

December 30, 2016

Jeremy Miller
Maintenance Manager
Department of Administrative Services
Enterprise Asset Management Division
1225 Ferry Street SE
Salem, Oregon 97301

Via email: Jeremy.W.MILLER@oregon.gov

Regarding: Drinking Water Sampling for Lead

Blind Commission Building 535 SE 12th Avenue Portland, Oregon 97214

PBS Project # 25103.003 Phase 0037

Dear Mr. Miller:

On October 19 and 22, 2016, PBS Engineering and Environmental Inc. (PBS) performed drinking water sampling at the Blind Commission building located at 535 SE 12th Avenue in Portland, Oregon. The testing was requested by State of Oregon Department of Administrative Services in an effort to ensure that concentrations of lead in drinking water remain below the EPA action level.

Sampling methodology and the interpretation of laboratory results were based on the EPA Lead and Copper Rule (LCR). Following LCR sampling guidelines, PBS collected the first 1000 milliliters (mL) of water from each test location (first draw) early in the morning following an overnight stagnation period. The LCR's stagnation period, and sampling protocol specifying the first 1000 mL samples, is designed to maximize the likelihood that the highest concentrations of lead are identified in water used for consumption. At each sample location, immediately following first draw sampling, a flush sample was collected after the water had been allowed to run for 30 seconds.

The water sampling process was supervised by a certified industrial hygienist (CIH) who is also an Oregon Health Authority certified lead risk assessor.

The action level set by the EPA for lead is 15 parts per billion (ppb). If the action level is exceeded in more than 10 percent of taps sampled, then action must be taken to control plumbing-material corrosion.

Eighteen first draw and flush drinking water samples were collected and delivered under chain of custody to BSK Laboratories in Vancouver, Washington for lead analysis. Initially, only first draw samples were analyzed. Any first draw sample that exceeded the EPA action level for lead had its associated flush sample analyzed.

Concentrations of lead in the first draw samples ranged from none detected to 1.8 ppb. Laboratory analysis indicates that all of these drinking water samples contained lead at concentrations below the EPA action level of 15 ppb.

The following table presents all first draw sample locations and lead concentrations in ppb.

#### First Draw Drinking Water Sample Locations and Lead Concentrations

Sample Number	Sample Location	Lead Concentration (ppb)
SK-BLC-001-FD	Break room first floor adjacent to Richard Hoover Conference room across from mail room	1.5
SK-BLC-003-FD	Student lounge adjacent to exercise room first floor kitchen sink	1.8
WF-BLC-005-FD	Water fountain between men's and women's bathrooms across from training kitchen	ND
SK-BLC-007-FD	Training kitchen sink northeast first floor	ND
SK-BLC-009-FD	Training kitchen sink southeast first floor	ND
SK-BLC-011-FD	Training kitchen sink southwest first floor	ND
SK-BLC-013-FD	Training kitchen sink northwest first floor	ND
WF-BLC-015-FD OR OSHA	Upper water fountain between room 118 and 120	1.2
SK-BLC-017-FD	First floor lab kitchen sink	ND

ND: None Detected

Please refer to the attached Chain of Custody form and laboratory data for greater details. It should be noted that quality control (QC) sample results are included at the end of laboratory information. The QC samples are both laboratory blanks and spiked samples used internally by the laboratory to assess accuracy.

Please feel free to contact me at 503.417.7602 or derek.may@pbsenv.com with any questions or comments.

Sincerely,

PBS Engineering and Environmental Inc.

Derek May, Principal

5. Dul sky

Attachments: Laboratory Results
Chain of Custody Form

DM::bmp

The information contained in this document is proprietary and shall not be duplicated, used, or disclosed in whole or in part to other parties without the permission of PBS.



A6J2557

Derek May PBS Environmental 4412 SW Corbett Ave Portland, OR 97239

RE: Report for A6J2557 Oregon DAS - Lead

Dear Derek May,

Thank you for using BSK Associates for your analytical testing needs. In the following pages, you will find the test results for the samples submitted to our laboratory on 10/20/2016. The results have been approved for release by our Laboratory Director as indicated by the authorizing signature below.

The samples were analyzed for the test(s) indicated on the Chain of Custody (see attached) and the results relate only to the samples analyzed. BSK certifies that the testing was performed in accordance with the quality system requirements specified in the 2009 TNI Standard. Any deviations from this standard or from the method requirements for each test procedure performed will be annotated alongside the analytical result or noted in the Case Narrative. Unless otherwise noted, the sample results are reported on an "as received" basis.

If additional clarification of any information is required, please contact your Project Manager, Debra Karlsson, at 559-497-2888.

