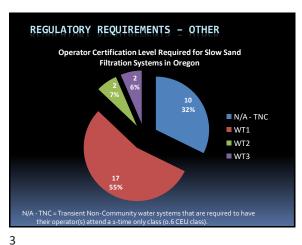


REGULATORY REQUIREMENTS Plan Review - OAR 333-061-0050(4)(c)(C) & (E) (C) Pilot studies shall be conducted by the water supplier to demonstrate the effectiveness of any filtration method other than conventional filtration. Pilot study protocol shall be approve advance by the Authority. Results of the pilot study shall be submitted to the Authority for review and approval. (E) All filtration systems shall be designed and operated so as to meet the requirements in OAR 333-061-0032(4) and (5) – i.e., meet turbidity limits and CT requirements. Design of the filtration system must be in keeping with accepted standard engineering references acknowledged by the Authority such as the Ten State dards, technical reports by the International Reference Center for Community Water Supply and Sanitation ("IRC publications from the World Health Organization ("WHO manual").

1 2



REGULATORY REQUIREMENTS 40 CFR 141.70 — 141.75 (applies to all SW and GWUDI systems a.k.a "Subpart H" systems) Required 3.0-log (99.9%) Giardia and 4-log (99.99%) virus removal/inactivation (filtration plus disinfection) Established turbidity limits (\leq 1 NTU in 95% of readings w/all \leq 5 NTU) Established disinfectant residual requirements 40 CFR 141.170 - 141.175 Added 2.0-log cryptosporidium treatment requirements 40 CFR 141.500 - 141.571 Extended IESWTR requirements for systems < 10,000 pop 40 CFR 141.700 – 141.723 & 40 CFR 141.211, Appendix A to Subpart Q Additional Cryptosporidium treatment requirements depending upon source sampling and resultant bin classification (more treatment if higher than bin 2) Addressed uncovered finished water reservoirs

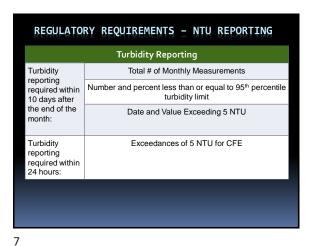
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REGULATORY REQUIREMENTS -PATHOGEN RMVL Applicability: PWSs that use SW or GWUDI that practice SSF, DE, or Regulated Pathogen 99.99% (4-log) removal/inactivation of viruses (SWTR) 99.9% (3-log) removal/inactivation of Giardia lamblia (SWTR) 99% (2-log) removal of Cryptosporidium (IESWTR/LT1) (> 2-log if Bin 2 or higher under LT2) Slow sand filtration is credited with removing: • 2.0-log Giardia & 2-log Cryptosporidium 1.0-log Giardia inactivation is needed through disinfection, 0.5-log of which must be obtained after filtration.

5

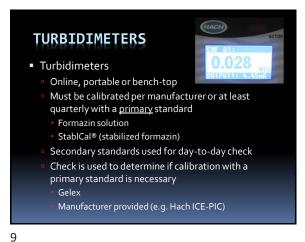
REGULATORY REQUIREMENTS - TURBIDITY **Turbidity Limits** Turbidity Turbidity readings are to 95% of CFE turbidity be monitored/recorded readings ≤ 1 NTU at the combined filter (≤ 1.49 NTU) effluent (CFE) at a frequency of at least once every 4 hours* All CFE turbidity readings ≤ 5 NTU (< 5.49 NTU) * Frequency may be reduced by the State to once per day.

