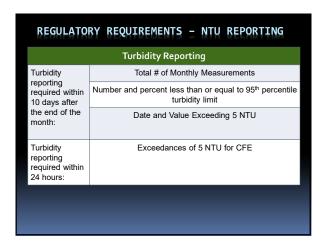
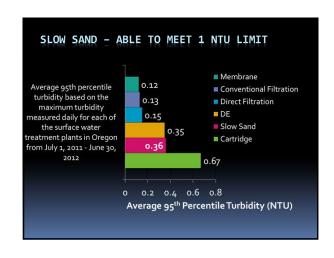
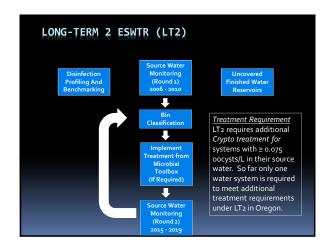


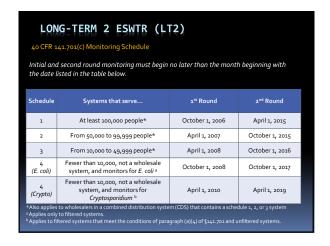
	Turbidity Limits	
Turbidity	Turbidity readings are to be monitored/recorded at the combined filter effluent (CFE) at a frequency of at least	95% of CFE turbidity readings ≤ 1 NTU (≤ 1.49 NTU)
	once every 4 hours*	All CFE turbidity readings ≤ 5 NTU (≤ 5.49 NTU)



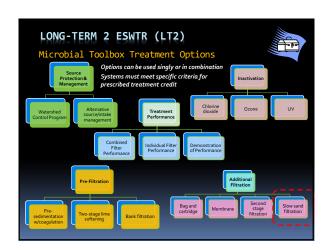


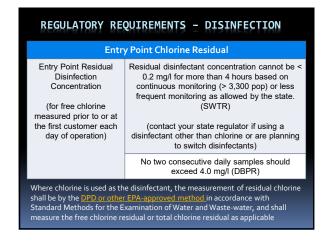




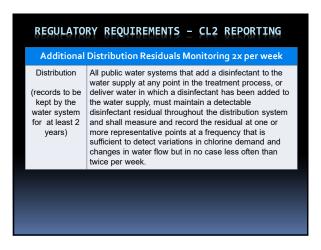


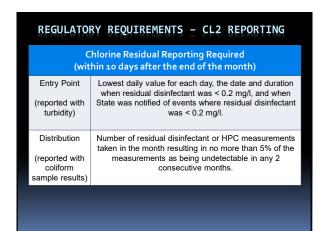
Filt	-TERM 2 ESWTR (LT2 ered System Additional Cryptosporid (based on their bin classification as di- and according to the sched	ium Treatment Requetermined under § 1.		
bin	Conventional Filtration (including softening), Slow Sand, or Diatomaceous Earth	Direct filtration	Alternative filtration technologies	
Bin 1	No Additio	nal Treatment		
Bin 2	1-log treatment	1.5-log treatment	RMVL + Inactivation > 4.0-log ¹	
Bin 3	2-log treatment	2.5-log treatment	RMVL + Inactivation ≥ 5.0-log ²	
Bin 4	2.5-log treatment	3-log treatment	RMVL + Inactivation > 5.5-log ³	





REGULATORY REQUIREMENTS - DISINFECTION **Distribution System Chlorine Residual** Distribution System Residual Residual disinfectant concentration cannot be undetectable in greater than Disinfection Concentration 5% of samples in a month, for any 2 (for free chlorine consecutive months. (SWTR) measured with coliform samples) (contact your state regulator if Not to exceed 4.0 mg/l MRDL* using a disinfectant other than (DBPR) chlorine or are planning to switch disinfectants) *The maximum residual disinfectant level (MRDL) is regulated under the Disinfection By-Products Rules (DBPR). Compliance is based upon chlorine residuals taken at the same location and frequency as that required for total coliform monitoring in the distribution system. The running annual average of monthly averages of samples, computed quarterly, must be <u><</u> 4.0 mg/l.