Thanks again for using BSK Associates. We value your business and appreciate your loyalty.

Sincerely,

Debra Karlsson, Project Coordinator



Accredited in Accordance with NELAP ORELAP #4021

Page 1 of 16





#### **Case Narrative**

Project and Report Details Invoice Details

Client: PBS Environmental Invoice To: PBS Environmental Report To: Derek May Invoice Attn: Accounts Payable

Project #: Blink Commission #25103.003 PH 37 Project PO#: -

**Received:** 10/20/2016 - 09:00

**Report Due:** 11/03/2016

**Sample Receipt Conditions** 

Cooler:Default CoolerContainers IntactTemperature on Receipt °C:19.6COC/Labels Agree

Received with no thermal preservation. Sample(s) split after receipt at the laboratory.

Initial receipt at BSK-VAL

#### **Data Qualifiers**

The following qualifiers have been applied to one or more analytical results:

\*\*\*None applied\*\*\*

#### **Report Distribution**

Recipient(s) Report Format CC:

Derek May FINAL.RPT





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

Sample ID: A6J2557-01 **Sample Date - Time:** 10/19/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-BLC-001-FD // Breakroom 1st Floor adjacent to Richard

Sample Type: First Draw

Hoover Conference room across from Mail Room

į	Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
L	ead	EPA 200.8	0.0015	0.0010	mg/L	1	A614767	10/26/16	10/27/16	





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

**Sample ID:** A6J2557-03 **Sample Date - Time:** 10/19/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-BLC-003-FD // Student lounge adjacent to exercise room 1st

Sample Type: First Draw

Floor kitchen sink

	Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
ī	Lead	EPA 200.8	0.0018	0.0010	mg/L	1	A614767	10/26/16	10/27/16	





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

**Sample ID:** A6J2557-05 **Sample Date - Time:** 10/19/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: WF-BLC-005-FD // Water fountain between men's/women's

bathroom across from training kitchen

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614767	10/26/16	10/27/16	





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

Sample ID: A6J2557-07Sample Date - Time: 10/19/16 - 00:00Sampled By: ClientMatrix: Drinking Water

Matrix: Drinking Water
Sample Type: First Draw

Sample Description: SK-BLC-007-FD // Training kitchen sink Northeast 1st Floor

# BSK Associates Fresno

	Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Ī	Lead	EPA 200.8	ND	0.0010	mg/L	1	A614767	10/26/16	10/27/16	

**Metals** 





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

Sample ID: A6J2557-09 **Sample Date - Time:** 10/19/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-BLC-009-FD // Training kitchen sink Southeast 1st Floor

Sample Type: First Draw

## **BSK Associates Fresno** Metals

	Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Ī	Lead	EPA 200.8	ND	0.0010	mg/L	1	A614767	10/26/16	10/27/16	

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Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

 Sample ID: A6J2557-11
 Sample Date - Time: 10/19/16 - 00:00

 Sampled By: Client
 Matrix: Drinking Water

Matrix: Drinking Water Sample Type: First Draw

Sample Description: SK-BLC-011-FD // Training kitchen sink Southwest 1st Floor

# BSK Associates Fresno

#### Metals

Analyte	Method	Result	RL	Units	RL Mult	Batch Prepared	Analyzed Qual
Lead	EPA 200.8	ND	0.0010	ma/L	1	A614767 10/26/16	10/27/16





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

Sample ID: A6J2557-13Sample Date - Time: 10/19/16 - 00:00Sampled By: ClientMatrix: Drinking Water

Matrix: Drinking Water
Sample Type: First Draw

Sample Description: SK-BLC-013-FD // Training kitchen sink Northwest 1st Floor

# BSK Associates Fresno

#### Metals

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614767	10/26/16	10/27/16	





Blink Commission #25103.003 PH 37

## **Certificate of Analysis**

**Sample ID:** A6J2557-15 **Sample Date - Time:** 10/19/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: WF-BLC-015-FD OR OSHA // Upper water fountain between

Sample Type: First Draw

Room 118 and 120

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0012	0.0010	mg/L	1	A614767	10/26/16	10/27/16	



