

November 22, 2016

Jeremy Miller
Maintenance Manager
Department of Administrative Services
Enterprise Asset Management Division
1225 Ferry Street SE
Salem, Oregon 97301

Via email: Jeremy.W.MILLER@oregon.gov

Regarding: Drinking Water Sampling for Lead

Labor & Industries Building 350 Winter Street NE Salem, Oregon 97301

PBS Project # 25103.003 Phase 0016

Dear Mr. Miller:

On October 13 and 15, 2016, PBS Engineering and Environmental Inc. (PBS) performed drinking water sampling at the Labor & Industries building located at 350 Winter Street NE in Salem, Oregon. The testing was requested by State of Oregon Department of Administrative Services in an effort to ensure that concentrations of lead in drinking water remain below the EPA action level.

Sampling methodology and the interpretation of laboratory results were based on the EPA Lead and Copper Rule (LCR). Following LCR sampling guidelines, PBS collected the first 1000 milliliters (mL) of water from each test location (first draw) early in the morning following an overnight stagnation period. The LCR's stagnation period, and sampling protocol specifying the first 1000 mL samples, is designed to maximize the likelihood that the highest concentrations of lead are identified in water used for consumption. At each sample location, immediately following first draw sampling, a flush sample was collected after the water had been allowed to run for 30 seconds.

The water sampling process was supervised by a certified industrial hygienist (CIH) who is also an Oregon Health Authority certified lead risk assessor.

The action level set by the EPA for lead is 15 parts per billion (ppb). If the action level is exceeded in more than 10 percent of taps sampled, then action must be taken to control plumbing-material corrosion.

Thirty first draw and flush drinking water samples were collected and delivered under chain of custody to BSK Laboratories in Vancouver, Washington for lead analysis. Initially, only first draw samples were analyzed. Any first draw sample that exceeded the EPA action level for lead had its associated flush sample analyzed.

Concentrations of lead in the first draw samples ranged from none detected to 3 ppb. Laboratory analysis indicates that all of these drinking water samples contained lead at concentrations below the EPA action level of 15 ppb.

The following table presents all first draw sample locations and lead concentrations in ppb.

First Draw Drinking Water Sample Locations and Lead Concentrations

Sample Number	Sample Location	Lead Concentration (ppb)
SK-LAB-001-FD	Room 410 kitchenette fourth floor across from conference room 4B, sink	1.5
SK-LAB-003-FD	Room 440 kitchenette fourth floor across from 4-N-134 cubicle, sink	ND
SK-LAB-005-FD	Break room fourth floor commons kitchen sink	ND
SK-LAB-007-FD	Room 300 kitchenette third floor across from 3-N-200 cubicle, sink	ND
SK-LAB-009-FD	Room 330 kitchenette third floor across from conference room 3B kitchen sink break room	1.2
SK-LAB-011-FD	Break room third floor commons, kitchen sink	ND
SK-LAB-013-FD	Break room second floor commons, kitchen sink	ND
WF-LAB-015-FD	Water fountain first floor main lobby upper east near room 160	1.6
WF-LAB-017-FD	Water fountain first floor main lobby lower east near room 160	1.5
WF-LAB-019-FD	Water fountain first floor main lobby upper west near room 150	3.0
WF-LAB-021-FD	Water fountain first floor main lobby lower east near room 150	2.5
SK-LAB-023-FD	Break room first floor treasury department kitchen sink	ND
SK-LAB-025-FD	Room 1 custodial break room basement kitchen sink	2.6
SK-LAB-027-FD	Cafe kitchen sink bake station	ND
SK-LAB-029-FD	Cafe kitchen sink food prep 1	2.3

ND: None Detected

Please refer to the attached Chain of Custody form and laboratory data for greater details. It should be noted that quality control (QC) sample results are included at the end of laboratory information. The QC samples are both laboratory blanks and spiked samples used internally by the laboratory to assess accuracy.

Please feel free to contact me at 503.417.7602 or derek.may@pbsenv.com with any questions or comments.

Sincerely,

PBS Engineering and Environmental Inc.

Derek May, Principal

Attachments: Laboratory Results
Chain of Custody Form

DM::bmp

The information contained in this document is proprietary and shall not be duplicated, used, or disclosed in whole or in part to other parties without the permission of PBS.



Derek May PBS Environmental 4412 SW Corbett Ave Portland, OR 97239

RE: Report for A6J2006 Oregon DAS - Lead

Dear Derek May,

Thank you for using BSK Associates for your analytical testing needs. In the following pages, you will find the test results for the samples submitted to our laboratory on 10/17/2016. The results have been approved for release by our Laboratory Director as indicated by the authorizing signature below.

