REGULATORY REQUIREMENTS

- Plan Review
 - Pilot Study
 - Approval to Construct
 - Final Approval
- 2. Operator Certification
 - Water Treatment 1 (Typical)
- 3. Monitoring
 - Chlorine/CT
 - Turbidity
- Reporting/Recordkeeping
 - Monthly Reporting (NTU, Chlorine, CT, etc.)

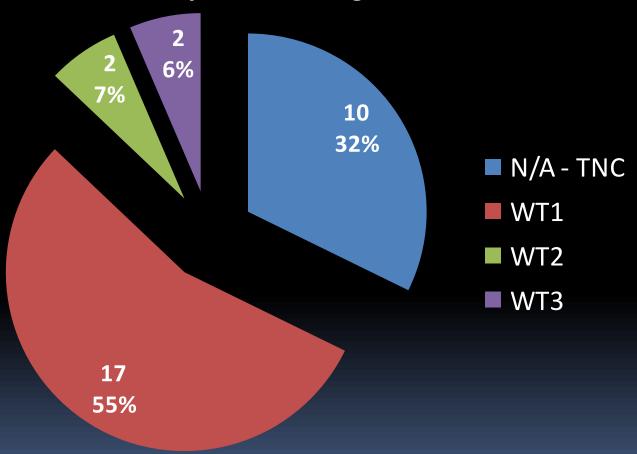
REGULATORY REQUIREMENTS

Plan Review – OAR 333-061-0050(4)(c)(C) & (E)

- (C) Pilot studies shall be conducted by the water supplier to demonstrate the effectiveness of any filtration method other than conventional filtration. Pilot study protocol shall be approved in advance by the Authority. Results of the pilot study shall be submitted to the Authority for review and approval.
- (E) All filtration systems shall be designed and operated so as to meet the requirements in OAR 333-061-0032(4) and (5) i.e., meet turbidity limits and CT requirements. Design of the filtration system must be in keeping with accepted standard engineering references acknowledged by the Authority such as the Ten States Standards, technical reports by the International Reference Center for Community Water Supply and Sanitation ("IRC manual"), or publications from the World Health Organization ("WHO manual").

REGULATORY REQUIREMENTS - OTHER

Operator Certification Level Required for Slow Sand Filtration Systems in Oregon



N/A - TNC = Transient Non-Community water systems that are required to have their operator(s) attend a 1-time only class (o.6 CEU class).

REGULATORY REQUIREMENTS

Surface Water Treatment Rule (SWTR), 1989

- 40 CFR 141.70 141.75 (applies to all SW and GWUDI systems a.k.a "Subpart H" systems)
- Required 3.o-log (99.9%) Giardia and 4-log (99.99%) virus removal/inactivation (filtration plus disinfection)
- Established turbidity limits (< 1 NTU in 95% of readings w/all < 5 NTU)
- Established disinfectant residual requirements

Interim Enhanced Surface Water Treatment Rule (IESWTR), 1998

- 40 CFR 141.170 141.175
- Added 2.o-log cryptosporidium treatment requirements

Long-Term 1 Enhanced Treatment Water Rule (LT1), 2002

- 40 CFR 141.500 141.571
- Extended IESWTR requirements for systems < 10,000 pop

Long-Term 2 ESWTR (LT2), 2006

- 40 CFR 141.700 141.723 & 40 CFR 141.211, Appendix A to Subpart Q
- Additional Cryptosporidium treatment requirements depending upon source sampling and resultant bin classification (more treatment if higher than bin 2)
- Addressed uncovered finished water reservoirs

REGULATORY REQUIREMENTS -PATHOGEN RMVL

Applicability: PWSs that use SW or GWUDI that practice SSF, DE, or Alternative Filtration

Regulated Pathogen

99.99% (4-log) removal/inactivation of viruses (SWTR)

99.9% (3-log) removal/inactivation of *Giardia lamblia (SWTR)*

99% (2-log) removal of Cryptosporidium (IESWTR/LT1)

(> 2-log if Bin 2 or higher under LT2)

Slow sand filtration is credited with removing:

- 2.0-log Giardia &
- 2-log Cryptosporidium

1.0-log Giardia inactivation is needed through disinfection, 0.5-log of which must be obtained after filtration.

REGULATORY REQUIREMENTS - TURBIDITY

Turbidity Limits									
Turbidity	Turbidity readings are to be monitored/recorded at the combined filter effluent (CFE) at a frequency of at least once every 4 hours*	95% of CFE turbidity readings ≤ 1 NTU (≤ 1.49 NTU) All CFE turbidity readings ≤ 5 NTU (≤ 5.49 NTU)							

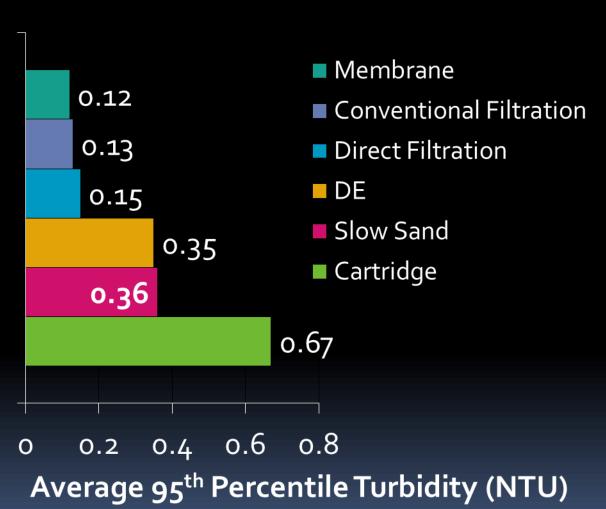
^{*} Frequency may be reduced by the State to once per day.

REGULATORY REQUIREMENTS - NTU REPORTING

Turbidity Reporting							
Turbidity	Total # of Monthly Measurements						
reporting required within 10 days after	Number and percent less than or equal to 95 th percentile turbidity limit						
the end of the month:	Date and Value Exceeding 5 NTU						
Turbidity reporting required within 24 hours:	Exceedances of 5 NTU for CFE						

SLOW SAND - ABLE TO MEET 1 NTU LIMIT

Average 95th percentile turbidity based on the maximum turbidity measured daily for each of the surface water treatment plants in Oregon from July 1, 2011 - June 30, 2012



TURBIDIMETERS

- Turbidimeters
 - Online, portable or bench-top
 - Must be calibrated per manufacturer or at least quarterly with a <u>primary</u> standard
 - Formazin solution
 - StablCal® (stabilized formazin)
 - Secondary standards used for day-to-day check
 - Check is used to determine if calibration with a primary standard is necessary
 - Gelex
 - Manufacturer provided (e.g. Hach ICE-PIC)



Disinfection Profiling And Benchmarking Source Water Monitoring (Round 1) 2006 - 2010



Bin Classification



Implement
Treatment from
Microbial
Toolbox
(If Required)



Source Water Monitoring (Round 2) 2015 - 2019 Uncovered Finished Water Reservoirs

Treatment Requirement
LT2 requires additional
Crypto treatment for
systems with ≥ 0.075
oocysts/L in their source
water. So far only one
water system is required
to meet additional
treatment requirements
under LT2 in Oregon.

40 CFR 141.701(c) Monitoring Schedule

Initial and second round monitoring must begin no later than the month beginning with the date listed in the table below.

Schedule	Systems that serve	1 st Round	2 nd Round	
1	At least 100,000 people* October 1, 2006		April 1, 2015	
2	From 50,000 to 99,999 people*	April 1, 2007	October 1, 2015	
3	From 10,000 to 49,999 people* April 1, 2008		October 1, 2016	
4 (E. coli)	Fewer than 10,000, not a wholesale system, and monitors for <i>E. coli</i> a	October 1, 2008	October 1, 2017	
4 (Crypto)	Fewer than 10,000, not a wholesale system, and monitors for Cryptosporidium b	April 1, 2010	April 1, 2019	

^{*}Also applies to wholesalers in a combined distribution system (CDS) that contains a schedule 1, 2, or 3 system

^a Applies only to filtered systems.

^b Applies to filtered systems that meet the conditions of paragraph (a)(4) of §141.701 and unfiltered systems.

Filtered System Additional Cryptosporidium Treatment Requirements

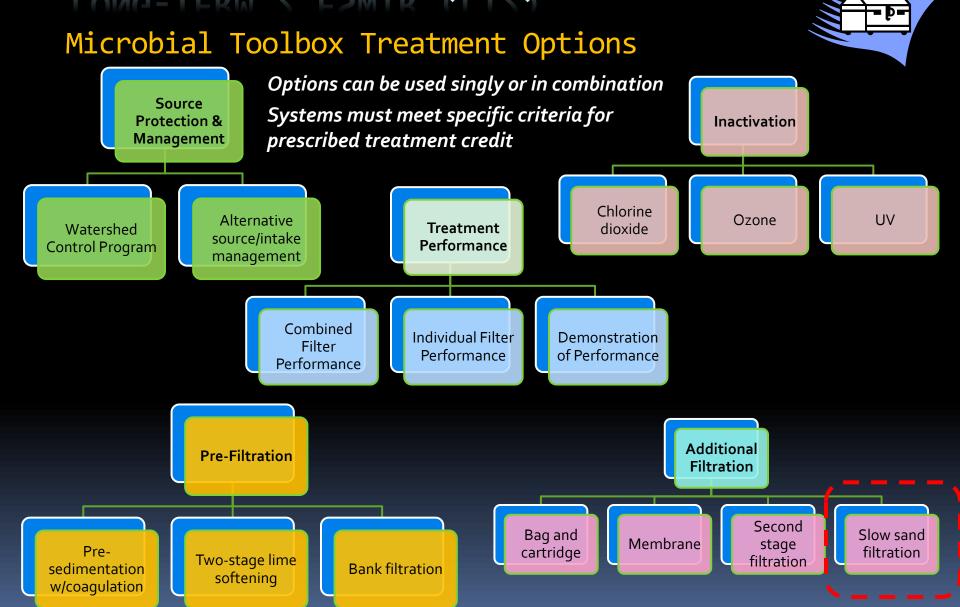
(based on their bin classification as determined under § 141.710 and according to the schedule in § 141.713)

bin	Conventional Filtration (including softening), Slow Sand, or Diatomaceous Earth	Direct filtration	Alternative filtration technologies			
Bin 1	No Additional Treatment					
Bin 2	1-log treatment	1.5-log treatment	RMVL + Inactivation			
Bin 3	2-log treatment	2.5-log treatment	RMVL + Inactivation > 5.0-log ²			
Bin 4	2.5-log treatment	3-log treatment	RMVL + Inactivation <u>></u> 5.5-log ³			

¹As determined by the State such that the total *Cryptosporidium* removal and inactivation is at least 4.0-log.