## BSK Associates Fresno Metals Quality Control Report

			<u> </u>								
Analyte	Result	RL	Units	Spike Level	Source Result	%REC	%REC Limits	RPD	RPD Limit	Date Analyzed Qual	
		EPA 20	00.8 - Q	uality Co	ntrol						_
Batch: A614767 Prep Method: EPA 200.2 - Pb/Cu I	Rule									Prepared: 10/26/ Analyst:	
Blank (A614767-BLK1)											
Lead	ND	0.0010	mg/L							10/27/16	
Blank Spike (A614767-BS1)											
Lead	0.096	0.0010	mg/L	0.10		96	85-115			10/27/16	
Blank Spike Dup (A614767-BSD1)											
Lead	0.097	0.0010	mg/L	0.10		97	85-115	1	20	10/27/16	
Matrix Spike (A614767-MS1), Soui	ce: A6J1964-74										
Lead	0.20	0.0020	mg/L	0.20	0.0035	99	70-130			10/27/16	
Matrix Spike (A614767-MS2), Soui	ce: A6J2557-01										
Lead	0.20	0.0020	mg/L	0.20	ND	97	70-130			10/27/16	
Matrix Spike Dup (A614767-MSD1	), Source: A6J1964-74										
Lead	0.20	0.0020	mg/L	0.20	0.0035	100	70-130	1	20	10/27/16	
Matrix Spike Dup (A614767-MSD2	), Source: A6J2557-01										
Lead	0.20	0.0020	mg/L	0.20	ND	98	70-130	0	20	10/27/16	



#### **Certificate of Analysis**

#### Notes:

- The Chain of Custody document and Sample Integrity Sheet are part of the analytical report.
- Any remaining sample(s) for testing will be disposed of according to BSK's sample retention policy unless other arrangements are made in advance.
- All positive results for EPA Methods 504.1 and 524.2 require the analysis of a Field Reagent Blank (FRB) to confirm that the results are not a contamination error from field sampling steps. If Field Reagent Blanks were not submitted with the samples, this method requirement has not been performed.
- · Samples collected by BSK Analytical Laboratories were collected in accordance with the BSK Sampling and Collection Standard Operating Procedures.
- J-value is equivalent to DNQ (Detected, not quantified) which is a trace value. A trace value is an analyte detected between the MDL and the laboratory reporting limit. This result is of an unknown data quality and is only qualitative (estimated). Baseline noise, calibration curve extrapolation below the lowest calibrator, method blank detections, and integration artifacts can all produce apparent DNQ values, which contribute to the un-reliability of these values.
- · (1) Residual chlorine and pH analysis have a 15 minute holding time for both drinking and waste water samples as defined by the EPA and 40 CFR 136. Waste water and ground water (monitoring well) samples must be field filtered to meet the 15 minute holding time for dissolved metals.
- · Summations of analytes (i.e. Total Trihalomethanes) may appear to add individual amounts incorrectly, due to rounding of analyte values occurring before or after the total value is calculated, as well as rounding of the total value.
- RL Multiplier is the factor used to adjust the reporting limit (RL) due to variations in sample preparation procedures and dilutions required for matrix interferences.
- Due to the subjective nature of the Threshold Odor Method, all characterizations of the detected odor are the opinion of the panel of analysts. The characterizations can be found in Standard Methods 2170B Figure 2170:1.
- The MCLs provided in this report (if applicable) represent the primary MCLs for that analyte.

#### **Definitions**

mg/L:	Milligrams/Liter (ppm)	MDL:	Method Detection Limit	MDA95:	Min. Detected Activity
mg/Kg:	Milligrams/Kilogram (ppm)	RL:	Reporting Limit: DL x Dilution	MPN:	Most Probable Number
μg/L:	Micrograms/Liter (ppb)	ND:	None Detected at RL	CFU:	Colony Forming Unit
μg/Kg:	Micrograms/Kilogram (ppb)	pCi/L:	Picocuries per Liter	Absent:	Less than 1 CFU/100mLs
%:	Percent Recovered (surrogates)	RL Mult:	RL Multiplier	Present:	1 or more CFU/100mLs
NR:	Non-Reportable	MCL:	Maximum Contaminant Limit		

Please see the individual Subcontract Lab's report for applicable certifications.

WA100008-008

BSK is not accredited under the NELAP program for the following parameters: \*\*NA\*\*

Certifications: Please refer to our website for a copy of our Accredited Fields of Testing under each certification.