The samples were analyzed for the test(s) indicated on the Chain of Custody (see attached) and the results relate only to the samples analyzed. BSK certifies that the testing was performed in accordance with the quality system requirements specified in the 2009 TNI Standard. Any deviations from this standard or from the method requirements for each test procedure performed will be annotated alongside the analytical result or noted in the Case Narrative. Unless otherwise noted, the sample results are reported on an "as received" basis.

If additional clarification of any information is required, please contact your Project Manager, Debra Karlsson, at 559-497-2888.

Thanks again for using BSK Associates. We value your business and appreciate your loyalty.

Sincerely,

Debra Karlsson, Project Coordinator



Accredited in Accordance with NELAP ORELAP #4021



Case Narrative

Project and Report Details Invoice Details

Client: PBS Environmental Invoice To: PBS Environmental Report To: Derek May Invoice Attn: Accounts Payable

Project #: Labor & Industries #25103.003 PH 16 Project PO#: -

Received: 10/17/2016 - 16:30 **Report Due:** 10/31/2016

Sample Receipt Conditions

Cooler:Default CoolerContainers IntactTemperature on Receipt °C: 20.5COC/Labels Agree

Received with no thermal preservation. Sample(s) split after receipt at the laboratory.

Initial receipt at BSK-VAL

Data Qualifiers

The following qualifiers have been applied to one or more analytical results:

Report Distribution

Recipient(s) Report Format CC:

Derek May FINAL.RPT

^{***}None applied***





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-01 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-LAB-001-FD // Room 410 Kitchenette 4th Floor across from

conference room 4B, kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0015	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-03 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-LAB-003-FD // Room 440 Kitchenette 4th Floor across from

Sample Type: First Draw

4-N-134 cubicle, kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-05 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-LAB-005-FD // Breakroom 4th Floor commons kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-07 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-LAB-007-FD // Room 300 Kitchenette 3rd Floor across from

3-N-200 cubicle, kitchen sink

BSK Associates Fresno

Metals

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-09 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-LAB-009-FD // Room 330 Kitchenette 3rd Floor across from

Sample Type: First Draw

conference room 3B kitchen sink breakroom

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual	
Lead	EPA 200.8	0.0012	0.0010	mg/L	1	A614462	10/20/16	10/20/16	_





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-11 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-LAB-011-FD // Breakroom 3rd Floor commons, kitchen sink

Sample Type: First Draw

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-13 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-LAB-013-FD // Breakroom 2nd Floor commons, kitchen sink

Sample Type: First Draw

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-15 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: WF-LAB-015-FD // Water fountain 1st Floor main lobby upper

Sample Type: First Draw

East near Rm 160

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	0.0016	0.0010	mg/L	1	A614462	10/20/16	10/20/16





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-17 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: WF-LAB-017-FD // Water fountain 1st Floor main lobby lower

Sample Type: First Draw

East near Rm 160

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0015	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-19 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: WF-LAB-019-FD // Water fountain 1stFloor main lobby upper

Sample Type: First Draw

West near Rm 150

	Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
•	Lead	EPA 200.8	0.0030	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-21 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: WF-LAB-021-FD // Water fountain 1st Floor main lobby lower

Sample Type: First Draw

East near Rm 150

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0025	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-23 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Description: SK-LAB-023-FD // Breakroom 1st Floor treasury department

Sample Type: First Draw

kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A614462	10/20/16	10/20/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J2006-25 **Sample Date - Time:** 10/13/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-LAB-025-FD // Room 1 Custodial breakroom basement

kitchen sink

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed Qual
Lead	EPA 200.8	0.0026	0.0010	mg/L	1	A614462	10/20/16	10/20/16



BSK Associates Fresno Metals Quality Control Report

					P						
Analyte	Result	RL	Units	Spike Level	Source Result	%REC	%REC Limits	RPD	RPD Limit	Date Analyzed	Qual
		EPA 20	00.8 - Q	uality Co	ntrol						
Batch: A614462										Prepared:	10/20/2016
Prep Method: EPA 200.2 - Pb/Cu R	ule									Ar	alyst: GNG
Blank (A614462-BLK1)											
Lead	ND	0.0010	mg/L							10/20/16	
Blank Spike (A614462-BS1)											
Lead	0.10	0.0010	mg/L	0.10		104	85-115			10/20/16	
Blank Spike Dup (A614462-BSD1)											
Lead	0.10	0.0010	mg/L	0.10		104	85-115	1	20	10/20/16	
Matrix Spike (A614462-MS1), Sourc	ce: A6J1309-06										
Lead	0.21	0.0020	mg/L	0.20	ND	105	70-130			10/20/16	
Matrix Spike (A614462-MS2), Sourc	ce: A6J2006-13										
Lead	0.20	0.0020	mg/L	0.20	ND	102	70-130			10/20/16	
Matrix Spike Dup (A614462-MSD1),	, Source: A6J1309-06										
Lead	0.21	0.0020	mg/L	0.20	ND	106	70-130	1	20	10/20/16	
Matrix Spike Dup (A614462-MSD2),	, Source: A6J2006-13										
Lead	0.20	0.0020	mg/L	0.20	ND	101	70-130	1	20	10/20/16	



