a certified industrial hygienist (CIH) who is also an Oregon Health Authority certified lead risk assessor.

The action level set by the EPA for lead is 15 parts per billion (ppb). If the action level is exceeded in more than 10 percent of taps sampled, then action must be taken to control plumbing-material corrosion.

One first draw and one flush drinking water samples were collected and delivered under chain of custody to BSK Laboratories in Vancouver, Washington for lead analysis. Only the first draw sample was analyzed. If the first draw sample had exceeded the EPA action level for lead, its associated flush sample would have been analyzed.

The lead concentration in the first draw sample was 1.8 ppb, indicating that this drinking water sample contained lead at a concentration below the EPA action level of 15 ppb.

The following table presents the first draw sample location and lead concentration in ppb.

Drinking Water Sampling for Lead, Department of Administrative Services Adolphson Building November 22, 2016 Page 2 of 2

Sample Number	Sample Location	Lead Concentration (ppb)
SK-AD-001-FD	First floor kitchen sink	1.8

ND: None Detected

Please refer to the attached Chain of Custody form and laboratory data for greater details. It should be noted that quality control (QC) sample results are included at the end of laboratory information. The QC samples are both laboratory blanks and spiked samples used internally by the laboratory to assess accuracy.

Please feel free to contact me at 503.417.7602 or derek.may@pbsenv.com with any questions or comments.

Sincerely, PBS Engineering and Environmental Inc.

S. Durl Hy

Derek May, Principal

Attachments: Laboratory Results Chain of Custody Form

DM::bmp

The information contained in this document is proprietary and shall not be duplicated, used, or disclosed in whole or in part to other parties without the permission of PBS.



**BSK Associates Fresno** 1414 Stanislaus St Fresno, CA 93706 559-497-2888 (Main)



Derek May **PBS Environmental** 4412 SW Corbett Ave Portland, OR 97239

### RE: Report for A6J1750 Oregon DAS - Lead

Dear Derek May,

Thank you for using BSK Associates for your analytical testing needs. In the following pages, you will find the test results for the samples submitted to our laboratory on 10/13/2016. The results have been approved for release by our Laboratory Director as indicated by the authorizing signature below.

The samples were analyzed for the test(s) indicated on the Chain of Custody (see attached) and the results relate only to the samples analyzed. BSK certifies that the testing was performed in accordance with the quality system requirements specified in the 2009 TNI Standard. Any deviations from this standard or from the method requirements for each test procedure performed will be annotated alongside the analytical result or noted in the Case Narrative. Unless otherwise noted, the sample results are reported on an "as received" basis.

If additional clarification of any information is required, please contact your Project Manager, Debra Karlsson , at 559-497-2888.

Thanks again for using BSK Associates. We value your business and appreciate your loyalty.

Sincerely,

Debra Karlsson, Project Coordinator



Accredited in Accordance with NELAP **ORELAP #4021** 



# **Case Narrative**

Project and	Report Details	Invoice Details	;
Client:	PBS Environmental	Invoice To:	PBS Environmental
Report To:	Derek May	Invoice Attn:	Accounts Payable
Project #:	Adolphson #25103.003 PH	2 Project PO#:	-
Received:	10/13/2016 - 09:00		
Report Due:	10/27/2016		
Sample Red	ceipt Conditions		
	ault Cooler on Receipt °C: 20.2	Containers Intact COC/Labels Agree Received with no thermal preservation. Sample(s) split after receipt at the laboratory. Initial receipt at BSK-VAL	
Data Quali	fiers		
The following	g qualifiers have been applie	d to one or more analytical results:	
***None applie	d***		
Report Dis	tribution		

Recipient(s)	Report Format	CC:
Derek May	FINAL.RPT	



# **Certificate of Analysis**

Sample ID: A6J1750-01 Sampled By: Client Sample Description: SK-AD-001-FD // 1st Floor kitchen sink Sample Date - Time: 10/10/16 - 00:00 Matrix: Drinking Water Sample Type: First Draw

### **BSK Associates Fresno**

Metals

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0018	0.0010	mg/L	1	A614384	10/19/16	10/19/16	