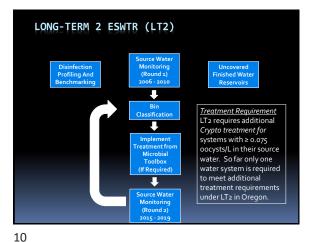
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SLOW SAND - ABLE TO MEET 1 NTU LIMIT Membrane 0.12 Average 95th percentile turbidity based on the ■ Conventional Filtration 0.13 ■ Direct Filtration maximum turbidity DE 0.15 measured daily for each of ■ Slow Sand the surface water 0.35 treatment plants in Oregon Cartridge 0.36 from July 1, 2011 - June 30, 2012 0.67 0.2 0.4 0.6 0.8 Average 95th Percentile Turbidity (NTU)

8

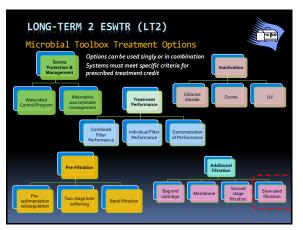




40 CFR 1	IG-TERM 2 ESWTR (L ⁻¹ 41-701(c) Monitoring Schedule 4 second round monitoring must begin isted in the table below.	1 3 9	n beginning with		
Schedule	Systems that serve	1 st Round	2 nd Round		
1	At least 100,000 people*	October 1, 2006	April 1, 2015		
2	From 50,000 to 99,999 people*	April 1, 2007	October 1, 2015		
3	From 10,000 to 49,999 people*	April 1, 2008 October 1, 2016			
4 (E. coli)	Fewer than 10,000, not a wholesale system, and monitors for <i>E. coli</i> ^a	October 1, 2008	October 1, 2017		
4 (Crypto)	Fewer than 10,000, not a wholesale system, and monitors for <i>Cryptosporidium</i> ^b	April 1, 2010	April 1, 2019		
^a Applies only	to wholesalers in a combined distribution syste to filtered systems. Itered systems that meet the conditions of para				

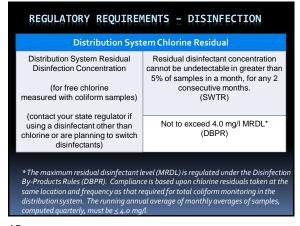
	ered System Additional Cryptosporidi (based on their bin classification as de and according to the sched	termined under § 12				
bin	Conventional Filtration (including softening), Slow Sand, or Diatomaceous Earth	Direct filtration	Alternative filtration technologies			
Bin 1	No Additional Treatment					
Bin 2	1-log treatment	1.5-log treatment	RMVL + Inactivation ≥ 4.0-log¹			
Bin 3	2-log treatment	2.5-log treatment	RMVL + Inactivation ≥ 5.0-log ²			
Bin 4	2.5-log treatment	3-log treatment	RMVL + Inactivation ≥ 5.5-log ³			

11 12



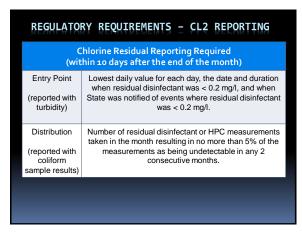
REGULATORY REQUIREMENTS - DISINFECTION **Entry Point Chlorine Residual** Entry Point Residual Residual disinfectant concentration cannot be < Disinfection 0.2 mg/l for more than 4 hours based on Concentration continuous monitoring (> 3,300 pop) or less frequent monitoring as allowed by the state. (for free chlorine (SWTR) measured prior to or at the first customer each (contact your state regulator if using a day of operation) disinfectant other than chlorine or are planning to switch disinfectants) No two consecutive daily samples should exceed 4.0 mg/l (DBPR) Where chlorine is used as the disinfectant, the measurement of residual chlorine shall be by the <u>DPD or other EPA-approved method</u> in accordance with Standard Methods for the Examination of Water and Waste-water, and shall measure the free chlorine residual or total chlorine residual as applicable

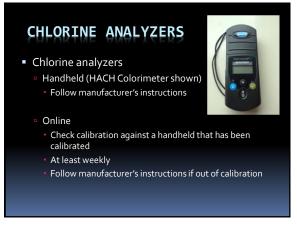
13 14



REGULATORY REQUIREMENTS - CL2 REPORTING Additional Distribution Residuals Monitoring 2x per week Distribution All public water systems that add a disinfectant to the water supply at any point in the treatment process, or (records to be deliver water in which a disinfectant has been added to kept by the the water supply, must maintain a detectable water system for at least 2 disinfectant residual throughout the distribution system and shall measure and record the residual at one or more representative points at a frequency that is vears) sufficient to detect variations in chlorine demand and changes in water flow but in no case less often than twice per week.