Certificate of Analysis

Notes:

- The Chain of Custody document and Sample Integrity Sheet are part of the analytical report.
- Any remaining sample(s) for testing will be disposed of according to BSK's sample retention policy unless other arrangements are made in
- All positive results for EPA Methods 504.1 and 524.2 require the analysis of a Field Reagent Blank (FRB) to confirm that the results are not a contamination error from field sampling steps. If Field Reagent Blanks were not submitted with the samples, this method requirement has not been performed.
- Samples collected by BSK Analytical Laboratories were collected in accordance with the BSK Sampling and Collection Standard Operating Procedures.
- J-value is equivalent to DNQ (Detected, not quantified) which is a trace value. A trace value is an analyte detected between the MDL and the laboratory reporting limit. This result is of an unknown data quality and is only qualitative (estimated). Baseline noise, calibration curve extrapolation below the lowest calibrator, method blank detections, and integration artifacts can all produce apparent DNQ values, which contribute to the un-reliability of these values.
- (1) Residual chlorine and pH analysis have a 15 minute holding time for both drinking and waste water samples as defined by the EPA and 40 CFR 136. Waste water and ground water (monitoring well) samples must be field filtered to meet the 15 minute holding time for dissolved
- Summations of analytes (i.e. Total Trihalomethanes) may appear to add individual amounts incorrectly, due to rounding of analyte values occurring before or after the total value is calculated, as well as rounding of the total value.
- RL Multiplier is the factor used to adjust the reporting limit (RL) due to variations in sample preparation procedures and dilutions required for matrix interferences.
- Due to the subjective nature of the Threshold Odor Method, all characterizations of the detected odor are the opinion of the panel of analysts. The characterizations can be found in Standard Methods 2170B Figure 2170:1.
- The MCLs provided in this report (if applicable) represent the primary MCLs for that analyte.

Definitions

mg/L:	Milligrams/Liter (ppm)	MDL:	Method Detection Limit	MDA95:	Min. Detected Activity
mg/Kg:	Milligrams/Kilogram (ppm)	RL:	Reporting Limit: DL x Dilution	MPN:	Most Probable Number
μg/L:	Micrograms/Liter (ppb)	ND:	None Detected at RL	CFU:	Colony Forming Unit
μg/Kg:	Micrograms/Kilogram (ppb)	pCi/L:	Picocuries per Liter	Absent:	Less than 1 CFU/100mLs
%:	Percent Recovered (surrogates)	RL Mult:	RL Multiplier	Present:	1 or more CFU/100mLs
NR:	Non-Reportable	MCL:	Maximum Contaminant Limit		

Please see the individual Subcontract Lab's report for applicable certifications.

BSK is not accredited under the NELAP program for the following parameters: **NA**

Certifications: Please refer to our website for a copy of our Accredited Fields of Testing under each certification.

Fresn	10
State	of

State of California - ELAP	1180	State of Hawaii	4021
State of Nevada	CA000792016-1	State of Oregon - NELAP	4021
EPA - UCMR3	CA00079	State of Washington	C997-16
Sacramento			

State of California - ELAP 2435

San Bernardino

State of California - ELAP 2993 State of Oregon - NELAP 4119-001

Vancouver

State of Washington State of Oregon - NELAP WA100008-008 C824-16

A6J2006 FINAL 10252016 1750

Printed: 10/25/2016

QA-RP-0001-10 Final.rpt



Engineering + Environmental

A6J2006 PBSEN1939 10/17/2016 10





25103.003

FACILITY NAME: LABOR & WOUSTRIES	PROJECT#: PH N
ANALYSIS REQUESTED: LEAD (PB) IN DRINKING WATER COPPER (CU) IN DRINKING WATER	DATE: 10 13 16
RELING'D BY/SIGNATURE: Mike Golden / Will De	DATE/TIME: 10 13/16 1400
RECEIVED BY/SIGNATURE: Jens Kangell	DATE/TIME: 10 17 16 1630
EMAIL RESULTS TO: derek may Epbseny com	TURN AROUND TIME: 7-10 days