REGULATORY REQUI	REMENTS - OTHER
Other SWTR/IE	SWTR/LT1 Requirements
Disinfection Profiling & Benchmarking	Systems must profile inactivation levels and generate a benchmark, if required due to disinfection changes (IESWTR & LT1)
Water System Surveys (State Requirement)	CWS: Every 3 years NCWS: Every 5 years (IESWTR & LT1)
Finished Water Reservoirs	New (post-1989) reservoirs must be covered under SWTR. Pre-SWTR reservoirs must be covered (or have additional treatment) under LT2
Operator Certification	Operated by Qualified Personnel as Specified by State (SWTR)
(CWS) Community Water System (I	NCWS) Non-community Water System

	Cyanotoxin Monitoring (OAR 333-061-0510 to -0580) Healthoregon.org/dwcyanotoxins
Who does this apply to?	Affects systems who have sources susceptible to cyanobacteria blooms (not everyone). See list systems and specific rule requirements on-line at www.healthoregon.org/dwcyanotoxins
What is required?	Raw water (intake) sampling for total microcystin and Cylindrospermopsin toxins every 2 weeks from May 1^π – October 31^π each year
What happens if detected?	1. Notify your regulator 2. If any toxins are greater than or equal to 0.3 µg/L in raw water or if there is a recreational use health advisory" upstream of the intake, sample raw and entry point weekly with the first EP sample taken within 1 business day. Weekly sampling continues until non-detect at EP and less than 0.3 µg/L in raw water in two consecutive sampless. 3. If detected at EP sample EP daily and optimize treatment for toxin removal. 4. If above Health Advisor up evel (HAL) at EE take confirmation sample within 24-hrs & monitor EP daily. 5. If confirmation sample is above the HAL, issue Do-Not-Drink Advisory 6. Advisory may only be lifted if 2 consecutive daily EP samples taken a minimum of 24-hrs apart are 5 HAL and two consecutive daily sets of distribution samples taken a minimum of 24 hours apart are 5 HAL and two consecutive daily sets of distribution samples taken a minimum of 24 hours apart are 5 HAL. "Recreational use health advisory" means a health advisory issued by the Oregon Health Authority for a water body when cyanotoxins are determined to be above any recreational use advisory levels.
What are the DW Health Advisory Levels (HALs)?	 Total Microcystins: 0.3 µg/L for vulnerable people; 1.6 µg/L for all persons Cylindrospermopsin: 0.7 µg/L for vulnerable people; 3 µg/L for all persons "Vulnerable people" means infants, children under the age of six, pregnant women, nursing mothers, those with pre-existing liver conditions, and those receiving dialysis treatment.

REVIEW

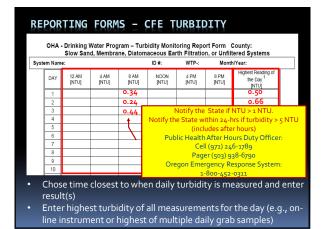
- 2.o-log Cryptosporidium removal is required (and credited) for slow sand filtration.
- Surface Water Treatment Rule (SWTR) requires 3log reduction of Giardia using a combination of disinfection and filtration and 4.0-log reduction of viruses.
- At least 2.0 -log Giardia removal is credited for slow sand filtration (per 1991 USEPA SWTR Manual)
- 1.0-log Giardia inactivation must be achieved through disinfection (0.5-log must be after filtration).
 1.0-log reduction of viruses must also be achieved after filtration.