15 16





17 18

REGULATORY REQUI	REMENTS - OTHER			
Other SWTR/IESWTR/LT1 Requirements				
Disinfection Profiling & Benchmarking	Systems must profile inactivation levels and generate a benchmark, if required due to disinfection changes (IESWTR & LT1)			
Water System Surveys (State Requirement)	CWS: Every 3 years NCWS: Every 5 years (IESWTR & LT1)			
Finished Water Reservoirs	New (post-1989) reservoirs must be covered under SWTR. Pre-SWTR reservoirs must be covered (or have additional treatment) under LT2			
Operator Certification	Operated by Qualified Personnel as Specified by State (SWTR)			
(CWS) Community Water System (I	NCWS) Non-community Water System			

Cyanotoxin Monitoring (QAR 333-061-0510 to -0580)
Healthoregon.org/dwcyanotoxins

Who does this apply to?

Affects systems who have sources susceptible to cyanobacteria blooms (not everyone). See list systems and specific rule requirements on-line at www.healthoregon.org/dwcyanotoxins

Raw water (intake) sampling for total microcystin and Cylindrospermopsin toxins every 2 weeks from May 1st — October 31st each year

What happens
If detected?

If any toxins are greater than or equal to 0.3 µg/L in raw water or if there is a recreational use health advisory* upstream of the intake, sample raw and entry point weekly with the first EP sample taken within 1 business day. Weekly sampling continues until non-detect at EP and less than 0.3 µg/L in raw water in two consecutive samples.

If detected at EP sample PE daily and optimize treatment for toxin removal.

If above Health Advisory Level (HAL) at EP take confirmation sample within 24-hrs & monitor EP daily.

If confirmation sample is above the HAL, issue Do-Not-Drink Advisory

Advisory may only be lifted if 2 consecutive daily EP samples taken a minimum of 24-hrs apart are https://www.healthoregov.nayon.na

19 20

REVIEW

21

- 2.o-log Cryptosporidium removal is required (and credited) for slow sand filtration.
- Surface Water Treatment Rule (SWTR) requires 3log reduction of Giardia using a combination of disinfection and filtration and 4.0-log reduction of viruses.
- At least 2.0 -log Giardia removal is credited for slow sand filtration (per 1991 USEPA SWTR Manual)
- 1.o-log Giardia inactivation must be achieved through disinfection (o.5-log must be after filtration).
 1.o-log reduction of viruses must also be achieved after filtration.

REPORTING FORMS - CFE TURBIDITY OHA - Drinking Water Program - Turbidity Monitoring Report Form County: Slow Sand, Membrane, Diatomaceous Earth Filtration, or Unfiltered Systems System Name ID#: WTP-: Month/Year: 8 PM [NTU] 0.24 0.44 Notify the State within 24-hrs if turbidity > 5 NTU (includes after hours) Public Health After Hours Duty Officer: Cell (971) 246-1789 Pager (503) 938-6790 Oregon Emergency Response System: Chose time closest to when daily turbidity is measured and enter Enter highest turbidity of all measurements for the day (e.g., online instrument or highest of multiple daily grab samples)

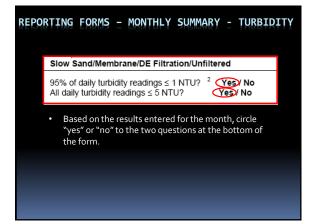
REPORTING FORMS

There are 4 forms:

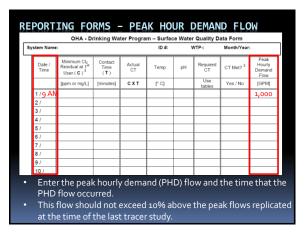
- Conventional/Direct
- Slow Sand / Membrane / DE / Unfiltered
- Cartridge
- UV (if used for Giardia credit)

Must use correct form because each has questions that must be answered that are specific to the filtration type