SAMPLE DATA FORM										
LAB	SAMPLE#	BUILDING	ROOM	LOCATION IN ROOM						
i	SK-LAB-001-FD.		410	Kitchenotte 4th Floor, Across						
2	SK-LA8-002-FL		4	From conference room 4B Kitchen Sink						
3	SIL- LAB- 003-FD		440	Kitchenethe, 4th Floor Across						
4	SIC-LAB-004-FL		4	from 4-N-134 (cubicle) Kitchen Style						
5	SIC- LAB- 005- FD		The second secon	Breakroom (4th Floor commons) 4th						
i,	SK-LAB-006-FL			Floor, Kitchen Sink						
7	SK-48-007-FD		300	Kitchenette, 3rd Floor, Across						
8	5K-LAB-008-FL		4	from 3-N-200 (cubicle) Kitchen S. at						
9	SK-LAB-009-FU	was a second or district	330	Kitchenette, 3rd Floor, Across						
10	6K-LAB-010-FL		4	from Conference Room 3B, Kritier State						
11	SK- LAB- 011- FO			Break room (3rd Floor Communs)						
12	5K-LAB-012-FL		34	Floor, Kitchen Sink						
13	SK-LAB-OB-FO			Breakroom (2nd Floor Commons)						
14	SK-LAB-014-FL			2nd Floor Kitchen Sink						
15	WF-LAB-015-FD			Water Fountain, 1st Floor, Main						
16	WF- LAB- 016-FL			Lobby (Uppor) East (new Rom 16)						
17	WF- LAB- 017-FO			Worker Fountain, 1st Floor Main						
18	WF- LAB- 015-FL			Lobby (Lower) He East (near						
19	WF-LA8-019-FO			water Fourtain 1st Floor, Main.						
20	WF-LAS 00-FU	,		Lubby Cupper West (new Ry 150						
21	WF-LAB-021-F0	NOTE OF THE PARTY		Woder Fountain 1st Floor, Moun						
22	WF-LAS 022-FI	<u> </u>		Lobber (10000) East (new Rm 150)						
23	SK-LAB. 023-FD			Breakroom, 184 Place Treasury						
24	SK-LAB-DU-FL			Deportment Kitchen Sink						
25	3K-LAB-025-FD		1	(ustadial Break room, Basements						
26	SK-LAB. 026-FL		4	KHELEN SINK						