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021
997-16
119-001
9

A6J2557 FINAL 11022016 1607

Printed: 11/2/2016

Vancouver

State of Oregon - NELAP

QA-RP-0001-10 Final.rpt

C824-16

State of Washington



# Engineering + Environmental

A6J2557 PBSEN1939



10/20/2016



25103.003

FACILITY NAME: BUND COMMISSION	PROJECT #: PH 3
ANALYSIS REQUESTED:  LEAD (PB) IN DRINKING WATER  COPPER (CU) IN DRINKING WATER	DATE: 10/19/16
RELING'D BY/SIGNATURE: Mike Golden With M	DATE/TIME: 10/19/16 1700
9.6 RECEIVED BY/SIGNATURE: John tangell	DATE/TIME: 10/20/16 1900
EMAIL RESULTS TO: CERK MAY Epbson, Com	TURN AROUND TIME: 7-10 days

		SAMPLE	ATA FO	DRM						
LAB	SAMPLE#	BUILDING	ROOM	LOCATION IN ROOM						
J.	SK- BLC- 001-F0	-C-DOI-FO BUND		Breakroom, 134 Floor, adjacent to						
2	SK-BLC- 002-FL	- COMMISSION		Richard Hoover Conference Room, acros						
				from Mail Room						
3	5K-BLC-003-FD			Student Course, adjacent to						
4	5K-BLC-004-FL			exercise Room, 18+ Floor, Kitchen Sink						
	WF BLC:005-FD			Water Fountain between news (wonens.						
6	WF-BLC-006-FL			buthroom, across from Kitchen (training						
_7	SK- SLC-007- FO			Bitchen (training), Kitchen						
8	BK-BLC-DOT-FL			SINK NORTHEAST (ST Floor						
9	5K-BLL-009-FO			Kitchen (training) Kitchen						
(0	St-BLC-010-FL			Sink Southeast, 15+ HOOR						
	SK-BLC-011-FU		-	Kitchen (training), Kitchen						
12	SK-BLC-02-FL		1	Sink, Southwest, 18+ Floor						
.3	EK-BLC - 013-Fi		<u> </u>	Kitcley (training) Kitcley Sink						
14	SK-BLL-014- FL	4	·	Northwest, 1st Floor						
815	WF- BLC - 015- FD	or psha		Water Fountain (upper)						
16	WF-BLC-016-FL	4		between Room 118 and 120						
	5K-BL6-017-FD	<u> </u>								
	EX-BU-OFF FL	•								
			<u> </u>							
	_	<u> </u>								
	1 16 0 1		-							
	* Halt full &	u								
		-	_							

	Associates SR-FL-0002-16					A6J2557	10/2	0/2016	
Sa	mple Integrity					PBSEN1939		0/2016 10	
BS	K Bottles: (Yes) No Page	e of _			The state of the s				
	Was temperature within range? Chemistry ≤ 6°C Micro < 10°C	1	(AN		re correct con				
9	If samples were taken today, is there evidence	1	X		received for the tests requested? Were there bubbles in the VOA vials?			۰۰, ۰۰۰ N	
COC Info	that chilling has begun?		NA)	(Vol	atiles Only)	Ye	es No (N		
00	Did all bottles arrive unbroken and intact?	Yes No				ount of sample rece		Yes) No	
O	Did all bottle labels agree with COC?  Was sodium thiosulfate added to CN sample(s)	Yes 1	/ok			hold time <72 hours	?	Yes (No	
	until chlorine was no longer present?	Yes No	NA)	PM:	s PM notified of o	By/Time:	Ye	es No (Ñ	
	250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V)	Checks	Pas	sed?	1-10	Zyr ranio.			
	Bacti Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	_		-					
	None (P)White Cap	_	-						
	Cr6 (P) Lt. Green Label/Blue Cap NH4OH(NH4)2SO4 DW	Cl, pH > 8	Y	N		· ·			
	Cr6 (P) Pink Label/Blue Cap NH40H(NH4)2SO4 WW	pH 9.3-9.7	Y	N					
the lab	Cr6 (P) Black Label/Blue Cap NH40H(NH4)2SO4 7199 ***24 HOUR HOLD TIME***	pH 9.0-9.5	Y	N					
.⊑	HNO <sub>3</sub> (P) Bed Cap or HCI (P) Purple Cap/Lt. Blue Label	_	_		10.				
performed	H <sub>2</sub> SO <sub>4</sub> (P) or (AG) Yellow Cap/Label	pH < 2	Υ	N					
John John John John John John John John	NaOH (P) Green Cap	Cl, pH >10	Y	N					
Be	NaOH + ZnAc (P)	pH > 9	Y	N					
are	Dissolved Oxygen 300ml (g)	pnza		iN					
A or									
<b>p</b> ≥	None (AG) 608/8081/8082, 625, 632/8321, 8151, 8270	<del>-</del>	-	-					
Receive are eith	HCI (AG) <sup>Lt, Blue Label</sup> O&G, Diesel		_	-					
	Ascorbic, EDTA, KH <sub>2</sub> Ct (AG) <sup>Pink Label</sup> 525	_	_	-					
	Na <sub>2</sub> O <sub>3</sub> S 250mL (AG) <sup>Neon Green Label</sup> 515	_	-						
Bottles ne checks	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 1 Liter (Brown P) 549	=	_	-					
Se cl	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (AG) <sup>Blue Label</sup> 548, THM, 524		-						
Chlorir	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (CG) <sup>Blue Label</sup> 504, 505, 547		# <u>===</u>	<u></u>					
5	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> + MCAA (CG) <sup>Orange Label</sup> 531	pH < 3	Υ	N					
ation	NH <sub>4</sub> Cl (AG) <sup>Purple Label</sup> 552	_		_					
e S	EDA (AG)Brown Label DBPs	_				And the second			
8F	HCL (CG) 524.2,BTEX,Gas, MTBE, 8260/624		APLE DR						
(n)	Buffer pH 4 (CG)								
nea	H <sub>3</sub> PO <sub>4</sub> (CG) <sup>Salmon Label</sup>								
	Other:								
= '	Asbestos 1Liter Plastic w/ Foil								
	Low Level Hg / Metals Double Baggie						AND AND ASSESSMENT OF THE PARTY		
_	Bottled Water		9 6						
	Clear Glass 250mL / 500mL / 1 Liter	-	_						
-	Soil Tube Brass / Steel / Plastic Tedlar Bag / Plastic Bag	_							
		 Time/Initials	7=		Container	Droom :=4::	+ D-1 -		
Split	s) P 250 \$	· intornitials	S	Р	Container	Preservative	Date/11	ime/Initials	
S	S P			Р			+		
Comments	* odd	numbe	-4		ly. Ruk	)	1		
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BSK Associates SR-FL-0002-16