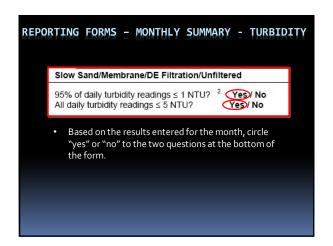
REPORTING FORMS

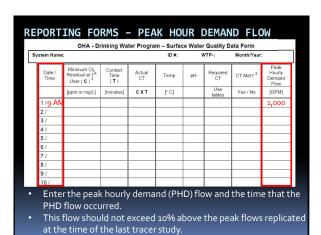
There are 4 forms:

- Conventional/Direct
- Slow Sand / Membrane / DE / Unfiltered
- Cartridge
- UV (if used for *Giardia credit*)

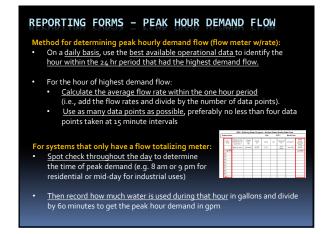
Must use correct form because each has questions that must be answered that are specific to the filtration type

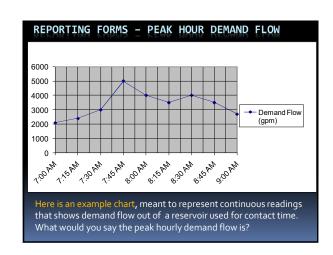


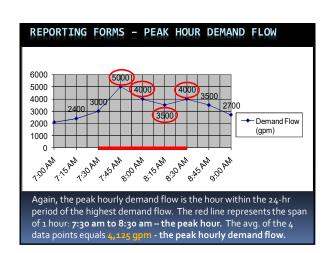


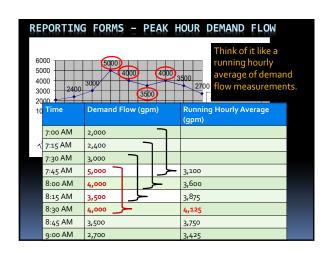


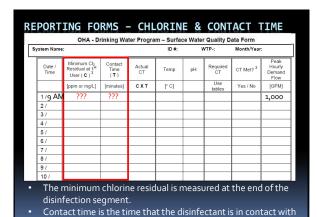
REPORTING FORMS - PEAK HOUR DEMAND FLOW | Cold-Design Prince Program- Future Name County Case Future | Cold-Design Prince Prince Peak County Case Future | Cold-Design Prince Peak County Coun



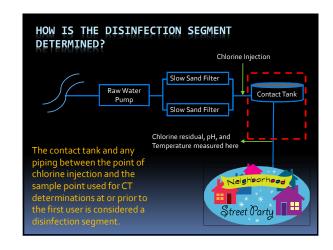








the water within the disinfection segment.

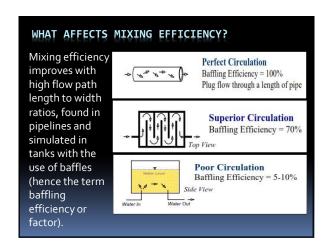


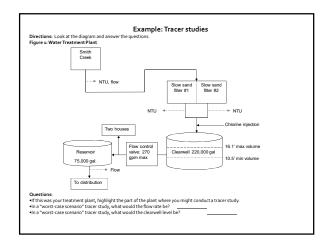
Tracer studies are used to determine contact time (T) which is used in calculating CT achieved, where CT = chlorine Concentration x contact Time. Contact time is the time that chlorine is in contact with the water from the point of injection to the point where it is measured (sometimes referred to as the "CT segment") May be at or before the 1st user May be more than one CT segment Tracer studies are often conducted to simulate a worst-case scenario where peak hour demand flows are high and reservoir.

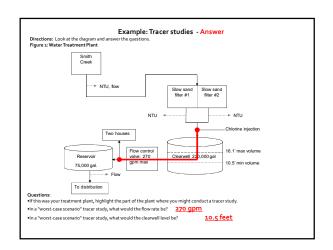
levels are low. This gives a conservative (i.e. lower) contact

time than would normally be expected.



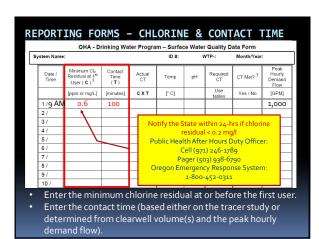






DO I REPORT CONTACT TIME?