Sample Integrity

A6J2006 PBSEN1939

10/17/2016

	ok Bottles: Yes No Page	\ of		Y4			nar ii a	
	Was temperature within range? Chemistry ≤ 6°C Micro < 10°C	Yes No (N.	A) Wei	e correct conta	iners and pres	ervatives	(Yes)	No NA
Info	If samples were taken today, is there evidence that chilling has begun?	Yes No (N	Wer	eived for the tes	in the VOA vi	als?	Yes	-
S	Did all bottles arrive unbroken and intact?	Yes No		atiles Only)	ount of commit			NO (NA
202	Did all bottle labels agree with COC?	Yes No	Do s	s a sufficient am samples have a	hold time <72	e received?		
×	Was sodium thiosulfate added to CN sample(s)	F	1 Was	PM notified of	discrepancies	7	Yes	No No
====	until chlorine was no longer present?	Yes No (N	PM:		By/Time:	Ť.	Yes	No (NA
	250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V)	Checks	Passed?	1-26				
	Bacti Na ₂ S ₂ O ₃	· (1) · (1)		ter il				
	None (P)White Cap		_					From the many region
	Cr6 (P) Lt Green Label/Blue Cap NH40H(NH4)2SO4 DW	. N. 2	Ϋ́N				10-0-57	3 PA 35
	Cr6 (P) Pink Label/Blue Cap NH4OH(NH4)2SO4 WW	pH 9.3-9.7	Y N					
4.7 4.7	C16 (P) Black Label/Blue Cap: NH4OH(NH4)2SO4 7199	pH 9.0-9.5	Y N					
2.	HNO3 (P) Red Copy OF HCI (P) Purple Cap/Lt. Blue Label	_	_	10				
bombo	H ₂ SO ₄ (P) or (AG) Yellow Cap/Label	pH×2	Y N	ER EFERS	167 Y 78 15	SECTION ES	E VEVI	6. T. P. D. L.
Š	NaOH (P) Green Cap	Cl, pH >10	YN					
č	NaOH + ZnAc (P)	.e < Hq	YN			- 1.0 to 1.5	S Office Act	र्वे को हाता हुन हुन है। इ.स.च्या का स्थापन
9	Dissolved Oxygen 300ml (g)	_						
Α/	None (AG) 608/8081/8082, 525, 632/8321, 8/151, 8270		3 <u>5</u> 5 3	to the second	Call San As	44 (18 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8		Paragraph of the Control of the Control
Ved	HCI (AG)LI Blue Label O&G, Diesel			The state of the s	238 A 10 10	STATE OF THE STATE	2000年	Property of
ei v	Ascorbic, EDTA, KH ₂ Ct (AG) ^{Pink Label} 525							
Received	Na ₂ O ₃ S 250mL (AG)Neon Green Label -515		- (*	randon (Sana)	terypole was to get	14.12.12.17.17.16.		
	Na ₂ S ₂ O ₃ 1 Liter (Brown P) 549		\$ See \$ 256		No. 10		4.17	
Bottles ne check	Na ₂ S ₂ O ₃ (AG) ^{Blue Label} 548, 111M, 524				5			
80	Na ₂ S ₂ O ₃ (CG) ^{Blue Label} 504, 505, 547	4. () 1. (() ()	<u> </u>			6-25-34	14-11	
hlor	And the late of th	-		on d'ann consider	Sheet 1275 24 4 4 12 2			
J/uo	Na ₂ S ₂ O ₃ + MCAA (CG)Orange Label 533	pH<3	Y. N		证证证法			
rvat	NH4CI (AG) ^{Purple Label} 552							
ese	EDA (AG)Brown Label DBPs	-	25-1					
Spr	HCL (CG) 524.2,BTEX,Gas, MTBE, 8260/624	and the second second second						
mean	Buffer pH 4 (GG)		i			6-1-1	34.34	18 433
Ε.	H ₃ PO ₄ (CG)Salmon tabel	学生 沙里长	V-13		经制度			
اء	Other: Asbestos 1Liter Plastic w/ Foil	0 (EAST 1-15-11	77 T 1947 T 1 T 1 T 1 T 1 T 1 T 1 T 1 T 1 T 1 T				
	Low Level Hg / Metals Double Baggie		_		terspiele in		200	
	Bottled Water							
	Clear Glass 250mL / 500mL / 1 Liter		_	X or or or or or		14 - 1 (1) (1) (1) (1)	980,86	
	Soil Tube Brass / Steel / Plastic			*****	2 141 141 141 141	30.24 (4.5)	To be designed	
	Tedlar Bag / Plastic Bag	_	-1			202 11 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		5 A. S.
4		ime/Initials		Container	Preserva	alive Da	ite/Time	/Initials
Split	s)p 250%		SP				ici iniie	/IIIIIais
0,	SP		SP					
Comments	+ odd num	bens ou	nly.	RIN				



Derek May PBS Environmental 4412 SW Corbett Ave Portland, OR 97239

RE: Report for A6J3318 Oregon DAS - Lead

Dear Derek May,

Thank you for using BSK Associates for your analytical testing needs. In the following pages, you will find the test results for the samples submitted to our laboratory on 10/26/2016. The results have been approved for release by our Laboratory Director as indicated by the authorizing signature below.

The samples were analyzed for the test(s) indicated on the Chain of Custody (see attached) and the results relate only to the samples analyzed. BSK certifies that the testing was performed in accordance with the quality system requirements specified in the 2009 TNI Standard. Any deviations from this standard or from the method requirements for each test procedure performed will be annotated alongside the analytical result or noted in the Case Narrative. Unless otherwise noted, the sample results are reported on an "as received" basis.

If additional clarification of any information is required, please contact your Project Manager, Debra Karlsson, at 559-497-2888.

Thanks again for using BSK Associates. We value your business and appreciate your loyalty.

Sincerely,

Debra Karlsson, Project Coordinator



Accredited in Accordance with NELAP
ORELAP #4021

Page 1 of 9



Case Narrative

Project and Report Details Invoice Details

Client: PBS Environmental Invoice To: PBS Environmental Report To: Derek May Invoice Attn: Accounts Payable

Project #: Labor & Industries #25103.003 PH 16 Project PO#: -

Received: 10/26/2016 - 17:15

Report Due: 11/10/2016

Sample Receipt Conditions

Cooler:Default CoolerContainers IntactTemperature on Receipt °C:20.4COC/Labels Agree

Received with no thermal preservation. Sample(s) split after receipt at the laboratory.