10212016

PBSEN1939

Turnaround: Standard

Due Date: 11/3/2016



PBS Environmental





_		mpla lateau	~				ĺ	АбЈ2557		10/20/	<sup>2</sup> 016
J	a	mple Integr	Ity					PBSEN19		10	)
В	Sł	K Bottles: (Ýe	s No Page	e of							
		Was temperature within Chemistry ≤ 6°C M	n range?	T 7	(AV		re correct con			(44 i 114 i 11	
.	2	If samples were taken	today, is there evidence	<del>                                     </del>	$\sim$		eived for the testre there bubbles				., NA
-		that chilling has begun'	?		<u>(AY</u>	(Vo	latiles Only)			Yes	No (NA)
+ 3	<del>ر</del> ج	Did all bottles arrive un Did all bottle labels agr			10		s a sufficient am				es) No
	⊦ د		e added to CN sample(s)	4	, dy		samples have a s PM notified of			Y	≶s (No)
		until chlorine was no lo	nger present?	Yes No	VA/	PM		By/Time:	es:	Yes	No (NA)
	-	250ml(A) 500ml(B) 1Li	iter(C) 40ml VOA(V)	Checks	Pas	sed?	1-16				$\top$
	_ <u></u>	Bacti Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>									
	- ⊢	None (P)White Cap	,	_	_	_					
	<b>—</b>	Cr6 (P) Lt. Green Label/Blue Cap NH40H(NH4)2SO4 DW		CI, pH > 8	Υ	N					
	_ +	Cr6 (P) Pink Label/Blue Cap		pH 9.3-9.7	Υ	N					
	the lab	Cr6 (P) Black Label/Blue Ca	<sup>ap</sup> NH40H(NH4)2SO4 <b>7199</b> HOLD TIME***	pH 9.0-9.5	Y	N					
	<u>=</u>	HNO3 (P) BROWN OF HO	_	_	_	10%					
		H <sub>2</sub> SO <sub>4</sub> (P) or (A	G) Yellow Cap/Label	pH < 2	Υ	N					
P P	erfo	NaOH (P) Green Cap		Cl, pH >10	Υ	N				<u> </u>	1 200
	d e	NaOH + ZnAc (P)		pH > 9	Υ	N			2.1		
	Dissolved Oxygen 300ml (g)			<u></u>				-			
	≰	None (AG) 608/8081/808	2, 625, 632/8321, 8151, 8270								
/ed		HCI (AG)L1. Blue Label O		_				. 1. 2. 2		1 1 1 1 1 1 1 1 1	
Sei	=-	Ascorbic, EDTA, KH <sub>2</sub>									
Received	9	Na <sub>2</sub> O <sub>3</sub> S 250mL (AG) <sup>N</sup>		<u>.                                    </u>							1000000
Bottles R	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 1 Liter (Brown P) 549				<u> </u>	_					
oftl	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> (AG) <sup>Blue Label</sup> 548, THM, 524								i Line of the second	<del> </del>	
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	듯┝┈	VH <sub>4</sub> Cl (AG) <sup>Purple Label</sup>	<u> </u>	privo		18		<u>-ii</u>			
	>\ <u> </u>	EDA (AG)Brown Label D				-					
	<b>65</b>			_							
		HCL (CG) 524.2,BTEX,G	5as, M⊺BE, 8260/624			-					
	15	Buffer pH 4 (CG) H <sub>3</sub> PO <sub>4</sub> (CG)Salmon Label									
z	<u> </u>	13PO4 (CG)9441101 Lase:			-						
2	· -		stic w/ Foil	1 - <u>12 - 1</u> (-2.1)							F-5
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	<del></del>	Bottled Water			_						
	$\vdash$	Clear Glass 250mL								_	
		Soil Tube Brass / Plas edlar Bag / Plas	Steel / Plastic tic Bag				14 17 24				A National Inc.
	Ľ	Container	<del></del>	Time/Initials	_		Container	Dress			
Split	s	1	- Dato	rancanitiais	s	P	Container	Prese	rvative	Date/Tim	e/Initials
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_abe	L-d	by:@	Labels check	red by:		— ര		RUSH Pac			