22



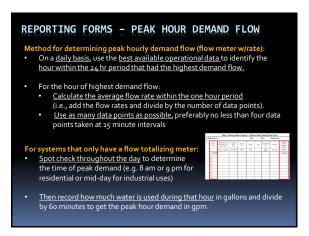
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REPORTING FORMS – PEAK HOUR DEMAND FLOW

| Control | Con

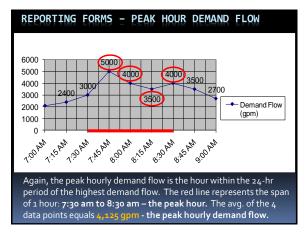
25 26



REPORTING FORMS - PEAK HOUR DEMAND FLOW 6000 5000 4000 3000 Demand Flow (gpm) 2000 1000 8:00 AM 8:15 AM 830AM 7:75 AM Here is an example chart, meant to represent continuous readings that shows demand flow out of a reservoir used for contact time. What would you say the peak hourly demand flow is?

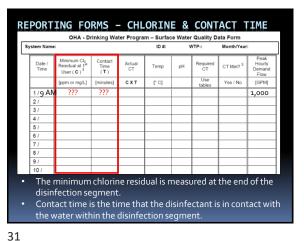
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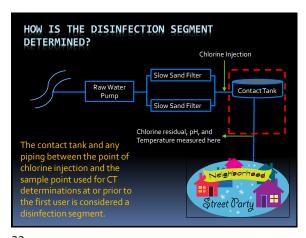
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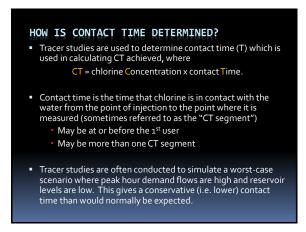


REPORTING FORMS - PEAK HOUR DEMAND FLOW 6000 running hourly 5000 everage of demand flow measurements. 4000 3500 4000 3500 3000 **Running Hourly Average** Time Demand Flow (gpm) 7:00 AM 2,000 7:15 AM 2,400 7:30 AM 3,000 7:45 AM 5,000 3,100 8:00 AM 4,000 3,600 8:15 AM 3,500 3,875 8:30 AM 4,000 4,125 8:45 AM 3,500 3,750 9:00 AM 2,700 3,425

29 30

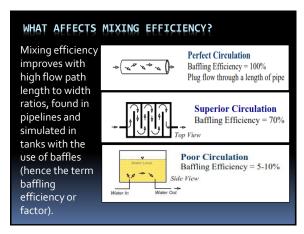


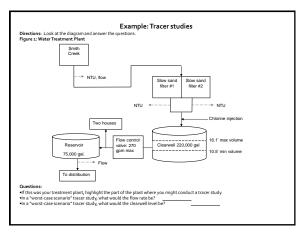




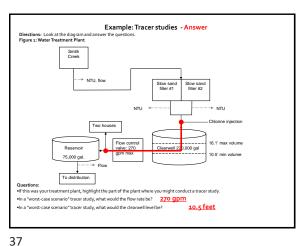
HOW IS CONTACT TIME DETERMINED? • The more efficient the mixing is in a reservoir or tank, the more contact time is available for disinfection. • Estimates of contact time based on tank or reservoir design are not allowed for calculating CT's for surface water!

33 34





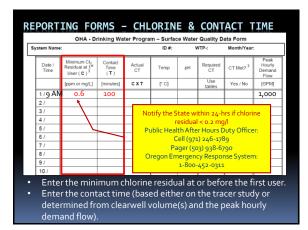
35 36



DO I REPORT CONTACT TIME?

- Use the time T from the tracer study on the monthly reporting form in the "Contact time (min)" column
 - Use the smallest T (highest flow) if the tracer study was done at multiple flow rates
- This may not be your exact time, but it represents your worst case (as long as the peak flow is less and clearwell volume is more than they were at the time of the tracer study)

38



CAN I USE A BAFFLING FACTOR?