## BSK Associates Fresno Metals Quality Control Report

				Spike	Source		%REC		RPD	Date	
Analyte	Result	RL	Units	Level	Result	%REC	Limits	RPD	Limit	Analyzed	Qual
EPA 200.8 - Quality Control											
Batch: A614384										Prepared:	10/19/2016
Prep Method: EPA 200.2										Ar	alyst: GNG
Blank (A614384-BLK1)											
Lead	ND	0.0010	mg/L							10/19/16	
		010010									
Blank Spike (A614384-BS1)											
Lead	0.098	0.0010	mg/L	0.10		98	85-115			10/19/16	
Blank Spike Dup (A614384-BSD1)											
Lead	0.098	0.0010	mg/L	0.10		98	85-115	0	20	10/19/16	
2000	0.000	0.0010	iiig/L	0.10		00	00 110	Ŭ	20	10,10,10	
Matrix Spike (A614384-MS1), Source:	A6J1785-21										
Lead	0.20	0.0020	mg/L	0.20	0.0029	101	70-130			10/19/16	
Matrix Spike (A614384-MS2), Source:	A6 11750-01										
Lead	0.20	0.0020	mg/L	0.20	ND	100	70-130			10/19/16	
2000	0.20	0.0020	iiig/L	0.20	NB	100	10 100			10/10/10	
Matrix Spike Dup (A614384-MSD1), So	ource: A6J1785-21										
Lead	0.20	0.0020	mg/L	0.20	0.0029	99	70-130	1	20	10/19/16	
Matrix Spike Dup (AG14294 MSDO) Sa	NUROOL & 6 14750 04										
Matrix Spike Dup (A614384-MSD2), So Lead	0.20	0.0020	ma/l	0.20	ND	102	70-130	2	20	10/19/16	
Leau	0.20	0.0020	mg/L	0.20	ND	102	70-130	2	20	10/19/10	



## **Certificate of Analysis**

#### Notes:

- The Chain of Custody document and Sample Integrity Sheet are part of the analytical report.
- Any remaining sample(s) for testing will be disposed of according to BSK's sample retention policy unless other arrangements are made in advance.
- All positive results for EPA Methods 504.1 and 524.2 require the analysis of a Field Reagent Blank (FRB) to confirm that the results are not a contamination error from field sampling steps. If Field Reagent Blanks were not submitted with the samples, this method requirement has not been performed.
- Samples collected by BSK Analytical Laboratories were collected in accordance with the BSK Sampling and Collection Standard Operating
  Procedures.
- J-value is equivalent to DNQ (Detected, not quantified) which is a trace value. A trace value is an analyte detected between the MDL and the laboratory reporting limit. This result is of an unknown data quality and is only qualitative (estimated). Baseline noise, calibration curve extrapolation below the lowest calibrator, method blank detections, and integration artifacts can all produce apparent DNQ values, which contribute to the un-reliability of these values.
- (1) Residual chlorine and pH analysis have a 15 minute holding time for both drinking and waste water samples as defined by the EPA and 40 CFR 136. Waste water and ground water (monitoring well) samples must be field filtered to meet the 15 minute holding time for dissolved metals.
- Summations of analytes (i.e. Total Trihalomethanes) may appear to add individual amounts incorrectly, due to rounding of analyte values occurring before or after the total value is calculated, as well as rounding of the total value.
- RL Multiplier is the factor used to adjust the reporting limit (RL) due to variations in sample preparation procedures and dilutions required for matrix interferences.
- Due to the subjective nature of the Threshold Odor Method, all characterizations of the detected odor are the opinion of the panel of analysts. The characterizations can be found in Standard Methods 2170B Figure 2170:1.
- The MCLs provided in this report (if applicable) represent the primary MCLs for that analyte.

#### Definitions

mg/L:	Milligrams/Liter (ppm)	MDL:	Method Detection Limit	MDA95:	Min. Detected Activity
mg/Kg:	Milligrams/Kilogram (ppm)	RL:	Reporting Limit: DL x Dilution	MPN:	Most Probable Number
µg/L:	Micrograms/Liter (ppb)	ND:	None Detected at RL	CFU:	Colony Forming Unit
µg/Kg:	Micrograms/Kilogram (ppb)	pCi/L:	Picocuries per Liter	Absent:	Less than 1 CFU/100mLs
%:	Percent Recovered (surrogates)	RL Mult:	RL Multiplier	Present:	1 or more CFU/100mLs
NR:	Non-Reportable	MCL:	Maximum Contaminant Limit		

Please see the individual Subcontract Lab's report for applicable certifications.

#### BSK is not accredited under the NELAP program for the following parameters: \*\*NA\*\*

#### Certifications: Please refer to our website for a copy of our Accredited Fields of Testing under each certification.