Derek May PBS Environmental 4412 SW Corbett Ave Portland, OR 97239

RE: Report for A6J3200 Oregon DAS - Lead

Dear Derek May,

Thank you for using BSK Associates for your analytical testing needs. In the following pages, you will find the test results for the samples submitted to our laboratory on 10/27/2016. The results have been approved for release by our Laboratory Director as indicated by the authorizing signature below.

The samples were analyzed for the test(s) indicated on the Chain of Custody (see attached) and the results relate only to the samples analyzed. BSK certifies that the testing was performed in accordance with the quality system requirements specified in the 2009 TNI Standard. Any deviations from this standard or from the method requirements for each test procedure performed will be annotated alongside the analytical result or noted in the Case Narrative. Unless otherwise noted, the sample results are reported on an "as received" basis.

If additional clarification of any information is required, please contact your Project Manager, Debra Karlsson, at 559-497-2888.

Thanks again for using BSK Associates. We value your business and appreciate your loyalty.

Sincerely,

Debra Karlsson, Project Coordinator



Accredited in Accordance with NELAP ORELAP #4021

Page 1 of 9



#### **Case Narrative**

Project and Report Details Invoice Details

Client: PBS Environmental Invoice To: PBS Environmental Report To: Derek May Invoice Attn: Accounts Payable

Project #: Blind Commission #25103.003 PH 37 Project PO#: -

**Received:** 10/27/2016 - 15:10

**Report Due:** 11/10/2016

**Sample Receipt Conditions** 

Cooler:Default CoolerContainers IntactTemperature on Receipt °C:20.4COC/Labels Agree

Received with no thermal preservation. Sample(s) split after receipt at the laboratory.

Initial receipt at BSK-VAL

#### **Data Qualifiers**

The following qualifiers have been applied to one or more analytical results:

\*\*\*None applied\*\*\*

#### **Report Distribution**

 Recipient(s)
 Report Format
 CC:

 Derek May
 FINAL.RPT
 beth.powers@pbsenv.com

A6J3200 FINAL 11092016 1141

Printed: 11/9/2016





Blind Commission #25103.003 PH 37

## **Certificate of Analysis**

Sample ID: A6J3200-01 **Sample Date - Time:** 10/22/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-BLC-017-FD // 1st Floor lab kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A615042	11/01/16	11/02/16	



## BSK Associates Fresno Metals Quality Control Report

			<u> </u>							
Analyte	Result	RL	Units	Spike Level	Source Result	%REC	%REC Limits	RPD	RPD Limit	Date Analyzed Qual
		EPA 20	00.8 - Q	uality Co	ntrol					
Batch: A615042 Prep Method: EPA 200.2										Prepared: 11/1/2016 Analyst: GNG
Blank (A615042-BLK1)										
Lead	ND	0.0010	mg/L							11/02/16
Blank Spike (A615042-BS1)										
Lead	0.10	0.0010	mg/L	0.10		100	85-115			11/02/16
Blank Spike Dup (A615042-BSD1)										
Lead	0.10	0.0010	mg/L	0.10		102	85-115	2	20	11/02/16
Matrix Spike (A615042-MS1), Source	e: A6J3197-01									
Lead	0.20	0.0020	mg/L	0.20	ND	98	70-130			11/02/16
Matrix Spike (A615042-MS2), Source	e: A6J3200-01									
Lead	0.19	0.0020	mg/L	0.20	ND	95	70-130			11/02/16
Matrix Spike Dup (A615042-MSD1), \$	Source: A6J3197-01									
Lead	0.20	0.0020	mg/L	0.20	ND	99	70-130	1	20	11/02/16
Matrix Spike Dup (A615042-MSD2), \$	Source: A6J3200-01									
Lead	0.19	0.0020	mg/L	0.20	ND	96	70-130	1	20	11/02/16