- As an alternative to using the tracer study contact time, you can use the results of the tracer study to determine the baffling factor of the clearwell
 - Baffling factor (%) = <u>Time (min) x Flow During Tracer Study (gpm)</u>
 Clearwell Volume During Tracer Study (gal)
- T can be adjusted based on flow (at flow < 110% of tracer study flow) with the following equation:
 - T = Current clearwell Volume (gal) x Baffling Factor (%) Peak Hourly Demand Flow (gpm)
- Contact the state for guidance on using baffling factors.

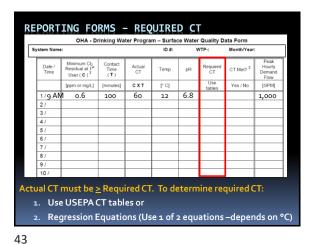
39 40

System Nam	e:			ID #:	V	VTP-:	Month/Year	:
Date / Time	Minimum CI ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? 3	Peak Hourly Demand Flow
	[ppm or mg/L]	[minutes]	СХТ	[° C]		Use tables	Yes / No	[GPM]
1/9 Al	VI 0.6	100	60					1,000
2/								
3 /								
4 /								
5/								
6 /								
7 /								
8 /								
9 /								
10 /								
		1.07		d that				

REPORTING FORMS - TEMPERATURE & PH OHA - Drinking Water Program – Surface Water Quality Data Form Residual at 1^s User (C) ³ Time (T) CT Met2 Flow схт [° C] Yes / No [GPM] 0.6 100 60 1,000 1/9 AN 12 8 / Enter the finished water temperature (°C) and pH measured at

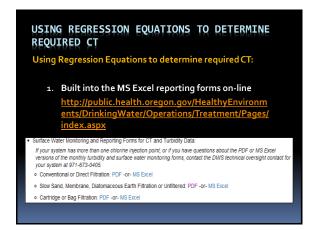
or prior to the first customer and after any storage (tank, reservoir, or pipeline) used for contact time.

41 42



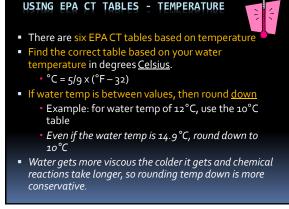
HOW IS REQUIRED CT CALCULATED? • We use the EPA tables (or "regression equations") to determine the CT required to inactivate Giardia (CT_{required}) 1-log inactivation of Giardia using chlorine results in at least 4.0-log inactivation of viruses. To determine CT, we need to know pH, temperature, and free chlorine residual at or before the first user. Then we compare the CT_{required} with the actual CT achieved in the water system (CT_{actual}) where: CT_{actual} = chlorine concentration (mg/l) x contact time (min) ■ Must keep CT_{actual} ≥ CT_{required}

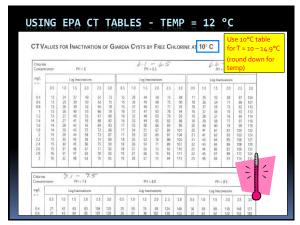
44



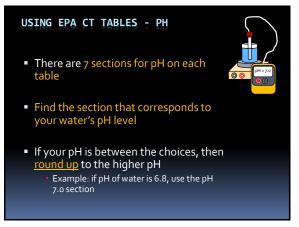
USING REGRESSION EQUATIONS, CONT. Using Regression Equations to determine required CT: 2. Regression equations can be programmed into plant SCADA or spreadsheets Regression Equation (for Temp < 12.5°C) CT = (0.353*L)(12.006 + e(2.46 Regression Equation (for Temp > 12.5°C) $CT = (0.361*L)(-2.261 + e^{(2.69-1)})$ CT = Product of Free Chlorine Residual and Time required L = number of log inactivation for Giardia (L = 1 for slow sand) T = temperature, in Celsius C = chlorine residual in mg/L pH = pH of water e = 2.7183, base for natural log (Smith, Clark, Pierce and Regli, 1995, from EPA's 1999 Guidance Manual for Disinfection Profiling and Benchmarking)