²As determined by the State such that the total *Cryptosporidium* removal and inactivation is at least 5.0-log.

³As determined by the State such that the total *Cryptosporidium* removal and inactivation is at least 5.5-log.



REGULATORY REQUIREMENTS - DISINFECTION

Entry Point Chlorine Residual

Entry Point Residual
Disinfection
Concentration

(for free chlorine measured prior to or at the first customer each day of operation) Residual disinfectant concentration cannot be < 0.2 mg/l for more than 4 hours based on continuous monitoring (> 3,300 pop) or less frequent monitoring as allowed by the state. (SWTR)

(contact your state regulator if using a disinfectant other than chlorine or are planning to switch disinfectants)

No two consecutive daily samples should exceed 4.0 mg/l (DBPR)

Where chlorine is used as the disinfectant, the measurement of residual chlorine shall be by the <u>DPD or other EPA-approved method</u> in accordance with Standard Methods for the Examination of Water and Waste-water, and shall measure the free chlorine residual or total chlorine residual as applicable

REGULATORY REQUIREMENTS - DISINFECTION

Distribution System Chlorine Residual

Distribution System Residual Disinfection Concentration

(for free chlorine measured with coliform samples)

(contact your state regulator if using a disinfectant other than chlorine or are planning to switch disinfectants)

Residual disinfectant concentration cannot be undetectable in greater than 5% of samples in a month, for any 2 consecutive months.

(SWTR)

Not to exceed 4.0 mg/I MRDL* (DBPR)

*The maximum residual disinfectant level (MRDL) is regulated under the Disinfection By-Products Rules (DBPR). Compliance is based upon chlorine residuals taken at the same location and frequency as that required for total coliform monitoring in the distribution system. The running annual average of monthly averages of samples, computed quarterly, must be $\leq 4.0 \text{ mg/l}$.

REGULATORY REQUIREMENTS - CL2 REPORTING

Additional Distribution Residuals Monitoring 2x per week

Distribution

(records to be kept by the water system for at least 2 years) All public water systems that add a disinfectant to the water supply at any point in the treatment process, or deliver water in which a disinfectant has been added to the water supply, must maintain a detectable disinfectant residual throughout the distribution system and shall measure and record the residual at one or more representative points at a frequency that is sufficient to detect variations in chlorine demand and changes in water flow but in no case less often than twice per week.

REGULATORY REQUIREMENTS - CL2 REPORTING

Chlorine Residual Reporting Required (within 10 days after the end of the month)

Entry Point

(reported with turbidity)

Lowest daily value for each day, the date and duration when residual disinfectant was < 0.2 mg/l, and when State was notified of events where residual disinfectant was < 0.2 mg/l.

Distribution

(reported with coliform sample results)

Number of residual disinfectant or HPC measurements taken in the month resulting in no more than 5% of the measurements as being undetectable in any 2 consecutive months.

CHLORINE ANALYZERS

- Chlorine analyzers
 - Handheld (HACH Colorimeter shown)
 - Follow manufacturer's instructions



- Online
 - Check calibration against a handheld that has been calibrated
 - At least weekly
 - Follow manufacturer's instructions if out of calibration

REGULATORY REQUIREMENTS - OTHER

Other SWTR/IESWTR/LT1 Requirements

Disinfection Profiling & Benchmarking	Systems must profile inactivation levels and generate a benchmark, if required due to disinfection changes (IESWTR & LT1)
Water System Surveys (State Requirement)	CWS: Every 3 years NCWS: Every 5 years (IESWTR & LT1)
Finished Water Reservoirs	New (post-1989) reservoirs must be covered under SWTR. Pre-SWTR reservoirs must be covered (or have additional treatment) under LT2
Operator Certification	Operated by Qualified Personnel as Specified by State (SWTR)

Cyanotoxin Monitoring (OAR 333-061-0510 to -0580) Healthoregon.org/dwcyanotoxins

Who does this apply to?	Affects systems who have sources susceptible to cyanobacteria blooms (not everyone). See systems and specific rule requirements on-line at www.healthoregon.org/dwcyanotoxins						
What is required?	Raw water (intake) sampling for total microcystin and Cylindrospermopsin toxins every 2 weeks from May 1 st – October 31 st each year						
What happens if detected?	 Notify your regulator If any toxins are greater than or equal to 0.3 μg/L in raw water or if there is a recreational use health advisory* upstream of the intake, sample raw and entry point weekly with the first EP sample taken within 1 business day. Weekly sampling continues until non-detect at EP and less than 0.3 μg/L in raw water in two consecutive samples. If detected at EP, sample EP daily and optimize treatment for toxin removal. If above Health Advisory Level (HAL) at EP, take confirmation sample within 24-hrs & monitor EP daily. If confirmation sample is above the HAL, issue Do-Not-Drink Advisory Advisory may only be lifted if 2 consecutive daily EP samples taken a minimum of 24-hrs apart are ≤ HAL and two consecutive daily sets of distribution samples taken a minimum of 24 hours apart are ≤ HAL "Recreational use health advisory" means a health advisory issued by the Oregon Health Authority for a water body when cyanotoxins are determined to be above any recreational use advisory levels. 						
What are the DW Health	 Total Microcystins: 0.3 μg/L for vulnerable people; 1.6 μg/L for all persons Cylindrospermopsin: 0.7 μg/L for vulnerable people; 3 μg/L for all persons 						

"Vulnerable people" means infants, children under the age of six, pregnant women, nursing mothers, those with pre-existing liver conditions, and those receiving dialysis treatment.

Advisory

Levels (HALs)?

REVIEW

- 2.0-log Cryptosporidium removal is required (and credited) for slow sand filtration.
- Surface Water Treatment Rule (SWTR) requires 3log reduction of *Giardia* using a combination of disinfection and filtration and 4.0-log reduction of viruses.
- At least 2.0 -log Giardia removal is credited for slow sand filtration (per 1991 USEPA SWTR Manual)
- 1.0-log Giardia inactivation must be achieved through disinfection (0.5-log must be after filtration). 1.0-log reduction of viruses must also be achieved after filtration.

REPORTING FORMS

There are 4 forms:

- Conventional/Direct
- Slow Sand / Membrane / DE / Unfiltered
- Cartridge
- UV (if used for Giardia credit)

Must use correct form because each has questions that must be answered that are specific to the filtration type

REPORTING FORMS - CFE TURBIDITY

OHA - Drinking Water Program – Turbidity Monitoring Report Form County: Slow Sand, Membrane, Diatomaceous Earth Filtration, or Unfiltered Systems

			.,	,			,	orea eyotomo			
System Name:					ID #:	WTP-:	Month	Month/Year:			
	DAY	12 AM [NTU]	4 AM [NTU]	8 AM [NTU]	NOON [NTU]	4 PM [NTU]	8 PM [NTU]	Highest Reading of the Day ¹ [NTU]			
	1			0.34				0.50			
	2			0.24				0.66			
	3			0.44	Notify the State if NTU > 1 NTU.						
	4			A	Notify th	e State w	ithin 24-h	rs if turbidity >	5 NTU		
	5				,	(incl	udes afte	r hours)			
	6				Pub	lic Health	After Ho	urs Duty Office	r:		
	7						l (971) 24	,			
	8				Pager (503) 938-6790						
	9				Oregon Emergency Response System:						
	10				016		-800-452	•			
				_							

- Chose time closest to when daily turbidity is measured and enter result(s)
- Enter highest turbidity of all measurements for the day (e.g., online instrument or highest of multiple daily grab samples)

REPORTING FORMS - MONTHLY SUMMARY - TURBIDITY

Slow Sand/Membrane/DE Filtration/Unfiltered

95% of daily turbidity readings ≤ 1 NTU?

All daily turbidity readings ≤ 5 NTU?



 Based on the results entered for the month, circle "yes" or "no" to the two questions at the bottom of the form.

OHA - Drinking Water Program - Surface Water Quality Data Form

	on the printer of the										
System Name:					ID #:	٧	VTP-:	Month/Year	:		
	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? ³	Peak Hourly Demand Flow		
		[ppm or mg/L]	[minutes]	схт	[° C]		Use tables	Yes / No	[GPM]		
	1/9 AN								1,000		
	2/										
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	4 /										
	5 /										
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	40.7										

- Enter the peak hourly demand (PHD) flow and the time that the PHD flow occurred.
- This flow should not exceed 10% above the peak flows replicated at the time of the last tracer study.

	OHA - Drinking Water Program – Surface Water Quality Data Form									
S	stem Name	:			ID#:	V	VTP-:	Month/Year		
	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? 3	Peak Hourly Demand Flow	
		[ppm or mg/L]	[minutes]	CXT	[° C]		Use tables	Yes / No	[GPM]	
	1/9 AN								1,000	
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Peak Hour Demand Flow:

- The greatest volume of water passing through the system during any one hour in a consecutive 24 hr period
- Not the same as Peak Instantaneous Flow
- Report demand flow: flow leaving the clear well, not plant flow (in most cases)

Method for determining peak hourly demand flow (flow meter w/rate):

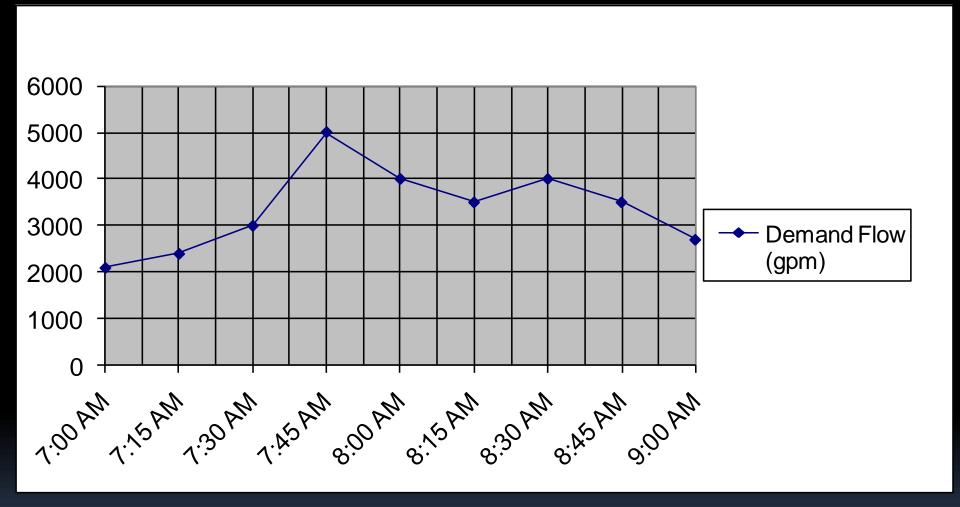
- On a <u>daily basis</u>, use the <u>best available operational data</u> to identify the hour within the 24 hr period that had the highest demand flow.
- For the hour of highest demand flow:
 - Calculate the average flow rate within the one hour period (i.e., add the flow rates and divide by the number of data points).
 - <u>Use as many data points as possible</u>, preferably no less than four data points taken at 15 minute intervals