Initial receipt at BSK-VAL

Data Qualifiers

The following qualifiers have been applied to one or more analytical results:

Report Distribution

 Recipient(s)
 Report Format
 CC:

 Derek May
 FINAL.RPT
 beth.powers@pbsenv.com

A6J3318 FINAL 11092016 1140

Printed: 11/9/2016

^{***}None applied***





Sampled By: Client

Oregon DAS - Lead

Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J3318-01 **Sample Date - Time:** 10/15/16 - 00:00

Matrix: Drinking Water

Sample Description: SK-LAB-027-FD // Cafe kitchen sink bake station Sample Type: First Draw

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	ND	0.0010	mg/L	1	A615129	11/02/16	11/02/16	





Labor & Industries #25103.003 PH 16

Certificate of Analysis

Sample ID: A6J3318-03 **Sample Date - Time:** 10/15/16 - 00:00 Sampled By: Client

Matrix: Drinking Water

Sample Type: First Draw Sample Description: SK-LAB-029-FD // Cafe kitchen sink food prep 1

BSK Associates Fresno

Metals

Analyte	Method	Result	RL	Units	RL Mult	Batch	Prepared	Analyzed	Qual
Lead	EPA 200.8	0.0023	0.0010	mg/L	1	A615129	11/02/16	11/02/16	



BSK Associates Fresno Metals Quality Control Report

Analyte	Result	RL	Units	Spike Level	Source Result	%REC	%REC Limits	RPD	RPD Limit	Date Analyzed Qual
		EPA 20	00.8 - Q	uality Co	ntrol					
Batch: A615129 Prep Method: EPA 200.2										Prepared: 11/2/2016 Analyst: GNG
Blank (A615129-BLK1)										
Lead	ND	0.0010	mg/L							11/02/16
Blank Spike (A615129-BS1)										
Lead	0.098	0.0010	mg/L	0.10		98	85-115			11/02/16
Blank Spike Dup (A615129-BSD1)										
Lead	0.098	0.0010	mg/L	0.10		98	85-115	0	20	11/02/16
Matrix Spike (A615129-MS1), Source:	A6J3315-03									
Lead	0.19	0.0020	mg/L	0.20	ND	97	70-130			11/02/16
Matrix Spike Dup (A615129-MSD1), S	ource: A6J3315-03	.								
Lead	0.19	0.0020	mg/L	0.20	ND	95	70-130	3	20	11/02/16



Certificate of Analysis

Notes:

- The Chain of Custody document and Sample Integrity Sheet are part of the analytical report.
- Any remaining sample(s) for testing will be disposed of according to BSK's sample retention policy unless other arrangements are made in advance.
- All positive results for EPA Methods 504.1 and 524.2 require the analysis of a Field Reagent Blank (FRB) to confirm that the results are not a contamination error from field sampling steps. If Field Reagent Blanks were not submitted with the samples, this method requirement has not been performed.
- Samples collected by BSK Analytical Laboratories were collected in accordance with the BSK Sampling and Collection Standard Operating Procedures.
- J-value is equivalent to DNQ (Detected, not quantified) which is a trace value. A trace value is an analyte detected between the MDL and the laboratory reporting limit. This result is of an unknown data quality and is only qualitative (estimated). Baseline noise, calibration curve extrapolation below the lowest calibrator, method blank detections, and integration artifacts can all produce apparent DNQ values, which contribute to the un-reliability of these values.
- (1) Residual chlorine and pH analysis have a 15 minute holding time for both drinking and waste water samples as defined by the EPA and 40 CFR 136. Waste water and ground water (monitoring well) samples must be field filtered to meet the 15 minute holding time for dissolved metals.
- Summations of analytes (i.e. Total Trihalomethanes) may appear to add individual amounts incorrectly, due to rounding of analyte values occurring before or after the total value is calculated, as well as rounding of the total value.
- · RL Multiplier is the factor used to adjust the reporting limit (RL) due to variations in sample preparation procedures and dilutions required for matrix interferences.
- Due to the subjective nature of the Threshold Odor Method, all characterizations of the detected odor are the opinion of the panel of analysts. The characterizations can be found in Standard Methods 2170B Figure 2170:1.
- The MCLs provided in this report (if applicable) represent the primary MCLs for that analyte.

Definitions

mg/L:	Milligrams/Liter (ppm)	MDL:	Method Detection Limit	MDA95:	Min. Detected Activity
mg/Kg:	Milligrams/Kilogram (ppm)	RL:	Reporting Limit: DL x Dilution	MPN:	Most Probable Number
μg/L:	Micrograms/Liter (ppb)	ND:	None Detected at RL	CFU:	Colony Forming Unit
μg/Kg:	Micrograms/Kilogram (ppb)	pCi/L:	Picocuries per Liter	Absent:	Less than 1 CFU/100mLs
%:	Percent Recovered (surrogates)	RL Mult:	RL Multiplier	Present:	1 or more CFU/100mLs
NR:	Non-Reportable	MCL:	Maximum Contaminant Limit		

Please see the individual Subcontract Lab's report for applicable certifications.