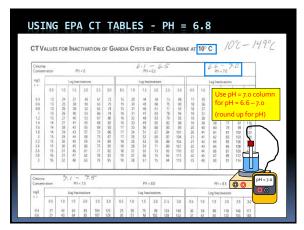
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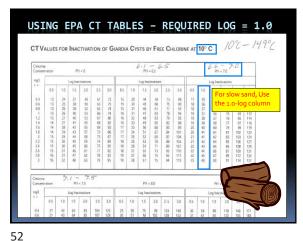


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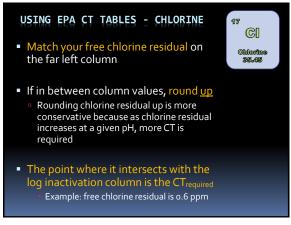


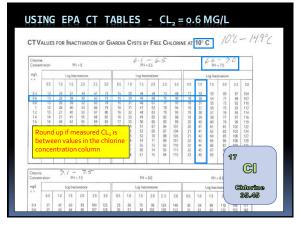




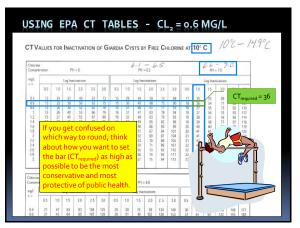


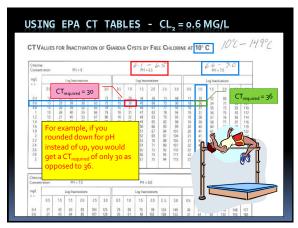
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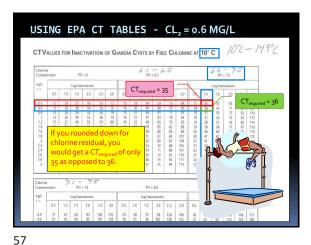


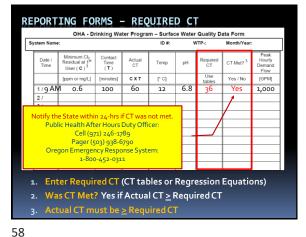


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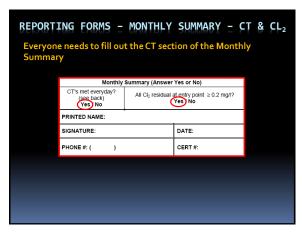


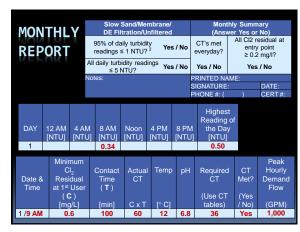






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59 60

FILLING OUT THE MONTHLY REPORT - COMMON MISTAKES

- Not calculating CT's daily
 - Don't wait until the end of the month to do the calculations because if you discover you didn't meet CT's, it's too late!
- If adjusting contact time according to flow rate, use the demand flow, not the plant flow.
- Failure to answer questions at bottom of form correctly (or
- Always answering "Yes" to the questions at the bottom of the form without actually looking at the numbers

61

FILLING OUT THE MONTHLY **REPORT - AVOIDING MISTAKES**

- Check how T is calculated at your plant
- Do all treatment plant operators understand it?
- Review spreadsheet equation for CTs (if applicable)
- Arrange for a tracer study if necessary
- Calculate CT and fill out monthly report daily
- wrong (contact State regulator & refer to **Emergency Response Plan)**

63

Write an SOP for CT determination

Know what to do and who to call when things go

FILLING OUT THE MONTHLY **REPORT - COMMON MISTAKES**

- Rounding errors when using EPA tables to determine CT_{required}
 - Must round <u>down</u> for temperature
 - Must round up for pH
 - Must round up for free chlorine residual
- Bad CT formulas in excel spreadsheets:
 - Make sure you understand your formula Wilkes Equation not allowed, must use Regression

62

STRIVE TO IMPROVE DATA QUALITY

- Make data reliability a plant goal
- Only collect data used for process control or compliance reporting
- Establish protocols for collection and recording of data
- Establish a data verification process that can be routinely used to confirm data integrity
- Turn data into information (e.g., draw the