For systems that only have a flow totalizing meter:

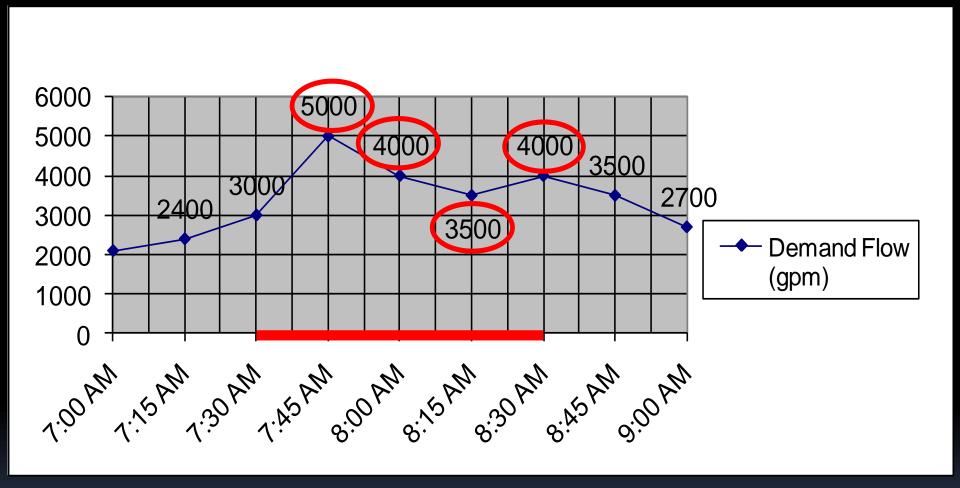
• Spot check throughout the day to determine the time of peak demand (e.g. 8 am or 9 pm for residential or mid-day for industrial uses)

OHA - Drinking Water Program – Surface Water Quality Data Form									
Sy	rstem Name	:			ID #:	V	VTP-:	Month/Year	:
	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? 3	Peak Hourly Demand Flow
		[ppm or mg/L]	[minutes]	CXT	[° C]		Use tables	Yes / No	[GPM]
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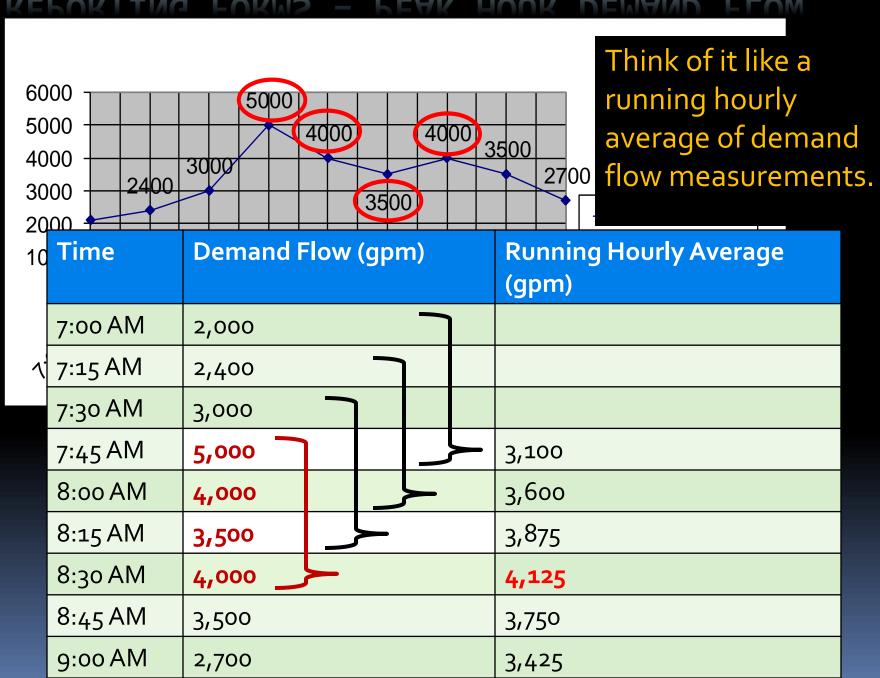
 Then record how much water is used during that hour in gallons and divide by 60 minutes to get the peak hour demand in gpm



Here is an example chart, meant to represent continuous readings that shows demand flow out of a reservoir used for contact time. What would you say the peak hourly demand flow is?



Again, the peak hourly demand flow is the hour within the 24-hr period of the highest demand flow. The red line represents the span of 1 hour: 7:30 am to 8:30 am – the peak hour. The avg. of the 4 data points equals 4,125 gpm - the peak hourly demand flow.



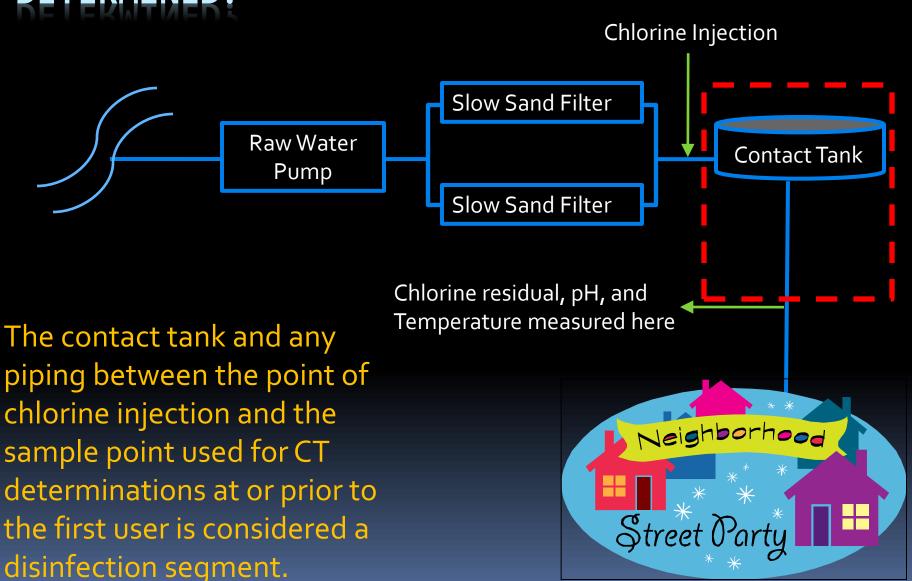
REPORTING FORMS - CHLORINE & CONTACT TIME

OHA - Drinking Water Program – Surface Water Quality Data Form

					-				
Sy	stem Name	:			ID #:	٧	VTP-:	Month/Year	:
Ì	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? ³	Peak Hourly Demand Flow
		[ppm or mg/L]	[minutes]	СХТ	[° C]		Use tables	Yes / No	[GPM]
	1/9 AN	???	???						1,000
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- The minimum chlorine residual is measured at the end of the disinfection segment.
- Contact time is the time that the disinfectant is in contact with the water within the disinfection segment.

HOW IS THE DISINFECTION SEGMENT DETERMINED?



HOW IS CONTACT TIME DETERMINED?

 Tracer studies are used to determine contact time (T) which is used in calculating CT achieved, where

CT = chlorine Concentration x contact Time.

- Contact time is the time that chlorine is in contact with the water from the point of injection to the point where it is measured (sometimes referred to as the "CT segment")
 - May be at or before the 1st user
 - May be more than one CT segment
- Tracer studies are often conducted to simulate a worst-case scenario where peak hour demand flows are high and reservoir levels are low. This gives a conservative (i.e. lower) contact time than would normally be expected.

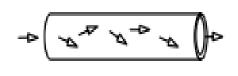
HOW IS CONTACT TIME DETERMINED?

- The more efficient the mixing is in a reservoir or tank, the more contact time is available for disinfection.
- Estimates of contact time based on tank or reservoir design are not allowed for calculating CT's for surface water!



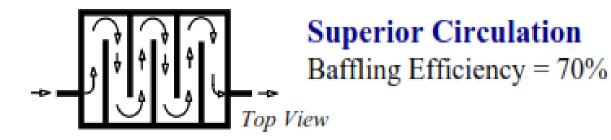
WHAT AFFECTS MIXING EFFICIENCY?

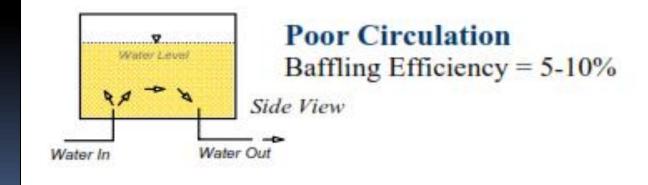
Mixing efficiency improves with high flow path length to width ratios, found in pipelines and simulated in tanks with the use of baffles (hence the term baffling efficiency or factor).



Perfect Circulation

Baffling Efficiency = 100% Plug flow through a length of pipe

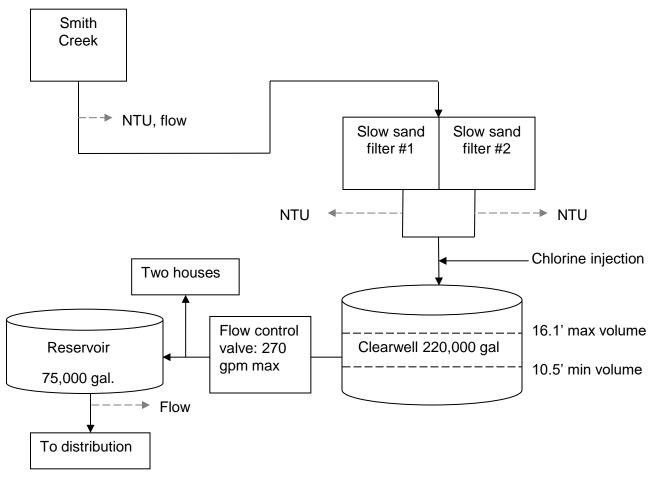




Example: Tracer studies

Directions: Look at the diagram and answer the questions.