Fresno			
State of California - ELAP	1180	State of Hawaii	4021
State of Nevada	CA000792016-1	State of Oregon - NELAP	4021
EPA - UCMR3	CA00079	State of Washington	C997-16
Sacramento			
State of California - ELAP	2435		
San Bernardino			
State of California - ELAP	2993	State of Oregon - NELAP	4119-001
Vancouver			
State of Oregon - NELAP	WA100008-008	State of Washington	C824-16

	PBS	Engineering + Environmenta		9	10/13/2016 10 LEAD IN DRINKING WATER TESTING PROGRAM 25103.003	
	FACILITY NAM	E: Adolphson			PROJECT #: PH	2
20-7	ANALYSIS RE		RINKING WATER	nd	DATE: 10/10/16 DATE/TIME: 10/13/16 0900 TURN AROUND TIME: 7-10 days	<del>1000</del> 08
			SAMPLE	DATA FO	RM	7
	LAB	SAMPLE#	BUILDING	ROOM	LOCATION IN ROOM	
	2	5K-AD-001-FD.		*	Kitchen, 1 <sup>st</sup> Floor, Kitchen Sink	_
		-				

-

PBS ENGINEERING + ENVIRONMENTAL, 4412 SW Corbett Avenue

						6J1750		10	/13/20 10	10
State State State State	Associates SR-FL-0002-16				P.	BSEN1939	) Tana mini kutok tiku ti	NI NA MANANA		******
Sa	Imple Integrity									
BS	K Bottles: Yes (No ) Page	( of	1			<b>B</b> in the second	<b>1121 11811 13861 8</b> 00 4			
	Was temperature within range2		1	Wer	e correct contai	iners and n	esen/ative		6	
0	Chemistry ≤ 6°C Micro < 10°C	Yes No	(A)		ived for the tes			5	Yes	No NA
COC Info	If samples were taken today, is there evidence that chilling has begun?	Yes No	VA)		e there bubbles atiles Only)	in the VOA	vials?		Yes	No NA
Ö	Did all bottles arrive unbroken and intact?	(Yes N	lo		a sufficient am	ount of san	nple receiv	ed?	(Yes	
U U U	Did all bottle labels agree with COC?	(Yes) N	0	Do s	amples have a	hold time <	72 hours?		Yes	
	Was sodium thiosulfate added to CN sample(s) until chlorine was no longer present?	Yes No (1	VA)	Was PM:	PM notified of	discrepanci By/Time:	es?		Yes	No NA
	250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V)	Checks	Pas	sed?	1-2	Dy/ Time.	1	1		
	Bacti Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	-	-							
	None (P) <sup>White Cap</sup>		1997 - 1998 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997 - 1997							
	Cr6 (P) Lt. Green Label/Blue Cap NH4OH(NH4)2SO4 DW	Cl, pH > 8	Y	Ν						
102	Cr6 (P) Pink Label/Blue Cap NH40H(NH4)2SO4 WW	pH 9.3-9.7	Y	Ν				and the second second		
the lab	Cr6 (P) Black Label/Blue Cap NH4OH(NH4)2SO4 7199	pH 9.0-9.5	Y	N						
			-							
nerformed in	H <sub>2</sub> SO <sub>4</sub> (P) or (AG) Yellow Cap/Label	pH < 2	Y	N		2470.200				
rfor	NaOH (P) Green Cap	Cl, pH >10	Y	N				1		and the second second
ue De		pH > 9	Y	N			Production of the second	e cess	0.85	
or ar	Dissolved Oxygen 300ml (g)	_	-	_					1.094	
A C	None (AG) 608/8081/8082, 625, 632/8321, 8151, 8270	_						C North		
er N				and the second						
Bottles Received	Ascorbic, EDTA, KH <sub>2</sub> Ct (AG) <sup>Pink Label</sup> 525		11 13 10 10 10 10 10 10 10 10 10 10 10 10 10	_				-		
	Na2O3S 250ml (AG)Neon Green Label 515		1			TRUE AS A STATE	AN REPORTATION	-		S. M. Marganes Marg
ttles F	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 1 Liter (Brown P) 549									
che che	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (AG) <sup>Blue Label</sup> 548, THM, 524	R. States and	_						CHO SA A	Contraction of the second
<b>B</b> C lorine			-					A Laborator	and the second	
chlo	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> + MCAA (CG) <sup>Orange Label 531</sup>	pH < 3	Y	N			Carl Alexandre		234243	
		p11<0	1							
rvat	EDA (AG) <sup>Brown Label</sup> DBPs	-	-					1		
preservation/		-	-	-						
			-	-	a state of the sta				The local days	
means	Buffer pH 4 (CG) H <sub>3</sub> PO <sub>4</sub> (CG) <sup>Salmon Label</sup>		-					•		
ے ا	Other:	-		-	•					
=	Asbestos 1Liter Plastic w/ Foil	_		_		(Carlor Carlor)				1
	Low Level Hg / Metals Double Baggie		-	-	1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 - 1999 -			and the second second		
	Bottled Water	-	-							
	Clear Glass 250mL / 500mL / 1 Liter Soil Tube Brass / Steel / Plastic		-	-		10 A.M. 10 A.M. 10 A.M.				
	Tedlar Bag / Plastic Bag	_	-	-		Sec. 19	Constant Service			
ىب	Container Preservative Date/	Time/Initials			Container	Pres	ervative	Dat	e/Time	/Initials
Split	(s)P 250₽		S	Р						
0)	S P		S	Р						
Comments	* Odd number	s ænly-	R	IR						