#### **Certificate of Analysis**

#### Notes:

- The Chain of Custody document and Sample Integrity Sheet are part of the analytical report.
- Any remaining sample(s) for testing will be disposed of according to BSK's sample retention policy unless other arrangements are made in advance.
- All positive results for EPA Methods 504.1 and 524.2 require the analysis of a Field Reagent Blank (FRB) to confirm that the results are not a contamination error from field sampling steps. If Field Reagent Blanks were not submitted with the samples, this method requirement has not been performed.
- · Samples collected by BSK Analytical Laboratories were collected in accordance with the BSK Sampling and Collection Standard Operating Procedures.
- J-value is equivalent to DNQ (Detected, not quantified) which is a trace value. A trace value is an analyte detected between the MDL and the laboratory reporting limit. This result is of an unknown data quality and is only qualitative (estimated). Baseline noise, calibration curve extrapolation below the lowest calibrator, method blank detections, and integration artifacts can all produce apparent DNQ values, which contribute to the un-reliability of these values.
- (1) Residual chlorine and pH analysis have a 15 minute holding time for both drinking and waste water samples as defined by the EPA and 40 CFR 136. Waste water and ground water (monitoring well) samples must be field filtered to meet the 15 minute holding time for dissolved metals.
- Summations of analytes (i.e. Total Trihalomethanes) may appear to add individual amounts incorrectly, due to rounding of analyte values occurring before or after the total value is calculated, as well as rounding of the total value.
- RL Multiplier is the factor used to adjust the reporting limit (RL) due to variations in sample preparation procedures and dilutions required for matrix interferences.
- Due to the subjective nature of the Threshold Odor Method, all characterizations of the detected odor are the opinion of the panel of analysts. The characterizations can be found in Standard Methods 2170B Figure 2170:1.
- The MCLs provided in this report (if applicable) represent the primary MCLs for that analyte.

#### **Definitions**

mg/L:	Milligrams/Liter (ppm)	MDL:	Method Detection Limit	MDA95:	Min. Detected Activity
mg/Kg:	Milligrams/Kilogram (ppm)	RL:	Reporting Limit: DL x Dilution	MPN:	Most Probable Number
μg/L:	Micrograms/Liter (ppb)	ND:	None Detected at RL	CFU:	Colony Forming Unit
μg/Kg:	Micrograms/Kilogram (ppb)	pCi/L:	Picocuries per Liter	Absent:	Less than 1 CFU/100mLs
%:	Percent Recovered (surrogates)	RL Mult:	RL Multiplier	Present:	1 or more CFU/100mLs
NR:	Non-Reportable	MCL:	Maximum Contaminant Limit		

Please see the individual Subcontract Lab's report for applicable certifications.

WA100008-008

BSK is not accredited under the NELAP program for the following parameters:

\*\*NA\*\*

#### Certifications: Please refer to our website for a copy of our Accredited Fields of Testing under each certification.

Fresno			
State of California - ELAP	1180	State of Hawaii	4021
State of Nevada	CA000792016-1	State of Oregon - NELAP	4021
EPA - UCMR3	CA00079	State of Washington	C997-16
Sacramento		_	
State of California - ELAP	2435		
San Bernardino			
State of California - ELAP	2993	State of Oregon - NELAP	4119-001

State of Washington

A6J3200 FINAL 11092016 1141

Printed: 11/9/2016

Vancouver

State of Oregon - NELAP

C824-16

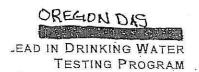


# Engineering + Environmental

A6J3200 PBSEN1939



10/27/2016



25103.003

	FACILITY NAME: BLIND COMMISSION	PROJECT#: PH 3
	ANALYSIS REQUESTED:  LEAD (PB) IN DRINKING WATER  COPPER (CU) IN DRINKING WATER	DATE: 10/22/16
	RELING'D BY/SIGNATURE: Mike Golden / Will De	DATE/TIME: 10/26/16 1300
0.4	RECEIVED BY/SIGNATURE:	DATE/TIME: 16/27/16 15:10
	EMAIL RESULTS TO: derek may Opbsenv. com	TURN AROUND TIME: 7-10 days