Figure 1: Water Treatment Plant



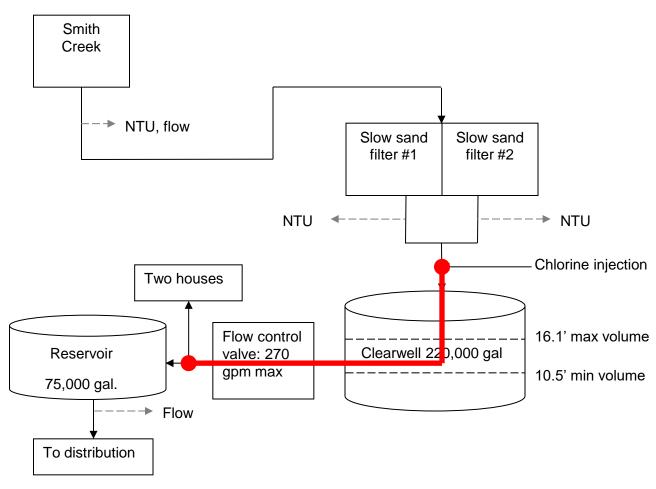
Questions:

- •If this was your treatment plant, highlight the part of the plant where you might conduct a tracer study.
- •In a "worst-case scenario" tracer study, what would the flow rate be?
- •In a "worst-case scenario" tracer study, what would the clearwell level be?

Example: Tracer studies - Answer

Directions: Look at the diagram and answer the questions.

Figure 1: Water Treatment Plant



Questions:

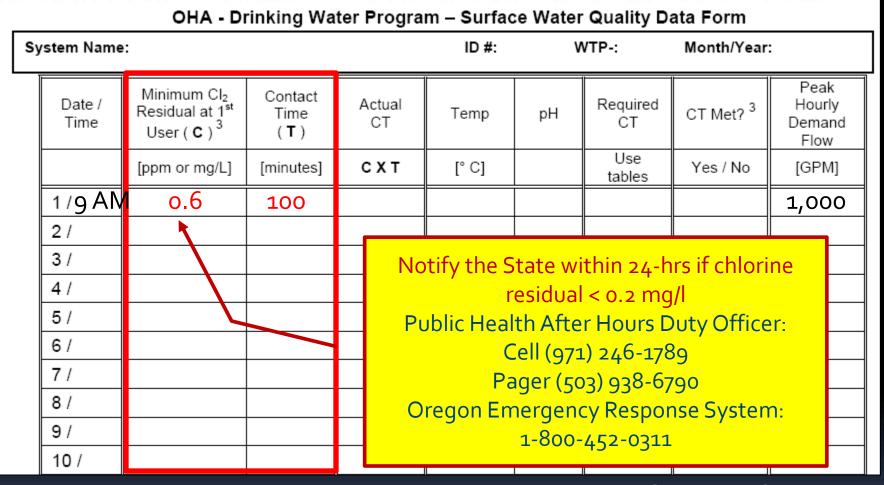
- •If this was your treatment plant, highlight the part of the plant where you might conduct a tracer study.
- •In a "worst-case scenario" tracer study, what would the flow rate be? 270 gpm
- •In a "worst-case scenario" tracer study, what would the clearwell level be?

10.5 feet

DO I REPORT CONTACT TIME?

- Use the time T from the tracer study on the monthly reporting form in the "Contact time (min)" column
 - Use the smallest T (highest flow) if the tracer study was done at multiple flow rates
- This may not be your exact time, but it represents your worst case (as long as the peak flow is less and clearwell volume is more than they were at the time of the tracer study)

REPORTING FORMS - CHLORINE & CONTACT TIME



- Enter the minimum chlorine residual at or before the first user.
- Enter the contact time (based either on the tracer study or determined from clearwell volume(s) and the peak hourly demand flow).

CAN I USE A BAFFLING FACTOR?

- As an alternative to using the tracer study contact time, you can use the results of the tracer study to determine the baffling factor of the clearwell
 - Baffling factor (%) = <u>Time (min) x Flow During Tracer Study (gpm)</u>
 Clearwell Volume During Tracer Study (gal)
- T can be adjusted based on flow (at flow < 110% of tracer study flow) with the following equation:
 - T = <u>Current clearwell Volume (gal) x Baffling Factor (%)</u>
 Peak Hourly Demand Flow (gpm)
- Contact the state for guidance on using baffling factors.

REPORTING FORMS - ACTUAL CT

OHA - Drinking Water Program - Surface Water Quality Data Form

System Name: ID #: WTP-: Month/Year:

Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? ³	Peak Hourly Demand Flow
	[ppm or mg/L]	[minutes]	СХТ	[° C]		Use tables	Yes / No	[GPM]
1/9 AN	1 0.6	100	60					1,000
2 /								
3 /								
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8 /								
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- Enter the actual CT achieved that day:
 - Actual CT = Chlorine Concentration (mg/l) x Contact Time (min)
- Do not confuse "CT" and "Contact Time"

REPORTING FORMS - TEMPERATURE & PH

OHA - Drinking Water Program - Surface Water Quality Data Form

System Name	: :			ID #:	V	VTP-:	Month/Year	:
Date / Time	Minimum Cl₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? ³	Peak Hourly Demand Flow
	[ppm or mg/L]	[minutes]	схт	[° C]		Use tables	Yes / No	[GPM]
1/9 AN	1 0.6	100	60	12	6.8			1,000
2 /								
3 /								
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6 /								
7 /								
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 Enter the finished water temperature (°C) and pH measured at or prior to the first customer and after any storage (tank, reservoir, or pipeline) used for contact time.

REPORTING FORMS - REQUIRED CT

OHA - Drinking Water Program - Surface Water Quality Data Form

				3			,		
Sy	stem Name	:			ID #:	W	/TP-:	Month/Year	:
	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? ³	Peak Hourly Demand Flow
		[ppm or mg/L]	[minutes]	схт	[° C]		Use tables	Yes / No	[GPM]
	1/9 AN	1 0.6	100	60	12	6.8			1,000
	2 /								
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Actual CT must be ≥ Required CT. To determine required CT:

1. Use USEPA CT tables or

9 /

10 /

Regression Equations (Use 1 of 2 equations –depends on °C)

HOW IS REQUIRED CT CALCULATED?

- We use the EPA tables (or "regression equations") to determine the CT required to inactivate Giαrdia (CT_{required})
 - 1-log inactivation of Giardia using chlorine results in at least 4.0-log inactivation of viruses.
 - To determine CT, we need to know pH, temperature, and free chlorine residual at or before the first user.
- Then we compare the CT_{required} with the actual CT achieved in the water system (CT_{actual}) where:
 - $CT_{actual} = chlorine concentration (mg/l) x contact time (min)$
- Must keep CT_{actual} ≥ CT_{required}

USING REGRESSION EQUATIONS TO DETERMINE REQUIRED CT

Using Regression Equations to determine required CT:

- Built into the MS Excel reporting forms on-line <u>http://public.health.oregon.gov/HealthyEnvironments/DrinkingWater/Operations/Treatment/Pages/index.aspx</u>
- Surface Water Monitoring and Reporting Forms for CT and Turbidity Data:
 - If your system has more than one chlorine injection point, or if you have questions about the PDF or MS Excel versions of the monthly turbidity and surface water monitoring forms, contact the DWS technical oversight contact for your system at 971-673-0405.
 - Conventional or Direct Filtration: PDF -or- MS Excel
 - Slow Sand, Membrane, Diatomaceous Earth Filtration or Unfiltered: PDF -or- MS Excel
 - Cartridge or Bag Filtration: PDF -or- MS Excel

USING REGRESSION EQUATIONS, CONT.

Using Regression Equations to determine required CT:

2. Regression equations can be programmed into plant SCADA or spreadsheets

Regression Equation (for Temp < 12.5°C)

 $CT = (0.353 \text{ L})(12.006 + e^{(2.46-0.073 \text{ T} + 0.125 \text{ C} + 0.389 \text{ pH})})$

Regression Equation (for Temp > 12.5°C)

 $CT = (0.361 \text{ L})(-2.261 + e^{(2.69-0.065 \text{ T}+0.111 \text{ C}+0.361 \text{ pH})})$

Variables:

CT = Product of Free Chlorine Residual and Time required

L = number of log inactivation for Giardia (L = 1 for slow sand)

T = temperature, in Celsius

C = chlorine residual in mg/L

pH = pH of water

e = 2.7183, base for natural log

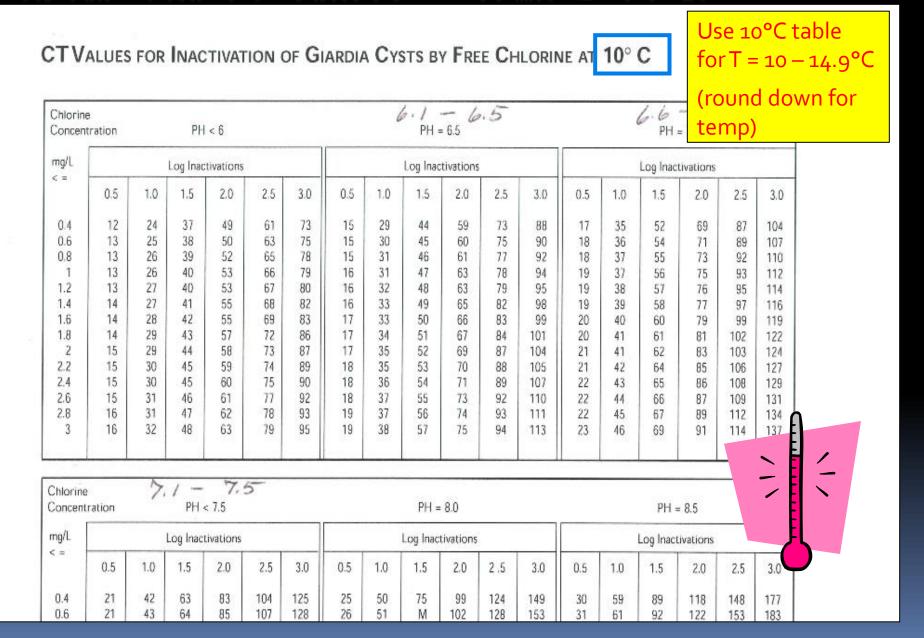
(Smith, Clark, Pierce and Regli, 1995, from EPA's 1999 Guidance

Manual for Disinfection Profiling and Benchmarking)

USING EPA CT TABLES - TEMPERATURE

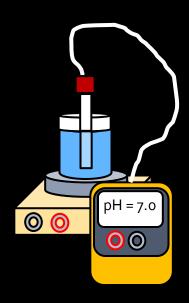
- There are six EPA CT tables based on temperature
- Find the correct table based on your water temperature in degrees <u>Celsius</u>.
 - $^{\circ}$ C = 5/9 x ($^{\circ}$ F 32)
- If water temp is between values, then round <u>down</u>
 - Example: for water temp of 12°C, use the 10°C table
 - Even if the water temp is 14.9°C, round down to 10°C
- Water gets more viscous the colder it gets and chemical reactions take longer, so rounding temp down is more conservative.