64

O&M MANUALS

Keep written procedures on:

- Instrument calibration methods and frequency
- Data handling/reporting
- Chemical dosage determinations
- Filter operation and cleaning
- CT determinations
- Responding to abnormal conditions (emergency response plan)

DISINFECTION





- Chemical (chlorine, chloramines, chlorine dioxide, ozone)
- Forms of chlorine
- NSF/ANSI Standard 6o



65 66

TYPES OF DISINFECTANTS - UV

- Works by subjecting water to ultraviolet (UV) light rays as water passes through a tube
- Drawbacks:
 - Interfering agents such as turbidity can screen pathogens from the UV light
 - Effective against Giardia and Cryptosporidium but not viruses at normal doses
 - No residual is present throughout the distribution system
 - For this reason, chlorination for residual maintenance is required when UV is used



Chlorine
 Chloramines
 Chlorine dioxide
 Ozone

TYPES OF DISINFECTANTS - CHEMICAL

67 68

TYPES OF DISINFECTANTS - CHLORINE

- The most widely used form of disinfection
- Also used as an oxidizing agent for iron, manganese and hydrogen sulfide and for controlling taste and odors
- Effectiveness as a disinfecting agent depends on factors such as pH, temperature, free chlorine residual, contact time and other interfering agents

FORMS OF CHLORINE

- Sodium Hypochlorite
- Onsite generated sodium hypochlorite
- Calcium Hypochlorite
- Chlorine Gas

69 70

FORMS OF CHLORINE - SODIUM HYPOCHLORITE

- The liquid form of chlorine
- Clear and has a slight yellow color
- Ordinary household bleach (~5% chlorine by solution) is the most common form
- Industrial strength: 12% and 15% solutions
- Can lose up to 4% of its available chlorine content per month; should not be stored for more than 60 to 90 days
- Very corrosive; should be stored and mixed away from equipment that can be damaged by corrosion

FORMS OF CHLORINE - SODIUM HYPOCHLORITE



Diaphragm pump with chlorine solution tank

71 72

ON-SITE GENERATED SODIUM HYPOCHLORITE

- o.8% sodium hypochlorite is produced on demand by combining salt, water & electricity
- Electrolysis of brine solution produces sodium hydroxide and chlorine gas, which then mix to form sodium hypochlorite
- Hydrogen gas byproduct; vented to atmosphere
- Alleviates safety concerns associated w/ hauling and storing bulk chlorine
- Higher initial cost, high power cost
- Mixed oxidants (proprietary)





FORMS OF CHLORINE - CALCIUM HYPOCHLORITE

- The solid form of chlorine
- Usually tablet or powder form (see photo below)
- Contains ~65% chlorine by weight
- White or yellowish-white granular material and is fairly soluble in water
- Important to keep in a dry, cool place
- More stable than liquid
- Common in small systems w/ low flows or no power



75

Erosion chlorinator Inside Hopper => 76

FORMS OF CHLORINE - CALCIUM HYPOCHLORITE

ON-SITE GENERATED SODIUM HYPOCHLORITE

Electrodes

FORMS OF CHLORINE - CHLORINE GAS

- 99.5% pure chlorine
- yellow-green color 2.5x heavier than air
- Liquefied at room temperature at ~107 psi hence the pressurized cylinders actually contain liquefied chlorine gas.
- Liquefied Cl₂ is released from tanks as chlorine gas, which is then injected into the water stream.
- usually used only by large water systems
- Smaller systems may find initial cost of operation prohibitive

FORMS OF CHLORINE - CHLORINE GAS Below: 1 ton cylinders. Note scale used to monitor product use. 150-lb cylinders Note security chain spare tank & labeling. 78

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FORMS OF CHLORINE - CHLORAMINES

- Chlorine + ammonia = chloramination
- Two advantages to regular chlorination:
 - produce a longer lasting chlorine residual (helpful to systems with extensive distribution systems)
 - may produce fewer by-products depending on the application
- Disadvantage:
 - Need a lot of contact time to achieve CTs compared to free chlorine (300 times more) which is why not used for primary disinfection
 - 2. Requires specific ratio of chlorine to ammonia or else potential water quality problems