WA100008-008

BSK is not accredited under the NELAP program for the following parameters: **NA**

Certifications: Please refer to our website for a copy of our Accredited Fields of Testing under each certification.

Fresno			
State of California - ELAP	1180	State of Hawaii	4021
State of Nevada	CA000792016-1	State of Oregon - NELAP	4021
EPA - UCMR3	CA00079	State of Washington	C997-16
Sacramento			
State of California - ELAP	2435		
San Bernardino			
State of California - ELAP	2993	State of Oregon - NELAP	4119-001
		=	

A6J3318 FINAL 11092016 1140

Printed: 11/9/2016

Vancouver

State of Oregon - NELAP

QA-RP-0001-10 Final.rpt

State of Washington

C824-16

A6J3318 PBSEN1939 10 26 2016 10





Engineering +
Environmental

FACILITY NAME: LABOR & INJOUSTRIES	25103.003 PROJECT#: PHIL
Analysis requested:	3
LEAD (PB) IN DRINKING WATER	DATE: 10 15 2016
COPPER (CU) IN DRINKING WATER	30 100 100
RELINO'D BY/SIGNATURE: Mike Golden / Will All	DATE/TIME: 10 21/2016 1500
RECEIVED BY/SIGNATURE:	DATESTIME: 10/26/16 5:15PM
EMAIL RESULTS TO: derek may Eposeny com	TURN AROUND TIME: 7-10 days

			DATA FO	T
LAB	SAMPLE#	BUILDING	ROOM	LOCATION IN ROOM
	SK-LAB-027-FD		CAFE	Kitchen Sink - Bato Stocking
2_	SK-LAB-039-FL SK-LAB-039-FO SK-LAB-030-FC			Kitchen Sink - Bake Stachion Kitchen Sink - Food Prop #1
3	5K-LAB-029-FD			Kitchen Sink - Food Pron #1
4	SK-148-030-FC		4	*
-		4 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -		
MARCHAELES DISE				
==				
-		The state of the s	1	
			1.	
40				
*		- Marie		
		200		
		-		

A6J3318 PBSEN1939 10/26/2016

10

BS	SK Bottles: (Yes) No Page	e of	1						
COC Info	Was temperature within range? Chemistry ≤ 6°C Micro < 10°C	Yes No (NA)	We	re correct contain	ners and preservativ	es (Yes	No. NA
	If samples were taken today, is there evidence	Yes No (\prec	We	eived for the test re there bubbles	s requested? in the VOA vials?		\sim	No NA
	that chilling has begun? Did all bottles arrive unbroken and intact?	- A		(Vo	latiles Only)				No (NA
	Did all bottle labels agree with COC?		10 10	Wa	s a sufficient am	ount of sample recei	ved?	Yes	
	Was sodium thiosulfate added to CN sample(s) until chlorine was no longer present?		VA	Wa	s PM notified of a		,	Yes	\rightarrow
	250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V)	Checks	T Bas	PM sed?	: 	By/Time:		165	No NA
	Bacti Na ₂ S ₂ O ₃	Officers	ras	seu!				1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	
	None (P)White Cap								
	Cr6 (P) Lt. Green Label/Blue Cap NH4OH(NH4)2SO4 DW	CI, pH > 8	Y	- NI			The Department	III locality in the	anti-interaction
	Cr6 (P) Pink Label/Blue Cap NH40H(NH4)2SO4 WW		10000000000	N					
4 0 7	* * * * * * * * * * * * * * * * * * * *	pH 9.3-9.7 pH 9.0-9.5	Y	N					•
1 4	***24 HOUR HOLD TIME***	pri 3.0-3.3	ľ	N					
2.	HNO ₃ (P) Red Cap or HCI (P) Purple Cap/Lt. Blue Label	_			10		1		
Derformed	H ₂ SO ₄ (P) or (AG) Yellow Cap/Label	pH < 2	Y	N					
	NaOH (P) Green Cap	CI, pH >10	Υ	N			2/2/3/8		
9	NaOH + ZnAc (P)	pH > 9	Υ	N					
0.00	Dissolved Oxygen 300ml (g)	_	_	_	W		+		
 ₹			_	JEST!					To 35 8 12 17 8 1 5
le d		_							
Seive	Ascorbic, EDTA, KH ₂ Ct (AG) ^{Pink Label} 525								
Received	Na ₂ O ₃ S 250mL (AG) ^{Neon Green Label} 515								
		7.3		-					
Bottles ne checks	Na ₂ S ₂ O ₃ (AG) ^{Blue Label} 548, THM, 524		_						
B ig	Na ₂ S ₂ O ₃ (CG) ^{Blue Label} 504, 505, 547			-					
lo di		Care Company of the							
ervation/ch	Na ₂ S ₂ O ₃ + MCAA (CG) ^{Orange Label} 531	pH < 3	Υ	N					
vati	NH ₄ Cl (AG) ^{Purple Label} 552	-	-	-					
sser	2574(10)			-					
pre	HCL (CG) 524.2,BTEX,Gas, MTBE, 8260/624	-	= ***					Control of	
ans	Buffer pH 4 (CG)								
, E	H ₃ PO ₄ (CG) ^{Salmon Label}								
اا	Other:								
	Asbestos 1Liter Plastic w/ Foil		-	•					
	Low Level Hg / Metals Double Baggie Bottled Water			- ALVENTED I					1.200
1	Clear Glass 250mL / 500mL / 1 Liter	<u> </u>	=						
Ī	Soil Tube Brass / Steel / Plastic						95000000	Carlos and	C Williams
	Tedlar Bag / Plastic Bag	_	_						
<u>=</u>		Time/Initials			Container	Preservative	Date	L /Time/	/Initials
Split	(s)P 250*		S	Р				* 1 111101	- Indiano
٠,	S P		S	Р				1000	
Comments	* Odd numbe	rs onl	y-	Re	R		L		
Labele	ed by:@ Labels check	ked by:		 @	F	RUSH Paged by:		Pag	e 8 of 9