USING EPA CT TABLES - TEMP = 12 °C



USING EPA CT TABLES - PH

 There are 7 sections for pH on each table



- Find the section that corresponds to your water's pH level
- If your pH is between the choices, then round up to the higher pH
 - Example: if pH of water is 6.8, use the pH 7.0 section

USING EPA CT TABLES - PH = 6.8

CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10°C 10°C - 14.9°C

Chlorin Concen			PH	l < 6				ı	PH:	- 6 = 6.5	.5				6.6 PH	-) = 7.0	>.0	
mg/L < =			Log Inac	tivations					Log Inac	ctivations					Log Inac	ctivations		
	0.5	1.0	1.5	2.0	2,5	3.0	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	U	se n	H =	7.o colum
0.4 0.6 0.8	12 13 13	24 25 26	37 38 39	49 50 52	61 63 65	73 75 78	15 15 15	29 30 31	44 45 46	59 60 61	73 75 77	88 90 92	17 18 18	35 36 37	fo	or pH	H = 6	5.6 – 7.0
1 1.2	13 13	26 27	40 40	53 53	66 67	79 80	16 16	31 32	47 48	63 63	78 79	94 95	19 19	37 38	(r	oun	d up	for pH)
1.4 1.6 1.8 2 2.2 2.4 2.6	14 14 14 15 15 15	27 28 29 29 30 30 31	41 42 43 44 45 45 46	55 55 57 58 59 60 61	68 69 72 73 74 75 77	82 83 86 87 89 90	16 17 17 17 18 18	33 34 35 35 36 37	49 50 51 52 53 54 55	65 66 67 69 70 71 73	82 83 84 87 88 89 92	98 99 101 104 105 107 110	19 20 20 21 21 21 22 22	39 40 41 41 42 43 44	58 60 61 62 64 65 66	77 79 81 83 85 86	97 99 102 103 106 108	116 19 122 24
2.8 3	16 16	31 32	47 48	62 63	78 79	93 95	19 19	37 38	56 57	74 75	93 94	111 113	22 23	45 46	67 69	89 91	112 114	
nlorine		>	/ -	7.	5												K	pH = 7.0
oncent				< 7.5					PH =	8.0					PH	= 8.5	0 (
g/L ≡			Log Inac	tivations					Log Inac	tivations					Log Inac	tivations		00
~ .	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0
0.4	21 21	42 43	63 64	83 85	104 107	125 128	25 26	50 51	75 M	99 102	124 128	149 153	30 31	59 61	89 92	118 122	148 153	177 183

USING EPA CT TABLES - 1-LOG



Use the 1-log inactivation column

(slow sand is granted 2.0-log removal credit for *Giardia*, which requires that 1.0-log *Giardia* inactivation is needed through disinfection)

USING EPA CT TABLES - REQUIRED LOG = 1.0

CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10°C 10°C - 14.9°C

Chlorine Concent			PH	< 6				6	/- / PH :	- 6.5	.5					-) = 7.0	>.0		
mg/L <=			Log Inac	tivations					Log Inac	tivations					Log Inac	tivations			
	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1,5	2.0	2.5	3.0	0.5	1.0	F	or slo	OW S	and	, Use
0.4 0.6 0.8	12 13 13	24 25 26	37 38 39	49 50 52	61 63 65	73 75 78	15 15 15	29 30 31	44 45 46	59 60 61	73 75 77	90 92	17 18 18	35 36 37					lumn
1 1.2 1.4 1.6	13 13 14 14 14	26 27 27 28 29	40 40 41 42 43	53 53 55 55	66 67 68 69 72	79 80 82 83 86	16 16 16 17 17	31 32 33 33	47 48 49 50	63 63 65 66	78 79 82 83	94 95 98 99	19 19 19 20	37 38 39 40	56 57 58 60	75 76 77 79	93 95 97 99	112 114 116 119	
1.8 2 2.2 2.4 2.6	15 15 15 15	29 30 30 31	45 45 45 46	57 58 59 60 61	73 74 75 77	87 89 90 92	17 18 18 18	34 35 35 36 37	51 52 53 54 55	67 69 70 71 73	84 87 88 89 92	101 104 105 107 110	20 21 21 22 22	41 41 42 43 44	61 62 64 65 66	81 83 85 86 87	102 103 106 108 109	122 124 127 129 131	
2.8	16 16	31 32	47 48	62 63	78 79	93 95	19 19	37 38	56 57	74 75	93 94	111	22 23	45 46	67 69	89 91	112 114	134 137	
hlorine		7.		7, : < 7.5	5				PH =	8.0					PH:	()			
ng/L			Log Inac						Log Inac						Log Inac		_		
=	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	6,5		
0.4 0.6	21 21	42 43	63 64	83 85	104 107	125 128	25 26	50 51	75 M	99 102	124 128	149 153	30 31	59 61	89 92	118 122	148 153	177 183	

USING EPA CT TABLES - CHLORINE

 Match your free chlorine residual on the far left column



- If in between column values, round <u>up</u>
 - Rounding chlorine residual up is more conservative because as chlorine residual increases at a given pH, more CT is required
- The point where it intersects with the log inactivation column is the CT_{required}
 - Example: free chlorine residual is 0.6 ppm

USING EPA CT TABLES - CL₂ = 0.6 MG/L

CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10°C 10°C - 14.9°C

Chlorine Concent			PH	< 6				6	/- / PH :	- 6.5	.5				6.6 PH	- > = 7.0	.0	
mg/L <=			Log Inac	tivations					Log Inac	tivations					Log Inac	tivations		
· -	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1,5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0
0.4	12	24	37	49	61	73	15	29	44	59	73	88	17	35	52	69	87	104
0.6	13	25	38	50	63	75	15	30	45	60	75	90	18	36	54	71	89	107
0.8	13	26	39	52	65	78	15	31	46	61	77	92	18	37	55	73	92	110
1	13	26	40	53	66	79	16	31	47	63	78	94	19	37	56	75	93	112
1.2	13	27	40	53	67	80	16	32	48	63	79	95	19	38	57	76	95	114
1.4	14	27	41	55	68	82	16	33	49	65	82	98	19	39	58	77	97	116
1.6	14	28	42	55	69	83	17	33	50 51	66 67	83 84	99	20 20	40 41	60	79	99	119
1 0																		
	ام میں		د			CI	:-	15		10.75.40111	177.20	2.500	70555	(50.50)	61 62	81	102	122
Ro	ound	up i	if m	easu		CL ₂	is	15	52	69	87	104	21	41	62	83	103	124
					ıred	_		5		10.75.40111	177.20	104 105	21 21	(50.50)	62 64	83 85	103 106	124 127
be	twe	en v	alue	es in	red the	chlc		5	52 53	69 70	87 88	104	21 21 22	41 42	62	83	103	124
be		en v	alue	es in	red the	chlc		5	52 53 54	69 70 71	87 88 89	104 105 107	21 21	41 42 43	62 64 65	83 85 86	103 106 108	124 127 129
be	twe	en v	alue	es in	red the	chlc		5	52 53 54 55	69 70 71 73	87 88 89 92	104 105 107 110	21 21 22 22	41 42 43 44	62 64 65 66	83 85 86 87 89	103 106 108 109	124 127 129
be	twe	en v	alue	es in	red the	chlc		5	52 53 54 55 56	69 70 71 73 74	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45	62 64 65 66 67	83 85 86 87 89	103 106 108	124 127 129
be	twe	en v	alue	es in	red the	chlc		5	52 53 54 55 56	69 70 71 73 74	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45	62 64 65 66 67	83 85 86 87 89	103 106 108 109	124 127 129
be	ncer	en v	alue	es in	the umn	chlc		5	52 53 54 55 56	69 70 71 73 74	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45	62 64 65 66 67	83 85 86 87 89	103 106 108 109	124 127 129
be co	ncer	en v	alue	es in colu	the umn	chlc		5	52 53 54 55 56	69 70 71 73 74 75	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45	62 64 65 66 67	83 85 86 87 8	103 106 108 109	124 127 129
Co Chlorine Concenti	ncer	en v	tion	es in colu	the umn	chlc		5 5 6 7 7 8	52 53 54 55 56 57	69 70 71 73 74 75	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45 46	62 64 65 66 67 69	93 95 96 87 89	103 106 108 109	124 127 129
co Chlorine	ncer	en v	tion	es in colu	the umn	chlc		5 5 6 7 7 8	52 53 54 55 56 57	69 70 71 73 74 75	87 88 89 92 93	104 105 107 110 111	21 21 22 22 22 22	41 42 43 44 45 46	62 64 65 66 67 69	93 95 96 87 89	103 106 108 109 177	124 127 129 131
Co Chlorine Concenti	ncer	en v	tion / - PH -	colu colu 7.5	the Jmn	chlo	orine	5 5 6 7 7 7 8	52 53 54 55 56 57 PH =	69 70 71 73 74 75 8.0	87 88 89 92 93 94	104 105 107 110 111 113	21 21 22 22 22 22 23	41 42 43 44 45 46	62 64 65 66 67 69 PH :	93 95 96 87 89	103 106 108 109 177	124 127 129 131

USING EPA CT TABLES - CL, = 0.6 MG/L

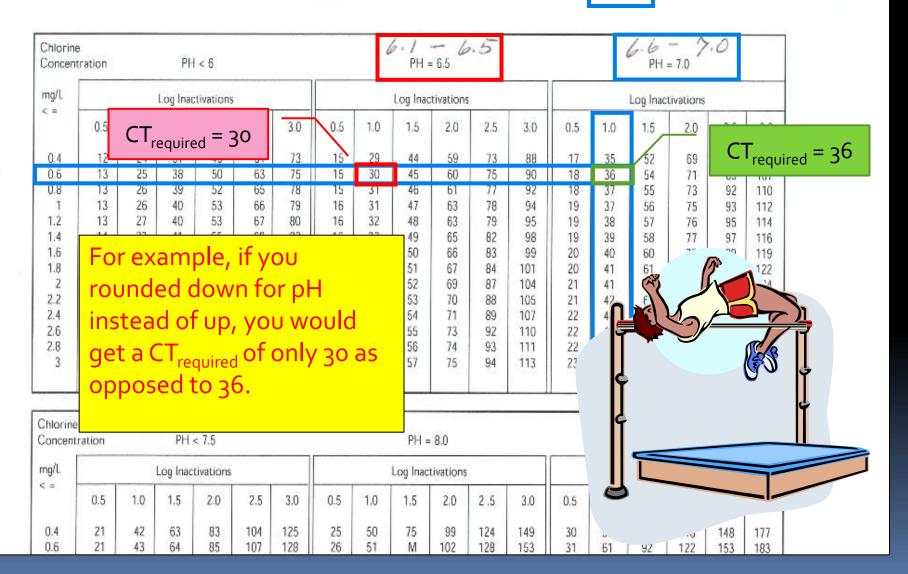
CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10°C 10°C - 14.9°C