CHLORINE DIOXIDE - ADVANTAGES

Advantages:

- More effective than chlorine and chloramines for inactivation of viruses, Cryptosporidium, and Giardia
- 2. Oxidizes iron, manganese, and sulfides
- 3. May enhance the clarification process
- 4. Controls T&O resulting from algae and decaying vegetation, as well as phenolic compounds
- 5. Under proper generation conditions halogen-substituted DBPs are not formed
- 6. Easy to generate
- 7. Provides residual

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CHLORINE DIOXIDE - DISADVANTAGES

Disadvantages

- 1. Forms the DBP chlorite
- Costs associated with training, sampling, and laboratory testing for chlorite and chlorate are high
- 3. Equipment is typically rented, and the cost of the sodium chlorite is high
- 4. Explosive, so it must be generated on-site
- 5. Decomposes in sunlight
- 6. Can lead to production noxious odors in some systems.

OZONE

- Colorless gas (O₃)
- Strongest of the common disinfecting agents
- Also used for control of taste and odor
- Extremely Unstable; Must be generated on-site
- Manufactured by passing air or oxygen through two electrodes with high, alternating potential difference

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OZONE - ADVANTAGES

Advantages:

- Short reaction time enables microbes (including viruses) to be killed within a few seconds
- Removes color, taste, and odor causing compounds
- 3. Oxidizes iron and manganese
- 4. Destroys some algal toxins
- 5. Does not produce halogenated DBPs

OZONE - DISADVANTAGES

Disadvantages:

- 1. Overfeed or leak can be dangerous
- 2. Cost is high compared with chlorination
- 3. Installation can be complicated
- May produce undesirable brominated byproducts in source waters containing bromide
- 5. No residual effect is present in the distribution system, thus post-chlorination is required
- 6. Much less soluble in water than chlorine; thus special mixing devices are necessary







RESOURCES FOR OPERATORS

• For surface water systems:
 www.healthoregon.gov/dwp

Click on "Water System Operations" on left-side menu list, then "Surface Water Treatment"

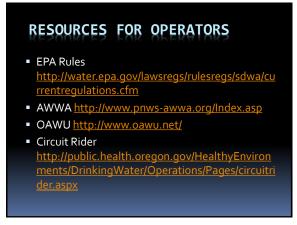
• Monthly Surface Water Quality Report form template

• Tracer Study form

• Surface Water Treatment Rule guidance manual, Appendix C: Determination of Disinfectant Contact Time

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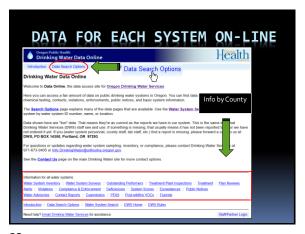


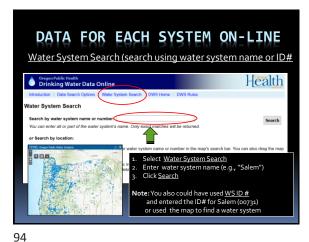


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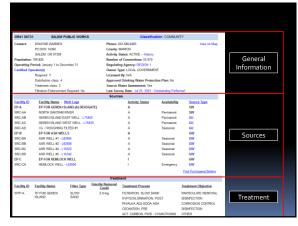






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DATA FOR EACH SYSTEM ON-LINE Select the Water System by Clicking on the PWS ID# 00731 PWS ID Water System Name Regulating Agency 05564 BPA-SALEM SUBSTATION POLK COUNTY Polk OVS Inactive DEPT OF AGRICULTURE Marion NTNC 95003 FORUM SALEM CAMPUS Active SALEM MOBILE ESTATES/SHADY ACRES MARION COUNTY Marion C Active REGION 1 SALEM PUBLIC WORKS Marion C Active 00768 SUBURBAN EAST SALEM WD REGION 1 Marion C Active



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