6 -





# PBSEN1939



PBS Environmental



# 10142016

Turnaround:	Standard
Due Date:	10/27/2016





Printed: 10/19/2016 10:54:54AM Page 1 of 1 Page 8 of 9

## BSK Associates SR-FL-0002-16

A6J1750	
PBSEN1939	

10/13/2016

10

	Associates SR-FL-0002-16 mple Integrity										
BS	K Bottles: Yes( <sup>(</sup> No / <sub>Pag</sub>	eof			•						
	Was temperature within range2 Chemistry ≤ 6°C Micro < 10°C		(A)	Were correct containers and preservatives (Yes) No							
coc Info	If samples were taken today, is there evidence that chilling has begun?	Yes No (	A	Were there bubbles in the VOA vials? (Volatiles Only)					Yes .	No	(N/
20	Did all bottles arrive unbroken and intact?	-+	lo	Was a sufficient amount of sample received? (Yes)					) —	No	
õ	Did all bottle labels agree with COC? Was sodium thiosulfate added to CN sample(s)	Yes N	10			a hold time <72 hours? of discrepancies?			Yes	5	(No
	until chlorine was no longer present?	Yes No (N	(A)	PM		oiscrepancie By/Time:	es?		Yes	No	K
	250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V)	Checks	Pas	sed?	1-2		· · · · ·				7
	Bacti Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>		-					1			
	None (P) <sup>White Cap</sup>		-				and the second second second				
	Cr6 (P) L1. Green Label/Blue Cap NH4OH(NH4)2SO4 DW	Cl, pH > 8	Y	N	-	· ·					`
	Cr6 (P) Pink Label/Blue Cap NH4OH(NH4)2SO4 WW	pH 9.3-9.7	Y	N							
the lab	Cr6 (P) Black Label/Blue Cap NH4OH(NH4)2SO4 7199	pH 9.0-9.5	Y	Ň							
<u>_</u>	HNO3 (P) Red Gap or HCI (P) Purple Cap/Lt. Blue Label			_	······································			<u> 1 </u>			
ned	H <sub>2</sub> SO <sub>4</sub> (P) or (AG) Yellow Cap/Label	pH<2	Y	N				1			
ror	NaOH (P) Green Cap	Cl, pH >10	Y	N	<u></u>		<u></u>	4		· ·	
pe	NaOH + ZnAc (P)	pH > 9	Y								
are	Dissolved Oxygen 300ml (g)				<u> </u>						
A or	None (AG) 608/8081/8082, 625, 632/8321, 8151, 8270						······································				
ed r N/A	HCI (AG) <sup>L1. Blue Label</sup> O&G, Diese!								<u>.</u>		
eive either		-						ļ			
Received are either N	Ascorbic, EDTA, KH <sub>2</sub> Ct (AG) <sup>Pink Label</sup> 525 Na <sub>2</sub> O <sub>3</sub> S 250mL (AG) <sup>Neon Green Label</sup> 515			-							
ഗ			-	-	· · · · · · · · · · · · · · · · · · ·					· 	
Bottles	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 1 Liter (Brown P) 549	-		-							
Bo Bo	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (AG) <sup>Bive Label</sup> 548, THM, 524			-							
lori	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (CG) <sup>Blue Label</sup> 504, 505, 547			- 							
ation/chlor	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> + MCAA (CG) <sup>Orange Label</sup> 531	pH < 3	Y	N							
/atic	NH4Cl (AG) <sup>Purple Label</sup> 552		-	-							
ser	EDA (AG) <sup>Brown Label</sup> DBPs	-	-	-				Carlon An		8 1	
pre	HCL (CG) 524.2, BTEX, Gas, MTBE, 8260/624			-							
means	Buffer pH 4 (CG)			-						Sector (	
e E	H <sub>3</sub> PO <sub>4</sub> (CG) <sup>Salmon Label</sup>	-	-	-							<del></del> -
ļ	Other:							<u></u>			<u></u>
ļ	Asbestos 1Liter Plastic w/ Foil	<u> </u>		-							
ŀ	Low Level Hg / Metals Double Baggie Bottled Water	<u>↓</u>		-							
ŀ	Clear Glass 250mL / 500mL / 1 Liter										
ł	Soil Tube Brass / Steel / Plastic	·		-							- <u>-</u>
	Tedlar Bag / Plastic Bag		_	-							
ij		e/Time/Initials			Container	Prese	rvative	Date	Time	e/Init	ials
L	s)p 2504		S	Р							
Comments	* Odd number alls	15 æhly-		P] N	; ;						
		cked by:		@ @		10/18					