Log nactivations	hlorine oncentr	ation		PH	l < 6				L	6 · / PH :	- 6 = 6.5	.5			(6.6 PH	-) = 7.0	>.0			
0.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 0.4 12 24 37 49 61 73 15 29 44 59 73 88 17 35 52 69 0.6 13 25 38 50 63 75 15 30 45 60 75 90 18 36 54 71 0.8 13 26 39 52 65 78 15 31 46 61 77 92 18 37 55 73 92 110 1 13 26 40 53 66 79 16 31 47 63 78 94 19 37 56 75 93 112 1.2 13 27 40 53 67 80 16 32 48 63 79 95 19 38 57 76 95 114 1.4 1.6 1.8 1.6 1.8 1.6 1.8 1.8 2 2 2.2 2.2 2.2 2.2 2.4 2.6 2.5 2.5 2.5 2.5 2.5 2.8 3 2.5 2.5 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5 Chlorine Concent protective of public health. 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.				Log Inac	tivations					Log Inac	tivations					Log Inac	tivations				
0.8	-	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1,5	2.0	2.5	3.0	0.5	1.0	1.5	2.0				
0.8														17		52	100000000000000000000000000000000000000	C	T_{requi}	red =	36
1 13 26 40 53 66 79 16 31 47 63 78 94 19 37 56 75 93 112 13 27 40 53 67 80 16 32 48 63 79 95 19 38 57 76 95 114 16 1.6 1.8 2 2.2 2.4 2.6 2.8 3 2.8 3 99 107 119 119 119 119 119 119 119 119 119 11			1000	555551	10000														197		
1.2	0.8	0.77	57/765	200000			1 2 2 3 6 3 5 5		10,000	18(2)7	1000	20000	0.0000					107900	2,000,000		
1.4 1.6 1.8 2 2 2 2.2 2.4 2.6 2.8 3 Chlorine Conservative and most protective of public health. mg/L 0.5 1.0 1.5 2.0 2.5 3.0 0.5	12		990,000	200000	150.00	20.20	0.00	1000000		100000	71,70	12.77	20,50,000	17,550	1000000	0.00 (20)			1,111,000		
If you get confused on which way to round, think about how you want to set the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. 1.6	A160	14	27	44	55	07	00	40	22	52	0.50							155,071	100000		
which way to round, think about how you want to set the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Solution S		If v	/OII	net	cor	ifus	ed d	n			66	1 - 1 - 1 - 1	99	20	85/40	1000	2,2	-50.0	7.17.01550		
about how you want to set the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. By 107 22 110 22 22 14 22 22 22 22 22 22 22 22 22 22 22 22 22				_								177.20	1.77	022000	05/50	61	<i>'</i>		122		
about how you want to set the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. The bar (CT _{required}) as high as possible to be the most conservative and most protective of public health.		wh	nich	wa	y to	rou	ınd,	thir	าk		1.1322.2	575.74	50525	2222.27	41	73	لر				
the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most conservative and most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health. Sector of the bar (CT _{required}) as high as possible to be the most protective of public health as high as possible to be the most protective of the bar (CT _{required}) as high as possible to be the most protective of the bar (CT _{required}) as high as protective of the bar (CT _{required}) as					,							1000000	7.557.757		⁴		1	4			
possible to be the most conservative and most protective of public health. Solution of public health. The ball (C required) as might as possible to be the most conservative and most protective of public health. PH = 8.0 Inactivations O.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5	2.6	ab	OUT	nov	w yc)U W	/ant	tos	set	55		(75,000)	0.000	22	7 \$		3	/	1/		
possible to be the most conservative and most protective of public health. PH = 8.0 Inactivations O.5	2.8	the	ha و	r (C	Т) as	hial	า ลร	56		100.00	1000								
Conservative and most protective of public health. Solution and most protective of public health. Output Description and most public health. Output Description and mo	3								ı us	57	75	94	113	2.7						J	
PH = 8.0 protective of public health. PH = 8.0	_	po	ssib	le t	o be	e th	e m	ost		_											
PH = 8.0 protective of public health. PH = 8.0	nlorine	COI	nsei	rvat	ive	and	l mo	st		_			-		¢						
mg/L = 0.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5		pro	nt a c	tive	a of	nuk	اماندا	ادمر	th	PH =	8.0										
< = 0.5 1.0 1.5 2.0 2.5 3.0 0.5 1.0 1.5 2.0 2.5 3.0 0.5	_{n/L} [Pro	JLEC	LIVE	2 01	hor	וטווכ	leal	cri.	n Iman	tiontine-										
	•	- 1	- 1							Ley mac	LIVALIONS										
0.4 21 42 63 83 104 125 25 50 75 99 124 149 30 30 31 148 177		0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0	0.5	0						
0.6 21 43 64 85 107 128 26 51 M 102 128 153 31 61 92 122 153 183	0.120	0.000.000	12.22.10	120000	- 2.57E-5	14.75.000	125	25		75	99	124	149	30	ù.			148	177		

USING EPA CT TABLES - CL, = 0.6 MG/L

CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10° C

10°C-14.9°



USING EPA CT TABLES - CL, = 0.6 MG/L

CT Values for Inactivation of Giardia Cysts by Free Chlorine at 10°C 10°C - 14.9°C

Chiorine Concent	Tarangan I		PH	< 6				ı	PH	- 6.5	,5			Ó	6-6 PH:	- 2 = 7.0	>.0			
mg/L < =			Log Inac	tivations				СТ		_ ~)				Log Inac	tivations				
N =	0.5	1.0	1.5	2.0	2.5	3.0	0.	C I _r	equire	d = 3	55	3.0	0.5	1.0	1.5	2.0	2.7			
0.4	12	24	37	49	61	73	15	29	44	59	73	88	17	35	52	69	C.	T _{requir}	$_{\text{red}} = 36$	
0.6 0.8 1 1.2 1.4 1.6 1.8 2 2.2 2.4 2.6 2.8 3	chl	/OU loring	ne ro get	esic : a C	lual, T _{rec}	, yo Juired	U		45 46 47 48 49 50 51 52 53 54 55 56 57	60 61 63 63 65 66 67 69 70 71 73 74 75	75 77 78 79 82 83 84 87 88 89 92 93 94	90 92 94 95 98 99 101 104 105 107 110 111	18 19 19 19 20 20 21 21 22 22 22 22	36 37 37 38 39 40 41 41 42	54 55 56 57 58 60 61	71 73 75 76 77 77 77	92 93 95 97	110 112 114 116 119 122		
hlorine oncent		7.	-	7. < 7.5 tivations	5				PH =	: 8.0 tivations				c						
=	0.5	1.0	1.5	2.0	2.5	3.0	0.5	1.0	1.5	2.0	2.5	3.0	0.5	3						
0.4 0.6	21 21	42 43	63 64	83 85	104 107	125 128	25 26	50 51	75 M	99 102	124 128	149 153	30 31	61	92	122	148 153	177 183		

REPORTING FORMS - REQUIRED CT

OHA - Drinking Water Program - Surface Water Quality Data Form

Sy	stem Name	:			ID #:	V	VTP-:	Month/Year	:
	Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	рН	Required CT	CT Met? ³	Peak Hourly Demand Flow
		[ppm or mg/L]	[minutes]	СХТ	[° C]		Use tables	Yes / No	[GPM]
	1/9 AN	1 0.6	100	60	12	6.8	36	Yes	1,000
	2 /							1	
	Publi	Pager (5 gon Emerger	er Hours 1) 246-17 03) 938-6	Duty Offi 89 790 Inse Syste	cer:				

- 1. Enter Required CT (CT tables or Regression Equations)
- 2. Was CT Met? Yes if Actual CT > Required CT
- 3. Actual CT must be ≥ Required CT

REPORTING FORMS - MONTHLY SUMMARY - CT & CL₂

Everyone needs to fill out the CT section of the Monthly Summary

Monthly	Summary (Answer	Yes or No)									
CT's met everyday? (see back) Yes No All Cl₂ residual at entry point ≥ 0.2 mg/l? Yes No											
PRINTED NAME:											
SIGNATURE:		DATE:									
PHONE #: ()		CERT #:									

MONTHLY REPORT

Slow Sand/Membra DE Filtration/Unfilte			thly Summa wer Yes or	
95% of daily turbidity readings ≤ 1 NTU? ²	Yes / No	CT's met everyday?	All Cl2 re entry ≥ 0.2	
All daily turbidity readings ≤ 5 NTU?	Yes / No	Yes / No	Yes	/ No
Notes:		PRINTED NA	ME:	
		SIGNATURE	:	DATE:
		PHONE #: ()	CERT #:

							Highest Reading of
DAY	12 AM	4 AM	8 AM	Noon	4 PM	8 PM	the Day
	[NTU]						
1			0.34				0.50

Date &	Minimum Cl ₂ Residual	Contact Time	Actual CT	Temp	рН	Required CT	CT Met?	Peak Hourly Demand
Time	at 1 st User	(T)				CI	IVICt:	Flow
Tille	(C)					(Use CT	(Yes	TIOW
	[mg/L]	[min]	CxT	[° C]		tables)	/No)	(GPM)
1 /9 AM	0.6	100	60	12	6.8	36	Yes	1,000

FILLING OUT THE MONTHLY REPORT - COMMON MISTAKES

- Not calculating CT's daily
 - Don't wait until the end of the month to do the calculations because if you discover you didn't meet CT's, it's too late!
- If adjusting contact time according to flow rate, use the demand flow, not the plant flow.
- Failure to answer questions at bottom of form correctly (or at all)
- Always answering "Yes" to the questions at the bottom of the form without actually looking at the numbers

FILLING OUT THE MONTHLY REPORT - COMMON MISTAKES

- Rounding errors when using EPA tables to determine CT_{required}
 - Must round <u>down</u> for temperature
 - Must round <u>up</u> for pH
 - Must round <u>up</u> for free chlorine residual

- Bad CT formulas in excel spreadsheets:
 - Make sure you understand your formula
 - Wilkes Equation not allowed, must use Regression Equation

FILLING OUT THE MONTHLY REPORT - <u>AVOIDING</u> MISTAKES

- Check how T is calculated at your plant
- Do all treatment plant operators understand it?
- Review spreadsheet equation for CTs (if applicable)
- Write an SOP for CT determination
- Arrange for a tracer study if necessary
- Calculate CT and fill out monthly report daily
- Know what to do and who to call when things go wrong (contact State regulator & refer to Emergency Response Plan)

STRIVE TO IMPROVE DATA QUALITY

- Make data reliability a plant goal
- Only collect data used for process control or compliance reporting
- Establish protocols for collection and recording of data
- Establish a data verification process that can be routinely used to confirm data integrity
- Turn data into information (e.g., draw the graph).