		SAMPLE	DATA FOR	M	
	SAMPLE#	BUILDING	ROOM	LOCATION IN ROOM	
	SK-BLC-017-FD			1st Floor, Lab, Kitchen Sink	
_	SK-BLC-017-FD SK-BLC-018-FL			Sink	
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A6J3200 PBSEN1939

10/27/2016

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RUSH Paged by:\_

BS	KB	ottles: (Ye	s No Pa	age of							and a second of
	Was	temperature within	range?	Yes No	(NA)	Wer	e correct contain	ולוס מווע פוסטט.	,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,		
0	Chen	nistry ≤ 6°C Mi	cro < 10°C	1	(3)	rece	eived for the tests	s requested?	4	Tes	INO NA
Info	that o	npies were taken t chilling has begun?	oday, is there evidenc	Yes No	(NA)	Wer	e there bubbles atiles Only)	2012 12250000000	Yes	No (NA	
200			oroken and intact?	Yes	No			ount of sample recei	rod2	Yes	\
္ပ	Did a	ll bottle labels agre	ee with COC?	(Yes	No			nold time <72 hours'		Yes	/
	Was sodium thiosulfate added to CN sample(s) until chlorine was no longer present?			Yes No	NA		PM notified of c			Yes	No (NA
			ter(C) 40ml VOA(V)	Checks	Pa	ssed?	1ーみ				
		Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub>		-							
	-	P)White Cap		_		-			473		
	Cr6 (	P) Lt. Green Label/Blue	Cap NH4OH(NH4)2SO4 D	W   Cl, pH > 8	3 Y	N					
	Cr6 (	P) Pink Label/Blue Cap	NH40H(NH4)2SO4 W	w pH 9.3-9.	7 Y	N					
the lab	Cr6 (P) Black Label/Blue Cap NH4OH(NH4)2SO4 7199  ***24 HOUR HOLD TIME***		99 pH 9.0-9.	5 Y	N						
	HNO <sub>3</sub> (P) Bed Cap or HCI (P) Purple Cap/Lt. Blue Label						IC				
are performed in	H <sub>2</sub> SO <sub>4</sub> (P) or (AG) Yellow Cap/Label			pH<2	Y	N N				9.0000	
ro Tuo		H (P) Green Cap			PROPERTY AND INCOME.	B Security Policy					
per				CI, pH >10	and salars				5000000		
are	NaOH + ZnAc (P)			pH > 9	Y	N					
ģ	Dissolved Oxygen 300ml (g)			_							
- 2 ×			2, 625, 632/8321, 8151, 8270			- 1					
eived either N/A	HCI (AG) <sup>L1. Blue Label</sup> O&G, Diesel			_	8	_					
Rec				-	123	_				E7-102 (A	
	Na <sub>2</sub> O <sub>3</sub> S 250mL (AG) <sup>Neon Green Label</sup> 515										
	Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> 1 Liter (Brown P) 549										
Bottles ne checks		2O <sub>3</sub> (AG) <sup>Blue Label</sup>									
Bott lorine ch	-	2O <sub>3</sub> (CG) Blue Label			OR BRIDE						
웆	AND DESCRIPTION	2O3 + MCAA (CG		-11 - 0	1				0 99.00		
ou/uo	AND GROUPS			pH < 3	Υ	N					
vati	Control Control	I (AG) <sup>Purple Label</sup>									
preser	EDA	(AG) <sup>Brown Label</sup> D	BPs			-					
b Dre	HCL (	(CG) 524.2,BTEX,0	Sas, MTBE, 8260/624	_		_					
aus	Buffe	r pH 4 (CG)				-					
me	H₃PO	4 (CG)Salmon Label		_							
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	Asbes		stic w/ Foil	- Lab <del>-</del> Lab							
		evel Hg / Metals	Double Baggie	_	-						
-		d Water	/ 5001 / 41:4-	<u> </u>							
	Soil T	Glass 250mL	/ 500mL / 1 Lite Steel / Plastic		05 607,615.0				0 000000000		
			tic Bag								
	<u>. odia</u>	Container		ate/Time/Initia	ls		Container	Preservative	Dot	VTir-	/ n 4 - -
Split	s) P	2501				6 P	Container	i reservative	Date	5/ 1 IITIE	e/Initials
S	SP	0.0				B P					7/0/10-1
Comments			* Odo	l numb			ily. Rik	)			
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Labeled by: \_\_\_\_\_ @ \_\_\_\_ Labels checked by: \_\_\_\_ @ \_\_\_\_







10282016

PBSEN1939

Turnaround: Standard

Due Date: 11/10/2016



PBS Environmental





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