- Use the time T from the tracer study on the monthly reporting form in the "Contact time (min)" column
 - Use the smallest T (highest flow) if the tracer study was done at multiple flow rates
- This may not be your exact time, but it represents your worst case (as long as the peak flow is less and clearwell volume is more than they were at the time of the tracer study)



CAN I USE A BAFFLING FACTOR? • As an alternative to using the tracer study contact time, you can use the results of the tracer study to determine the baffling factor of the clearwell Baffling factor (%) = <u>Time (min) x Flow During Tracer Study (gpm)</u> Clearwell Volume During Tracer Study (gal) T can be adjusted based on flow (at flow < 110% of tracer study flow) with the following equation: T = Current clearwell Volume (gal) x Baffling Factor (%) Peak Hourly Demand Flow (gpm)

• Contact the state for guidance on using baffling factors.

OHA - Drinking Water Program - Surface Water Quality Data Form								
rstem Name:				ID #:	WTP-:		Month/Year:	
Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? 3	Peak Hourly Demand Flow
	[ppm or mg/L]	[minutes]	CXT	[° C]		Use tables	Yes / No	[GPM]
1/9 AM	l 0.6	100	60					1,000
2/								
3 /								
4/								
5/								
6/								
7/								
8 /								
9/						Î		
10 /								

Actual CT = Chlorine Concentration (mg/l) x Contact Time (min)

• Do not confuse "CT" and "Contact Time"

		illikilig vvu	ter i rogra	m – Surfa					
stem Name:				ID #:	WTP-:		Month/Year		
Date / Time	Minimum Cl ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? 3	Peak Hourly Demand Flow	
	[ppm or mg/L]	[minutes]	СХТ	[° C]		Use tables	Yes / No	[GPM]	
1/9 AM	0.6	100	60	12	6.8			1,000	
2/									
3 /									
4/									
5/									
6/									
7/									
8 /									
9/									
10 /									

or prior to the first customer and after any storage (tank,

OHA - Drinking Water Progra ystem Name:				ID#:	WTP-:		Month/Year:	
Date / Time	Minimum CI ₂ Residual at 1 st User (C) ³	Contact Time (T)	Actual CT	Temp	pН	Required CT	CT Met? 3	Peak Hourly Deman
	[ppm or mg/L]	[minutes]	CXT	[° C]		Use tables	Yes / No	[GPM]
1/9 AM	l 0.6	100	60	12	6.8			1,000
2/								
3 /								
4/								
5/								
6/								
7/								
8 /								
9/								
10 /								

Actual CT must be ≥ Required CT. To determine required CT:

- 1. Use USEPA CT tables or
- 2. Regression Equations (Use 1 of 2 equations –depends on °C)

HOW IS REQUIRED CT CALCULATED?

- We use the EPA tables (or "regression equations") to determine the CT required to inactivate Giardia (CT_{required})
 - 1-log inactivation of Giardia using chlorine results in at least 4.0-log inactivation of viruses.
 - To determine CT, we need to know pH, temperature, and free chlorine residual at or before the first user.
- Then we compare the CT_{required} with the actual CT achieved in the water system (CT_{actual}) where:

CT_{actual} = chlorine concentration (mg/l) x contact time (min)

■ Must keep CT_{actual} ≥ CT_{required}

USING REGRESSION EQUATIONS TO DETERMINE REQUIRED CT Using Regression Equations to determine required CT:

- 1. Built into the MS Excel reporting forms on-line http://public.health.oregon.gov/HealthyEnvironments/DrinkingWater/Operations/Treatment/Pages/
- If your system has more than one chlorine injection point, or if you have questions about the PDF or MS Excel versions of the monthly furthally and surface water monitoring forms, contact the DWS technical oversight contact for your system at 971-673-6405.
- Conventional or Direct Filtration: PDF -or- MS Excel
- Slow Sand, Membrane, Diatomaceous Earth Filtration or Unfiltered: PDF -or- MS Excel
- · Cartridge or Bag Filtration: PDF -or- MS Excel

USING REGRESSION EQUATIONS, CONT.

ession Equations to determine required CT:

2. Regression equations can be programmed into plant SCADA or spreadsheets

Regression Equation (for Temp < 12.5°C) $CT = (0.353^{\circ}L)(12.006 + e^{(2.46-0.073^{\circ}T+0.125^{\circ}C+0.389^{\circ}PH)})$

Regression Equation (for Temp > 12.5°C)

 $CT = (0.361*L)(-2.261 + e^{(2.69-0)})$

<u>Variables:</u>
CT = Product of Free Chlorine Residual and Time required

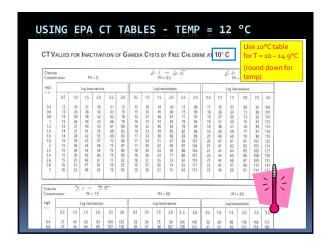
- L = number of log inactivation for Giardia (L = 1 for slow sand)
- T = temperature, in Celsius
- C = chlorine residual in mg/L pH = pH of water

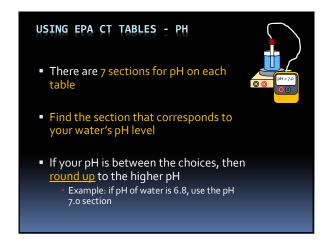
pr = pr of variety e = 2.7183, base for natural log (Smith, Clark, Pierce and Regli, 1995, from EPA's 1999 Guidance Manual for Disinfection Profiling and Benchmarking)

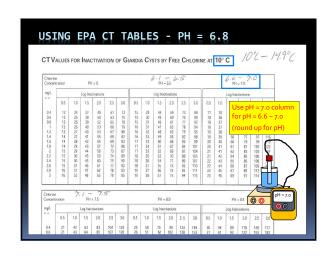
USING EPA CT TABLES - TEMPERATURE

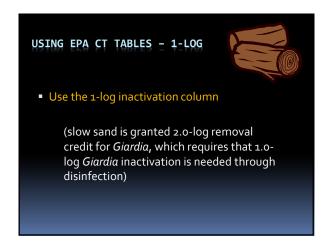


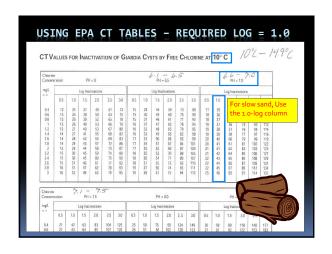
- There are six EPA CT tables based on temperature
- Find the correct table based on your water temperature in degrees Celsius.
 - °C = 5/9 x (°F 32)
- If water temp is between values, then round down
 - Example: for water temp of 12°C, use the 10°C
 - Even if the water temp is 14.9°C, round down to
- Water gets more viscous the colder it gets and chemical reactions take longer, so rounding temp down is more conservative.