O&M MANUALS



Keep written procedures on:

- Instrument calibration methods and frequency
- Data handling/reporting
- Chemical dosage determinations
- Filter operation and cleaning
- CT determinations
- Responding to abnormal conditions (emergency response plan)

DISINFECTION

SIEMENS Watter & Tierrary Barrier M

- Types of disinfectants
 - Radiation (UV)
 - Chemical (chlorine, chloramines, chlorine dioxide, ozone)
- Forms of chlorine
- NSF/ANSI Standard 6o



TYPES OF DISINFECTANTS - UV

- Works by subjecting water to ultraviolet (UV) light rays as water passes through a tube
- Drawbacks:
 - Interfering agents such as turbidity can screen pathogens from the UV light
 - Effective against *Giardia* and *Cryptosporidium* but not viruses at normal doses
 - No residual is present throughout the distribution system
 - For this reason, chlorination for residual maintenance is required when UV is used



TYPES OF DISINFECTANTS - CHEMICAL

- 1. Chlorine
- 2. Chloramines
- 3. Chlorine dioxide
- 4. Ozone

TYPES OF DISINFECTANTS - CHLORINE

- The most widely used form of disinfection
- Also used as an oxidizing agent for iron, manganese and hydrogen sulfide and for controlling taste and odors
- Effectiveness as a disinfecting agent depends on factors such as pH, temperature, free chlorine residual, contact time and other interfering agents

FORMS OF CHLORINE

- Sodium Hypochlorite
- Onsite generated sodium hypochlorite
- Calcium Hypochlorite
- Chlorine Gas

FORMS OF CHLORINE - SODIUM HYPOCHLORITE

- The liquid form of chlorine
- Clear and has a slight yellow color
- Ordinary household bleach (~5% chlorine by solution) is the most common form
- Industrial strength: 12% and 15% solutions
- Can lose up to 4% of its available chlorine content per month; should not be stored for more than 60 to 90 days
- Very corrosive; should be stored and mixed away from equipment that can be damaged by corrosion

FORMS OF CHLORINE - SODIUM HYPOCHLORITE



Diaphragm pump with chlorine solution tank

ON-SITE GENERATED SODIUM HYPOCHLORITE

- o.8% sodium hypochlorite is produced on demand by combining salt, water & electricity
- Electrolysis of brine solution produces sodium hydroxide and chlorine gas, which then mix to form sodium hypochlorite
- Hydrogen gas byproduct; vented to atmosphere
- Alleviates safety concerns associated w/ hauling and storing bulk chlorine
- Higher initial cost, high power cost
- Mixed oxidants (proprietary)

ON-SITE GENERATED SODIUM HYPOCHLORITE



FORMS OF CHLORINE - CALCIUM HYPOCHLORITE

- The solid form of chlorine
- Usually tablet or powder form (see photo below)
- Contains ~65% chlorine by weight
- White or yellowish-white granular material and is

fairly soluble in water

- Important to keep in a dry, cool place
- More stable than liquid
- Common in small systems w/ low flows or no power



FORMS OF CHLORINE - CALCIUM HYPOCHLORITE



Erosion chlorinator

Inside Hopper =>

FORMS OF CHLORINE - CHLORINE GAS

- 99.5% pure chlorine
- yellow-green color 2.5x heavier than air
- Liquefied at room temperature at ~107 psi –
 hence the pressurized cylinders actually contain
 liquefied chlorine gas.
- Liquefied Cl₂ is released from tanks as chlorine gas, which is then injected into the water stream.
- usually used only by large water systems
- Smaller systems may find initial cost of operation prohibitive

FORMS OF CHLORINE - CHLORINE GAS



150-lb cylinders

Note security chain spare tank & labeling.

Below: 1 ton cylinders. Note scale used to monitor product use.



FORMS OF CHLORINE - CHLORAMINES

- Chlorine + ammonia = chloramination
- Two advantages to regular chlorination:
 - produce a longer lasting chlorine residual (helpful to systems with extensive distribution systems)
 - may produce fewer by-products depending on the application
- Disadvantage:
 - Need a lot of contact time to achieve CTs compared to free chlorine (300 times more) which is why not used for primary disinfection
 - Requires specific ratio of chlorine to ammonia or else potential water quality problems

CHLORINE DIOXIDE - ADVANTAGES

Advantages:

- 1. More effective than chlorine and chloramines for inactivation of viruses, *Cryptosporidium*, and *Giardia*
- 2. Oxidizes iron, manganese, and sulfides
- 3. May enhance the clarification process
- 4. Controls T&O resulting from algae and decaying vegetation, as well as phenolic compounds
- Under proper generation conditions halogen-substituted DBPs are not formed
- 6. Easy to generate
- 7. Provides residual

CHLORINE DIOXIDE - DISADVANTAGES

Disadvantages

- Forms the DBP chlorite
- 2. Costs associated with training, sampling, and laboratory testing for chlorite and chlorate are high
- 3. Equipment is typically rented, and the cost of the sodium chlorite is high
- 4. Explosive, so it must be generated on-site
- 5. Decomposes in sunlight
- 6. Can lead to production noxious odors in some systems.

OZONE

- Colorless gas (O₃)
- Strongest of the common disinfecting agents
- Also used for control of taste and odor



- Extremely Unstable; Must be generated on-site
- Manufactured by passing air or oxygen through two electrodes with high, alternating potential difference

OZONE - ADVANTAGES

Advantages:

- Short reaction time enables microbes (including viruses) to be killed within a few seconds
- 2. Removes color, taste, and odor causing compounds
- 3. Oxidizes iron and manganese
- 4. Destroys some algal toxins
- 5. Does not produce halogenated DBPs

OZONE - DISADVANTAGES

Disadvantages:

- 1. Overfeed or leak can be dangerous
- 2. Cost is high compared with chlorination
- 3. Installation can be complicated
- 4. May produce undesirable brominated byproducts in source waters containing bromide
- 5. No residual effect is present in the distribution system, thus post-chlorination is required
- 6. Much less soluble in water than chlorine; thus special mixing devices are necessary

NSF/ANSI STANDARD 60 - CHEMICALS

- Addresses the health effects implications of treatment chemicals and related impurities.
- The two principal questions addressed are:
 - Is the chemical safe at the maximum dose, and
 - 2. Are impurities below the maximum acceptable levels?



NSF/ANSI STANDARD 60 - CHEMICALS

http://info.nsf.org/Certified/PwsChemicals/

NSF/ANSI 60 Drinking Water Treatment Chemicals - Health Effects

Morton Salt, Inc.

123 North Wacker Drive Chicago, IL 60606-1743 United States 312-807-2000

Facility: 0111 Glendale, AZ

Sodium Chloride[1] [CL]

Trade Designation Product Function Other Bulk White Crystal Solar Salt Bulk w/Crystal Crse. Sol. Salt Other Fine Solar Salt Other Morton® Commercial Grade Water Softening Pellets Other Other Morton® System Saver® II Formula Pellets White Crystal Solar Salt Other White Crystal Water Softening Solar Salt Other



NSF/ANSI STANDARD 61 - COMPONENTS & MEDIA

http://info.nsf.org/Certified/Pws Components/index.asp?stand ard=061

Cemex

5180 Golden Foothill Parkway Suite 200 El Dorado Hills, CA 95762 United States 925-426-8787

Facility: Marina, CA

Size

.2 mm - 3 mm



Trade Designation Size

Sand

Lapis Lustre Sand .2 mm - 3 mm

Water Water
Contact Contact
Temp Material

NET WEIGHT - 100 LBS.

(45.4 kg)

CLEAN-GRADED-KILN DRIED MONTEREY SANDS

CALIFORNIA ARB CERTIFIED FOR PERMISSIBLE DRY OUTDOOR BLASTING

CLD 23 SLDOX

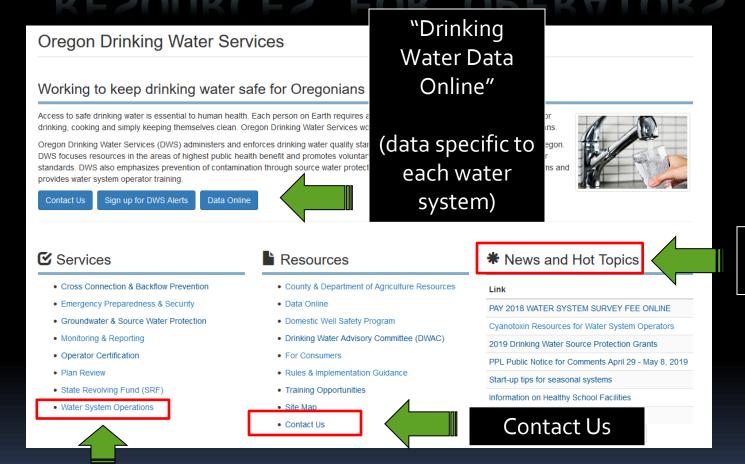
[1] Certified products include F-101 through F-112.

NOTE: Certified for water treatment plant applications.

This product has not been evaluated for point of use applications.

- For surface water systems:www.healthoregon.gov/dwp
- Click on "Water System Operations" on left-side menu list, then "Surface Water Treatment"
 - Monthly Surface Water Quality Report form template
 - Tracer Study form
- Surface Water Treatment Rule guidance manual, Appendix C: Determination of Disinfectant Contact Time

- EPA Rules
 <u>http://water.epa.gov/lawsregs/rulesregs/sdwa/cu</u>
 rrentregulations.cfm
- AWWA http://www.pnws-awwa.org/Index.asp
- OAWU http://www.oawu.net/
- Circuit Rider
 http://public.health.oregon.gov/HealthyEnvironments/DrinkingWater/Operations/Pages/circuitrider.aspx



Current News and Events

Water System Operations

www.healthoregon.gov/dwp

Resources for Oregon Water System Operators

Drinking Water Services

Water System Operations

Surface Water Treatment

Capacity Development

Public Notice Resources & Templates

Fact Sheets & Best Management Practices

Water System Surveys & Outstanding Performance

Circuit Rider Program

Pipeline Newsletter

Contact Us

Surface Water Treatment

Water systems that treat surface water sources have to deal with complex regulatory requirements, constantly changing raw water quality, and costly management of various assets. The Surface Water Treatment site provides information and tools needed to optimize water treatment processes and maximize public health protection without costly capital improvements.

Capacity Development

Water system capacity is the technical, managerial and financial capability of a water system to achieve and maintain compliance with drinking water standards and consistently provide safe drinking water. The Capacity Development site provides information and resources for drinking water systems to help build their capacity.

Public Notice Resources & Templates

Water systems are required to issue public notices to alert consumers under specific circumstances (for example, when exceeding a Maximum Contaminant Level, failing to complete required tests, failing to report the results, or failing to meet treatment technique requirements). This page includes information on public notification requirements and templates for issuing public notices, as well as translations and FAQs for effective communication with partners and the public.