WO10019 10/26/2016 Sample Integrity PBSEN1939 BSK Bottles: (Yes No Page of. Was temperature within range? Were correct containers and preservatives Yes No (NA Chemistry ≤ 6°C Micro < 10°C received for the tests requested? No NA Info If samples were taken today, is there evidence Were there bubbles in the VOA vials? Yes No (NA that chilling has begun? No (NA (Volatiles Only) Yes (Yes Did all bottles arrive unbroken and intact? No Was a sufficient amount of sample received? Yes Nα Did all bottle labels agree with COC? Yes No Do samples have a hold time <72 hours? No. Yes Was sodium thiosulfate added to CN sample(s) Was PM notified of discrepancies? No NA Yes until chlorine was no longer present? Yes NA 250ml(A) 500ml(B) 1Liter(C) 40ml VOA(V) Checks Passed? Bacti Na₂S₂O₃ None (P)White Cap Cr6 (P) Lt. Green Label/Blue Cap NH40H(NH4)2SO4 DW CI, pH > 8 YN Cr6 (P) Pink Label/Blue Cap NH40H(NH4)2SO4 WW pH 9.3-9.7 Υ Cr6 (P) Black Label/Blue Cap NH4OH(NH4)2SO4 7199 pH 9.0-9.5 N ***24 HOUR HOLD TIME*** HNO₃ (P) Red Cap or HCI (P) Purple Cap/Lt, Blue Label (C) are performed H₂SO₄ (P) or (AG) Yellow Cap/Label pH < 2 Y N NaOH (P) Green Cap CI, pH >10 Y Ν NaOH + ZnAc (P) pH > 9 Y N Dissolved Oxygen 300ml (g) ö None (AG) 608/8081/8082, 625, 632/8321, 8151, 8270 Received HCI (AG)Lt. Blue Label O&G, Diesel Ascorbic, EDTA, KH2Ct (AG)Pink Label 525 Na₂O₃S 250mL (AG)Neon Green Label 515 Bottles Na₂S₂O₃ 1 Liter (Brown P) 549 Na₂S₂O₃ (AG)^{Blue Label} 548, THM, 524 Na₂S₂O₃ (CG) Blue Label 504, 505, 547 Na₂S₂O₃ + MCAA (CG)^{Orange Label} 531 pH < 3YN NH₄CI (AG)^{Purple Label} 552 EDA (AG)Brown Label DBPs HCL (CG) 524.2,BTEX,Gas, MTBE, 8260/624 Buffer pH 4 (CG) H₃PO₄ (CG)Salmon Label Other: Asbestos 1Liter Plastic w/ Foil Low Level Hg / Metals Double Baggie Bottled Water Clear Glass 250mL / 500mL / 1 Liter Soil Tube Brass / Steel / Plastic Tedlar Bag Plastic Bag Container Preservative Date/Time/Initials Preservative Container Date/Time/Initials S/P 250¥ * Odd numbers only. Rea Comments All containers received intact no 11-1 Labeled by: _____ @ ____ Labels checked by: _____ @ ____ RUSH Paged by:____ Page 9 of 9