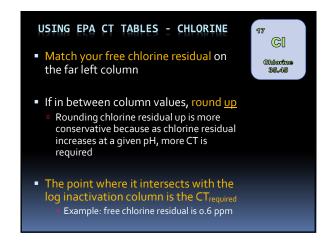


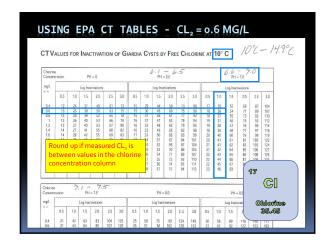


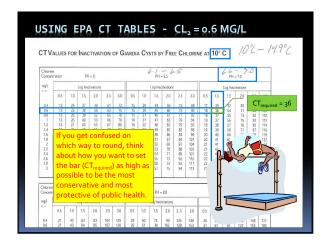


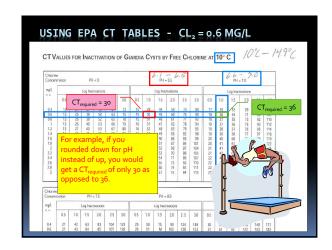


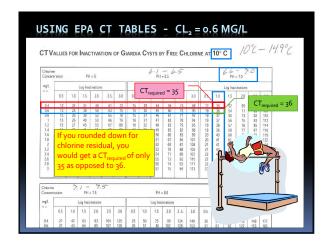


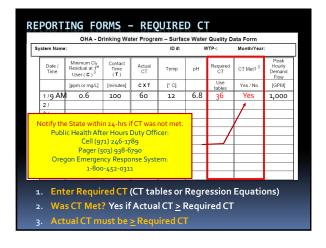


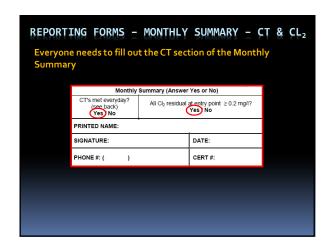


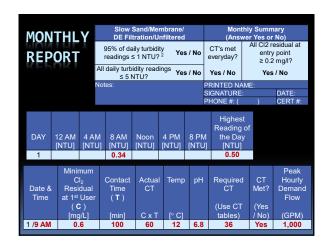












FILLING OUT THE MONTHLY REPORT - COMMON MISTAKES

- Not calculating CT's daily
 - Don't wait until the end of the month to do the calculations because if you discover you didn't meet CT's, it's too late!
- If adjusting contact time according to flow rate, use the demand flow, not the plant flow.
- Failure to answer questions at bottom of form correctly (or at all)
- Always answering "Yes" to the questions at the bottom of the form without actually looking at the numbers

FILLING OUT THE MONTHLY REPORT - COMMON MISTAKES

- Rounding errors when using EPA tables to determine CT_{required}
 - Must round <u>down</u> for temperature
 - Must round <u>up</u> for pH
 - Must round <u>up</u> for free chlorine residual
- Bad CT formulas in excel spreadsheets:
 - Make sure you understand your formula
 Wilkes Equation not allowed, must use Regression
 Equation

FILLING OUT THE MONTHLY REPORT - AVOIDING MISTAKES

- Check how T is calculated at your plant
- Do all treatment plant operators understand it?
- Review spreadsheet equation for CTs (if applicable)
- Write an SOP for CT determination
- Arrange for a tracer study if necessary
- Calculate CT and fill out monthly report daily
- Know what to do and who to call when things go wrong (contact State regulator & refer to Emergency Response Plan)

STRIVE TO IMPROVE DATA QUALITY

- Make data reliability a plant goal
- Only collect data used for process control or compliance reporting
- Establish protocols for collection and recording of data
- Establish a data verification process that can be routinely used to confirm data integrity
- Turn data into information (e.g., draw the graph).



O&M MANUALS

Keep written procedures on:

- Instrument calibration methods and frequency
- Data handling/reporting
- Chemical dosage determinations
- Filter operation and cleaning
- CT determinations
- Responding to abnormal conditions (emergency response plan)

DISINFECTION





- 2. Chemical (chlorine, chloramines, chlorine dioxide, ozone)
- Forms of chlorine
- NSF/ANSI Standard 6o



TYPES OF DISINFECTANTS - UV

- Works by subjecting water to ultraviolet (UV) light rays as water passes through a tube
- Drawbacks:
 - Interfering agents such as turbidity can screen pathogens from the UV light
 - Effective against Giardia and Cryptosporidium but not viruses at normal doses
 - No residual is present throughout the distribution system
 - For this reason, chlorination for residual maintenance is required when UV is used



TYPES OF DISINFECTANTS - CHEMICAL

- 1. Chlorine
- 2. Chloramines
- 3. Chlorine dioxide
- 4. Ozone

TYPES OF DISINFECTANTS - CHLORINE

- The most widely used form of disinfection
- Also used as an oxidizing agent for iron, manganese and hydrogen sulfide and for controlling taste and odors
- Effectiveness as a disinfecting agent depends on factors such as pH, temperature, free chlorine residual, contact time and other interfering agents

FORMS OF CHLORINE

- Sodium Hypochlorite
- Onsite generated sodium hypochlorite
- Calcium Hypochlorite
- Chlorine Gas