Fact Sheets & Best Management Practices

Key Resources

- . Drinking Water Data Online
- For Consumers
- Site Map

Oregon Drinking Water Services

Working to keep drinking water safe for Oregonians

Access to safe drinking water is essential to human health. Each person on Earth requires a drinking, cooking and simply keeping themselves clean. Oregon Drinking Water Services we

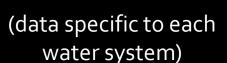
Oregon Drinking Water Services (DWS) administers and enforces drinking water quality star DWS focuses resources in the areas of highest public health benefit and promotes voluntar standards. DWS also emphasizes prevention of contamination through source water protect provides water system operator training.

Contact Us

Sign up for DWS Alerts

Data Online

"Drinking Water Data Online"





Services

- · Cross Connection & Backflow Prevention
- · Emergency Preparedness & Security
- · Groundwater & Source Water Protection
- · Monitoring & Reporting
- Operator Certification
- Plan Review
- · State Revolving Fund (SRF)
- · Water System Operations

Resources

- · County & Department of Agriculture Resources
- Data Online
- . Domestic Well Safety Program
- . Drinking Water Advisory Committee (DWAC)
- For Consumers
- · Rules & Implementation Guidance
- · Training Opportunities
- Site Map
- Contact Us

* News and Hot Topics

Link

PAY 2018 WATER SYSTEM SURVEY FEE ONLINE

Cyanotoxin Resources for Water System Operators

2019 Drinking Water Source Protection Grants

PPL Public Notice for Comments April 29 - May 8, 2019

Start-up tips for seasonal systems

Information on Healthy School Facilities

Pipeline Newsletter, October 2018

View archived hot topics and news items

www.healthoregon.gov/dwp

DATA FOR EACH SYSTEM ON-LINE



Health

Introduction : Da

Data Search Options

Data Search Options

Drinking Water Data Online

Welcome to Data Online, the data access site for Oregon Drinking Water Services.

Here you can access a fair amount of data on public drinking water systems in Oregon. You can find data chemical testing, contacts, violations, enforcements, public notices, and basic system information.

Info by County

The <u>Search Options</u> page explains many of the data pages that are available. Use the <u>Water System Se</u> system by water system ID number, name, or location.

Data shown here are "live" data. That means they're as current as the reports we have in our system. This is the same do not be a system of the system of the system personnel, county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.) find a report is missing, please forward a county staff, lab staff, etc.)

For questions or updates regarding water system sampling, inventory, or compliance, please contact Drinking Water Ser 971-673-0405 or lnfo.Drinking-water@odhsoha.oregon.gov.

See the Contact Us page on the main Drinking Water site for more contact options.

Information for all water systems:

Water System Inventory :: Water System Surveys :: Outstanding Performers :: Treatment Plant Inspections :: Treatment :: Plan Reviews

Alerts :: Violations :: Compliance & Enforcement :: Deficiencies :: System Scores :: Exceedances :: Public Notices

Water Advisories :: Contact Reports :: Cyanotoxins :: PFAS :: Post-wildfire VOCs :: Fluoride

Introduction :: Data Search Options :: Water System Search :: DWS Home :: DWS Rules

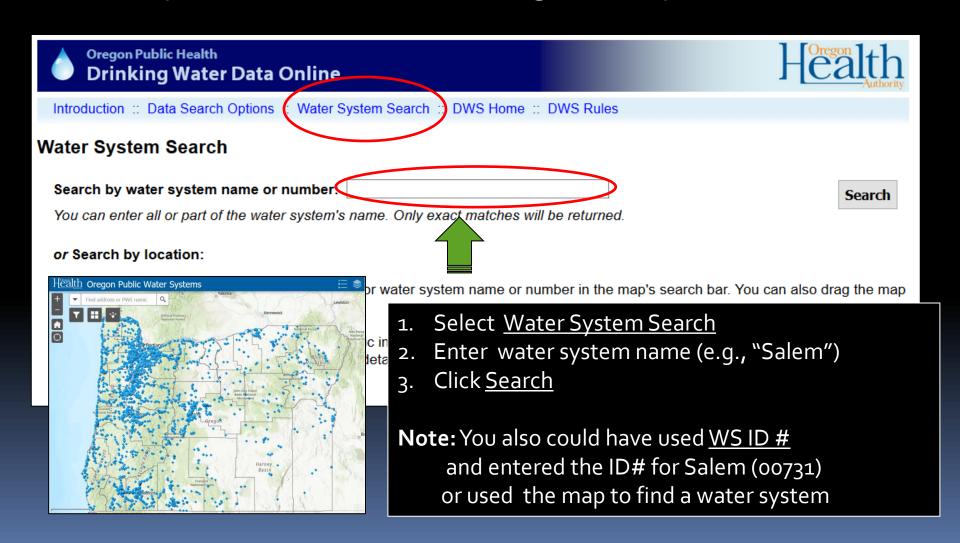
Need help? Email Drinking Water Services for assistance.

Staff/Partner Login

s at

DATA FOR EACH SYSTEM ON-LINE

Water System Search (search using water system name or ID#



DATA FOR EACH SYSTEM ON-LINE

Select the Water System by Clicking on the PWS ID# 00731

Search results: 5 systems found. Select a water system by clicking on its row (opens in a new tab).

PWS ID	Water System Name	Regulating Agency	County Served	System Type	Activity Status
05564	BPA-SALEM SUBSTATION	POLK COUNTY	Polk	ovs	Inactive
95003	FORUM SALEM CAMPUS	DEPT OF AGRICULTURE	Marion	NTNC	Active
00779	SALEM MOBILE ESTATES/SHADY ACRES	MARION COUNTY	Marion	С	Active
00731	SALEM PUBLIC WORKS	REGION 1	Marion	С	Active
00768	SUBURBAN EAST SALEM WD	REGION 1	Marion	С	Active

OR41 00731	SALEM PU	BLIC WORKS		Classifica			
Contact:	DWAYNE BARNES		Phone: 5	Phone: 503-588-6483 View on Map			
	PO BOX 14300		County: 1	County: MARION			
	SALEM, OR 97309		Activity S	Status: ACTIVE History			
Population: 199,820			Number	Number of Connections: 55,970			General
Operating Period: January 1 to December 31			Regulati	Regulating Agency: REGION 1			Information
Certified Operator(s)			Owner T	Owner Type: LOCAL GOVERNMENT			IIIIOIIIIatioii
Required: Y			Licensed	Licensed By: N/A			
	Distribution class: 4						
	Treatment class: 3		Source V	Nater Assessment: Yes			
	Filtration Endorsement F	Required: No	Last Sun	vey Date: Jul 25, 2023 - O			
Facility ID	Facility Name - We	ell Logs		Activity Status	<u>Availability</u>	Source Type	
EP-A	EP FOR GEREN ISLAND (ALDERSGATE)			Α		SW	
SRC-AA	NORTH SANTIAM RIVER			Α	Permanent	SW	
SRC-AB	GEREN ISLAND EAST WELL - L75842			Α	Permanent	GU	
SRC-AC	GEREN ISLAND WEST WELL - L75839			Α	Permanent	GU	
SRC-AD	I.G. / ROUGHING FILTER #1			Α	Seasonal	GU	
EP-B	EP FOR ASR WELLS			Α		GW	Sources
SRC-BA	ASR WELL #1 - L82685			Α	Seasonal	GW	3001003
SRC-BB	ASR WELL #2 - L82688			Α	Seasonal	GW	
SRC-BC	ASR WELL #4 - L10522			Α	Seasonal	GW	
SRC-BD	ASR WELL #5 - L16342			Α	Seasonal	GW	
EP-C	EP FOR HEMLOCK WELL			1		GW	
SRC-CA	HEMLOCK WELL - L	.62600		T	Emergency	GW	
						Find Purchasers/Sellers	
Facility ID	Facility Name	Filter Type G	iardia Removal <u>Credit</u>	Treatment Process		Treatment Objective	
WTP-A	TP FOR GEREN ISLAND	SLOW SAND	2.0-log	FILTRATION, SLOW SAN HYPOCHLORINATION, F PH/ALKA ADJ-SODA AS OZONATION, PRE	POST	PARTICULATE REMOVAL DISINFECTION CORROSION CONTROL DISINFECTION	Treatment

ACT. CARBON, PWD - CYANOTOXINS OTHER

System Classification

Classification: COMMUNITY



OR41 00731 SALEM PUBLIC WORKS

Contact: DWAYNE BARNES

PO BOX 14300

SALEM, OR 97309

Population: 199,820

Operating Period: January 1 to December 31

Certified Operator(s)

Required: Y

Distribution class: 4

Treatment class: 3

Filtration Endorsement Required: No

Phone: 503-588-6483

County: MARION

Activity Status: ACTIVE -- History

Number of Connections: 55,970

Regulating Agency: REGION 1

Owner Type: LOCAL GOVERNMENT

Licensed By: N/A

Approved Drinking Water Protection Plan: No

Source Water Assessment: Yes

Last Survey Date: Jul 25, 2023 - Outstanding Performer!

All written correspondence goes to this person (e.g., violation notices, general mailings, etc.)

View a list of Certified Operators



More information for this water system:

DWIS ID 3906

System Info :: Report for Lenders : Alerts :: Violations :: Compliance & Enforcement :: Contacts & Advisories :: Site Visits :: Public Notice

Coliform Summary :: Coliform Results :: Coliform Schedules :: LT2 :: GW/GWUDI Source Details :: Plan Review :: Annual Fee

Chemical Summary :: Chemical Results :: Chemical Schedules :: Chemical Schedule Summary :: Arsenic RAA :: Cyanotoxins :: PFAS

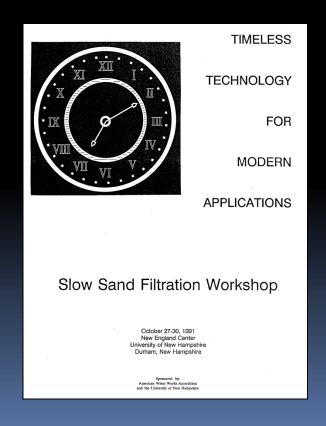
Lead & Copper :: Corrosion Control (LCR) :: DBP Sample Sites :: FANLs :: MRDL :: GWR 4-Log :: Turbidity :: SWTR :: LRAA

- 1. Alerts, Violations, & Compliance
- 2. Contacts, Advisories, & Public Notices
- 3. Site Visits
- 4. Plan Review
- Coliform and Chemical Sampling Schedules
- 6. Sampling Results



MORE QUESTIONS?

- Call your technical services contact at the State.
 State Drinking Water Services
 - General Info: (971) 673-0405





Astoria, OR 5 MGD plant (photo taken by Frank Wolf)