FORMS OF CHLORINE - SODIUM HYPOCHLORITE

- The liquid form of chlorine
- Clear and has a slight yellow color
- Ordinary household bleach (~5% chlorine by solution) is the most common form
- Industrial strength: 12% and 15% solutions
- Can lose up to 4% of its available chlorine content per month; should not be stored for more than 60 to 90 days
- Very corrosive; should be stored and mixed away from equipment that can be damaged by corrosion

FORMS OF CHLORINE - SODIUM HYPOCHLORITE

Diaphragm pump with chlorine solution tank

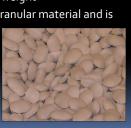
ON-SITE GENERATED SODIUM HYPOCHLORITE

- o.8% sodium hypochlorite is produced on demand by combining salt, water & electricity
- Electrolysis of brine solution produces sodium hydroxide and chlorine gas, which then mix to form sodium hypochlorite
- Hydrogen gas byproduct; vented to atmosphere
- Alleviates safety concerns associated w/ hauling and storing bulk chlorine
- Higher initial cost, high power cost
- Mixed oxidants (proprietary)



FORMS OF CHLORINE - CALCIUM HYPOCHLORITE

- The solid form of chlorine
- Usually tablet or powder form (see photo below)
- Contains ~65% chlorine by weight
- White or yellowish-white granular material and is fairly soluble in water
- Important to keep in a dry, cool place
- More stable than liquid
- Common in small systems w/ low flows or no power



FORMS OF CHLORINE - CALCIUM HYPOCHLORITE Erosion chlorinator Inside Hopper =>

FORMS OF CHLORINE - CHLORINE GAS

- 99.5% pure chlorine
- yellow-green color 2.5x heavier than air
- Liquefied at room temperature at ~107 psi hence the pressurized cylinders actually contain liquefied chlorine gas.
- Liquefied Cl₂ is released from tanks as chlorine gas, which is then injected into the water stream.
- usually used only by large water systems
- Smaller systems may find initial cost of operation prohibitive



FORMS OF CHLORINE - CHLORAMINES

- Chlorine + ammonia = chloramination
- Two advantages to regular chlorination:
 - produce a longer lasting chlorine residual (helpful to systems with extensive distribution systems)
 - may produce fewer by-products depending on the application
- Disadvantage:
 - Need a lot of contact time to achieve CTs compared to free chlorine (300 times more) which is why not used for primary disinfection
 - Requires specific ratio of chlorine to ammonia or else potential water quality problems

CHLORINE DIOXIDE - ADVANTAGES

Advantages:

- 1. More effective than chlorine and chloramines for inactivation of viruses, *Cryptosporidium*, and *Giardia*
- 2. Oxidizes iron, manganese, and sulfides
- 3. May enhance the clarification process
- ControlsT&O resulting from algae and decaying vegetation, as well as phenolic compounds
- Under proper generation conditions halogen-substituted DBPs are not formed
- 6. Easy to generate
- 7. Provides residual

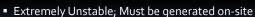
CHLORINE DIOXIDE - DISADVANTAGES

Disadvantages

- 1. Forms the DBP chlorite
- 2. Costs associated with training, sampling, and laboratory testing for chlorite and chlorate are high
- 3. Equipment is typically rented, and the cost of the sodium chlorite is high
- 4. Explosive, so it must be generated on-site
- 5. Decomposes in sunlight
- 6. Can lead to production noxious odors in some systems.

OZONE

- Colorless gas (O₃)
- Strongest of the common disinfecting agents
- Also used for control of taste and odor







OZONE - ADVANTAGES

Advantages:

- Short reaction time enables microbes (including viruses) to be killed within a few seconds
- Removes color, taste, and odor causing compounds
- 3. Oxidizes iron and manganese
- 4. Destroys some algal toxins
- 5. Does not produce halogenated DBPs

OZONE - DISADVANTAGES

Disadvantages:

- 1. Overfeed or leak can be dangerous
- 2. Cost is high compared with chlorination
- 3. Installation can be complicated
- May produce undesirable brominated byproducts in source waters containing bromide
- No residual effect is present in the distribution system, thus post-chlorination is required
- 6. Much less soluble in water than chlorine; thus special mixing devices are necessary









